

WINDERMERE LAKE

**WATER QUALITY CHANGES OVER THE PAST 300 YEARS AS
DETERMINED FROM A BOTTOM SEDIMENT CORE**

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Pollution Prevention
Kootenay Region
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Executive Summary

Windermere Lake is an important recreational feature in the East Kootenays. It is particularly so to the growing population around the lake who also rely on the lake as a source of drinking water. The question for residents and those involved in land and water management is whether past and present human activity (agriculture, industry and, in particular, residential development with on-site sewage disposal) has had any negative effects on lake water quality?

Various water quality monitoring efforts over the past 25 years have concluded the lake has reasonably good water quality, is not showing definite signs of eutrophication, nor is there serious bacterial contamination from the septic field disposal of sewage of the population on the east side of the lake. More subtle indications of concern, such as possible slight increases in lake-wide phosphorus and septic field leachate “hot spots” with associated high bacterial counts, have been offset by the knowledge that the lake has a high flushing rate.

It is difficult to rely on water quality sampling to determine subtle, long-term changes in lakes. Sampling seldom begins prior to human influence, which may extend into the last century. Years, if not decades, of sampling may be necessary to detect changes, which must be distinguished from seasonal and annual variation and unavoidable analytical error. Chemical and biological evidence from bottom sediment cores has become an invaluable tool in reconstructing the water quality history of lakes, often uncovering changes in flora and fauna that could never be detected through water sampling.

A sediment core from Windermere Lake, sampled in 1998, shows that water quality began to change around 1950, concurrent with population growth. There is some evidence of slight eutrophication (increasing nutrients causing more algal growth) coupled with a change in the composition of the phytoplankton community. Types of algae commonly associated with clean water and low nutrient levels, have been displaced by forms that may cause taste and odour problems in water supplies. Water users on systems with intakes drawing from the bottom of the lake **may** detect unpleasant odours or taste during the winter when the lake is covered with ice and snow.

The findings of this study do not mean that there is currently a serious problem in drinking water quality in Windermere Lake. A consumer's personal preferences and experience heavily influence the acceptability of drinking water. Windermere Lake water quality was probably better prior to 1950 than it is today, though it is not certain that a consumer would be able to distinguish the change.

The most important finding of this study is that there is a significant amount of evidence that Windermere Lake water quality is slowly deteriorating. The study has produced little hard evidence linking this change to increased human settlement in the basin but this is circumstantially the most logical cause.

Acknowledgements

The coring apparatus and skill to use it was kindly provided by Dr. Richard Nordin during his lake coring tour of southern B.C. in the summer of 1998. Dr. Gordon Kan of the Pacific Environmental Science Centre handled the management of the various university contracts. The report was reviewed by Ms. Julia Beatty, Dr. Richard Nordin, and Mr. Robert Wetham.

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Introduction and Methods

On July 23, 1998 a core of bottom sediment was taken from the deepest point in Windermere Lake, off the Akiskanook resort, by staff of the Ministry of Environment, Lands and Parks. Thin 5 mm sections of the core were stored in plastic cups and forwarded to Dr. Brian Cumming at Queen's University in Kingston, Ontario. Dr. Cumming's laboratory analyzed some of the sections for fossil algal remains. These single-celled algae are called diatoms and their cell walls, comprised of silica and called "frustules", remain after death like shells. These frustules are unique to each species and allow the make-up of the diatom community in each slice of the sediment core to be determined.

Changes to the diatom community occur when water quality, particularly the nutrient phosphorus, increase or decrease. The eutrophication of a lake (increase in algal growth) is usually caused by an increase in phosphorus and is often the result of human settlement (accelerated erosion, sewage, lawn fertilizer, etc.) or other human endeavours such as agriculture and industry. Lake eutrophication is often a slow process taking decades before effects can be perceived by other than sophisticated technical studies. By the time water use (drinking supplies, recreation, etc.) become noticeably impaired, the effort required to alter the human activity related cause can be substantial and expensive.

Sub-samples of the core were forwarded by Dr. Cumming to MYCORE Ltd. laboratories for Lead-210 (^{210}Pb) dating and to Dr. Peter Leavitt at the University of Regina for plant pigment analysis. Remaining sediment from these analyses was returned to MELP and analyzed by our contract laboratory for a number of metallic elements. Plant pigments, characteristic of certain types of algae, also change in relation to algal and larger aquatic plant community changes. Lead-210 dating allows sections of the core to be chronologically dated so changes in the algal community and water quality can be related to historical events in the lake basin such as forest fires, agriculture or human settlement.

Two technical reports by Drs. Cumming and Leavitt are appended; this brief cover report is an attempt to provide a less technical summary of these findings combined with the elemental analysis performed by MELP.

The intent of this work is to determine, to the extent possible, whether Windermere Lake water quality has changed over the recent past and, if so, how significant the change. There is some discussion about the reasons for down-core change and whether human settlement may be a cause. This information should be useful to local government in future land-use planning.

Field Observations

The core was taken in a depth of approximately 3.5 meters (maximum lake depth is about 6 meters, off Timber Ridge). The lake is 1.6 km across at the sample point so the core was about 800 meters from either shore. There was a small amount of *Chara sp.* or stonewort on the surface of the core but no other aquatic vegetation in the immediate vicinity. The core had no obvious banding, though the top 3 cm was light brown and the rest was a uniform dark grey. The

core had a strong sulphide odour indicating an absence of oxygen, a condition in which organic material, like plant pigments, are well preserved.

Core Dating and Diatom Community Changes

The following is a summary of Dr. Cumming's report *Paleolimnological analysis of Windermere Lake, B.C.—Final Report (March 1999)* (Appendix I) with some possible explanations for the findings as they relate to lake basin history.

The ^{210}Pb profile and percent organic matter suggest some mixing of surface sediments which is not surprising in a shallow lake with a large surface area like Windermere Lake. The mixing of surface sediments due to wind and currents is not a desirable feature for lake coring because older and newer sediments near the top or the core are mixed during randomly occurring wind events. The sulphide smell of the core, taken in mid-summer after considerable wind exposure, indicates little or no oxygen in the sediments suggesting that sediment mixing is confined to the upper few millimeters.

The fact that the bottom of the 35 cm core dates back to 1700 indicates a slow sedimentation rate, an average of 1.17 mm/year. Sediment deposited in the center of lakes is derived from erosion of the basin, transported in via inflowing streams, and from settling biological material that has grown in the lake. Windermere Lake has a relatively small volume compared to the amount of water flowing through, mostly via the Columbia River. This means the lake has a high flushing rate or short water residence (McKean and Nordin 1985, reported the lake volume replaced every 47 days). Rapid flushing means the internal production of sediment is relatively low, a condition less than ideal for sediment core analysis. The inflowing Columbia River passes through a series of wetlands (Mud Lake and Tatley Slough) before entering the lake, which may serve to reduce sediment loads.

The sedimentation rate and the percent organic matter began to increase around 1920, peaking in the late 1940s, then falling back to historic levels from 1960 to the present. This may have been due to land disturbance from logging or cattle ranching, which increase erosion and sediment transport to the lake. Forest fires may also have been a factor, though response to such events is usually more dramatic and short lived, and surface sediment mixing due to wind might obliterate the evidence of such events. Whatever the cause of the increased sedimentation rate, it does not appear related to settlement, which did see an increase from 1920 to 1960 but has greatly accelerated since 1970.

The diatom assemblages found in Windermere Lake sediments were dominated by benthic species (those living on the surface of the sediments) with some epiphytic forms (those living on aquatic plants). Planktonic species (those living in the water column), the most abundant forms in deeper lakes with smaller littoral zones, were relatively rare in Windermere Lake. In deeper lakes (>15 meters) benthic species would be a minor component, only living in the shallow margins because there is insufficient light penetration for photosynthesis. Windermere Lake is shallow enough throughout so that sufficient light reaches the bottom for algal growth and there is abundant large aquatic plant growth upon which epiphytic algae grow. The database used by Cumming to infer historical water quality is comprised of the over 200 B.C. lakes that are

generally deeper, with minor benthic components (Wilson *et al.* 1996). This means that diatom analysis is less useful in reconstructing historic water quality in Windermere Lake than other core methods employed.

Dr. Cumming's analysis does show some minor changes in the diatom community structure starting around 1950, with the loss of a few species and the appearance of others. Unfortunately the species involved are not well represented in the database from other B.C. lakes, which is used to infer changes in lake phosphorus. This means that predictions of the change in total phosphorus in Windermere Lake over time must be considered preliminary.

The observed changes in the diatom community do generally correspond with an increase in settlement around the lake. Of a total of 1290 lots subdivided on the east side of the lake from 1940 to 1986, 623 (48%) were established between 1947 and 1957 (Whetham, personal communication 1999). Although the diatom community change evidence from the core is not strong, it does suggest a subtle change in water quality, probably an increase in nutrients, and is circumstantially associated with accelerated settlement since 1950.

Fossil Plant Pigments

The following is a summary of Dr. Leavitt's report *Analysis of Fossil Pigments from Windermere Lake (March 1999)* (Appendix II).

Many algal photosynthetic pigments are unique to certain taxonomic groups and the various types of algae found can influence water quality. These pigments decompose in the presence of oxygen but are preserved in sediments when oxygen in the overlying water is low or when diffusion into the sediment is limited. Being shallow and wind-blown, the overlying water in Windermere Lake is rarely depleted of oxygen, save perhaps under ice and snow cover. The low sedimentation rate and compact nature of the bottom sediment would seem to be limiting the downward diffusion of oxygen to the upper most layer. The sulphide smell of the core, evidence of an absence of oxygen, confirms this.

Leavitt concludes, based on the increase in newer sediments of certain pigments common to all algae, that there has been a slight increase in algal production (eutrophication) since about 1960. Other changes in specific pigments indicate a shift in the algal community away from diatom dominance and toward green, blue-green and chrysophyte types over this period. Unlike diatoms, these other forms of algae do not leave behind fossil "shells", so photosynthetic pigments, preserved in sediments void of oxygen, are the only evidence of their historical representation in the phytoplankton community. Leavitt, who was given no prior information about Windermere Lake, describes these algae as "bloom-forming" types and states that the pigment evidence does not allow distinction between surface forms that people would see or deep-water forms that would occur unnoticed. Windermere Lake, of course, has no deep water and these algae have not yet reached bloom proportions. The lake water might appear less clear and greener in color today compared to pre-1960, if one could make such a visual comparison, but the difference would probably be subtle.

Sampling done by MELP in the lake in 1982 and 1983 indeed showed the dominance of two blue-green genera, *Gomphosphaeria* and *Chroococcus*, and the chrysophyte *Dinobryon* in the late-summer/fall and the diatoms *Synedra* and *Cyclotella glomerata* (McKean and Nordin 1985). Standard Methods for the Examination of Water and Wastewater (APHA 1975) identifies *Gomphosphaeria*, *Chroococcus*, *Dinobryon*, and *Synedra* as types of algae that can cause taste and odour problems in drinking water.

Algal pigment analysis also shows the presence of purple-sulphur bacteria in Windermere Lake. These photosynthetic bacteria are usually found in deeper lakes with permanent stratification and a bottom layer of water (hypolimnion) completely void of oxygen. The bacteria grow in the region at the top of this anoxic layer where there is low but sufficient light. Their photosynthetic pigments often give the water in this region a purple color, thus the name. The presence of these bacteria is not desirable in a lake because the permanent lower strata of the lake, in addition to being void of oxygen and thus uninhabitable to fish, accumulates odorous by-products of organic decay like methane, ammonia and hydrogen sulphide. Windermere Lake is too shallow to develop such stratification. These bacteria, which Leavitt states are at lower levels than strongly stratified lakes, probably grow on the surface of the sediments during winter ice and snow cover when oxygen depletion occurs in a small zone immediately above the sediments where low but sufficient light penetrates. Another requirement for the growth of these bacteria is sufficient quantities of sulphur, seldom in short supply in permanently stratified lakes. Windermere Lake receives large quantities of sulphate from the hot springs upstream and the gypsum (calcium sulphate) deposits in Windermere Creek.

Of particular concern is Leavitt's finding that the purple-sulphur bacterial pigment has increased 5 to 10 fold in the lake since about 1960. The consequence of such a change to drinking water supplies may be a deterioration in aesthetic water quality during the winter as quantities of sulphides, ammonia and methane increase, particularly in water intakes drawing from near the bottom. Leavitt states that the reasons for this increase are greater oxygen deficits in bottom waters due to stronger stratification or increased light penetration. Given the shallow depth of the lake neither of these causes seems likely. Because these increases in purple-sulphur bacteria occur at the same time as the shift in the phytoplankton community from diatoms to blue-green and chrysophytes, the causes are probably the same. This is also the period in the lake's recent history of the greatest human settlement. Though these associations seem reasonable cause-effect relationships the available evidence is purely circumstantial. Unfortunately conclusive evidence that human-settlement has caused a deterioration in water quality may only be forthcoming if the degradation becomes severe.

Elemental Analyses

The results of elemental analysis of selected sections of the sediment core are listed in Table 1 and summarized in Figures 1- 4. Unfortunately there was no sediment left in the sections from the upper 3 cm of the core because it was all required for other analyses. Elemental analysis thus covers a time period from about 1700, at the core bottom, to 1983.

The alkaline-earth metals, (in order of abundance) calcium, magnesium, barium, strontium, and beryllium (Figure 1, beryllium not graphed, see Table 1), are among the most common cations in

the lakes and streams of the East Kootenays. Calcium and magnesium are by far the most abundant and are the main causes of water hardness leading to scaling problems in potable water. These elements are found throughout the watershed as carbonates such as limestone or chalk (calcium carbonate), magnesite (magnesium carbonate) or as sulphates, e.g. gypsum (calcium sulphate), barite (barium sulphate). Weathering and erosion release these minerals which are then transported via tributaries to the lake. Human activities such as agriculture, mining, logging, road construction and settlement can accelerate erosion leading to increases in the deposition of these elements in lake sediments (Wetzel 1983).

Why calcium and strontium show increasing trends, commencing around 1950, while magnesium and barium show decreasing trends over the same period (Figure 1) is unknown. Gypsum deposits (calcium sulphate) in the south fork of Windermere Creek are unlikely the source of increasing calcium because sulphur shows an opposite, decreasing trend (Figure 3).

A number of other elements exhibit similar patterns of change over time. Quantities of iron (Figure 2), potassium and sulphur (Figure 3), aluminium, and titanium (Figure 4) all begin to decline around 1950. The inputs of extremely inert elements such as titanium are used to determine erosion rates in lake basins (Wetzel 1983). Falling titanium concentrations in this core indicate that erosion rates are decreasing in the Windermere Lake drainage basin from 1950 to 1983, and this seems corroborated by Cumming's sedimentation rate estimates which also decline after 1950.

Boron (Figure 3) shows a different pattern than any other element. Core concentrations remained uniformly low at around 50 µg/g from the bottom (1700) until around 1960 when they increased to over 200 µg/g. This may be an indication of human related sources entering the lake as boron is found in detergents, water softeners, pesticides, and fertilizers. This increase in boron, an algal micronutrient, may be a contributing factor in the change in the phytoplankton community, observed at a similar level in the core, by providing a competitive advantage to certain species that did not previously exist (Wetzel 1983).

Phosphorus, the most critical nutrient controlling algal and weed growth, appears to have steadily increased since about 1800, almost doubling by 1983. Iron, and to a lesser extent manganese, are important in controlling phosphorus concentrations in lakes by chemically binding the phosphorus in insoluble forms in the sediments. Sulphur "competes" with phosphorus for iron and manganese and the addition of sulphate to lakes has been used as a means of "fertilizing" by increasing the availability of the phosphorus already present in the lake (Wetzel 1983). Windermere Lake has ample quantities of sulphate from its drainage basin so that iron complexing of phosphorus is probably less important than in a lake with lower sulphur levels. The close association between iron and sulphur in the sediments can be seen in the core graphs of each element (Figures 2 and 3) which are almost identical. The presence of large quantities of ferrous sulphide in the sediments was confirmed by the "rotten egg" smell of the core. The reason for the simultaneous decrease in iron and sulphur, commencing around 1950, is unknown.

Calcium, which has increased 87% in the core from 1950 to 1983, probably exists mainly as calcium carbonate, which can also bind phosphorus. While the cause of the increase is unknown, this may explain an accelerated increase in phosphorus over the same period. Unless the inflow

of phosphorus to the lake has also increased, the consequence of an increase in sediment phosphorus would be a decrease in available phosphorus in the water column. McKean and Nordin (1985) reported an apparent increase in phosphorus in the lake between 1973 and 1983, probably relating to human activities, but caution that differences may only be seasonal variation. Monitoring subtle changes to available phosphorus in lakes over decades is a difficult proposition because of annual and seasonal variation and technical challenges, though the effects on the phytoplankton community may be significant.

Conclusions

- Diatom community reconstruction from sediment core sectioning are a less useful tool for determining historic water quality in Windermere Lake than in deeper lakes. Light penetration to the bottom throughout the lake means that the benthic component is much larger compared to the phytoplankton component than in most of the B.C. lakes used to develop Cummings' total phosphorus inference model.
- Notwithstanding the above, there have been subtle changes in the diatom community in the lake starting around 1950 which are probably associated with a change in water quality.
- Algal pigment analysis indicates a slight eutrophication since 1960 and a shift from clean-water diatom species to varieties of phytoplankton that are potentially bloom-forming and are associated with taste and odour problems in potable water supplies.
- There has been an increase in photosynthetic purple sulphur bacteria in the lake sediments since 1960. This means that dissolved oxygen is completely absent from the interstitial water in the uppermost sediments, probably during winter ice cover. A consequence of this condition could be the accumulation of undesirable anoxic by-products such as hydrogen sulphide, ammonia and methane in water near the bottom of the lake, under winter ice, where some water supplies draw from.
- Chemical analysis of the core shows levels of a number of elements decreasing in abundance since 1950 and a small number increasing over the same period. This indicates a change in water quality in the lake since 1950, which corresponds with core observed changes in the algal communities and, circumstantially, with the sustained increase in human activity, mainly in the form of settlement.
- An increase in phosphorus in the sediments since 1950 may be an indication of eutrophication, caused by human activity. The increase matches a similar increase in calcium which, as calcium carbonate, probably serves as a phosphorus-binding agent. Were this latter process alone occurring, without an increase in phosphorus entering the lake, the result would be a more oligotrophic state (less nutrient, clearer water) which is the converse of the algal evidence.

Recommendations

Future lake monitoring programs should include some effort to confirm the findings of this sediment core study, including the following:

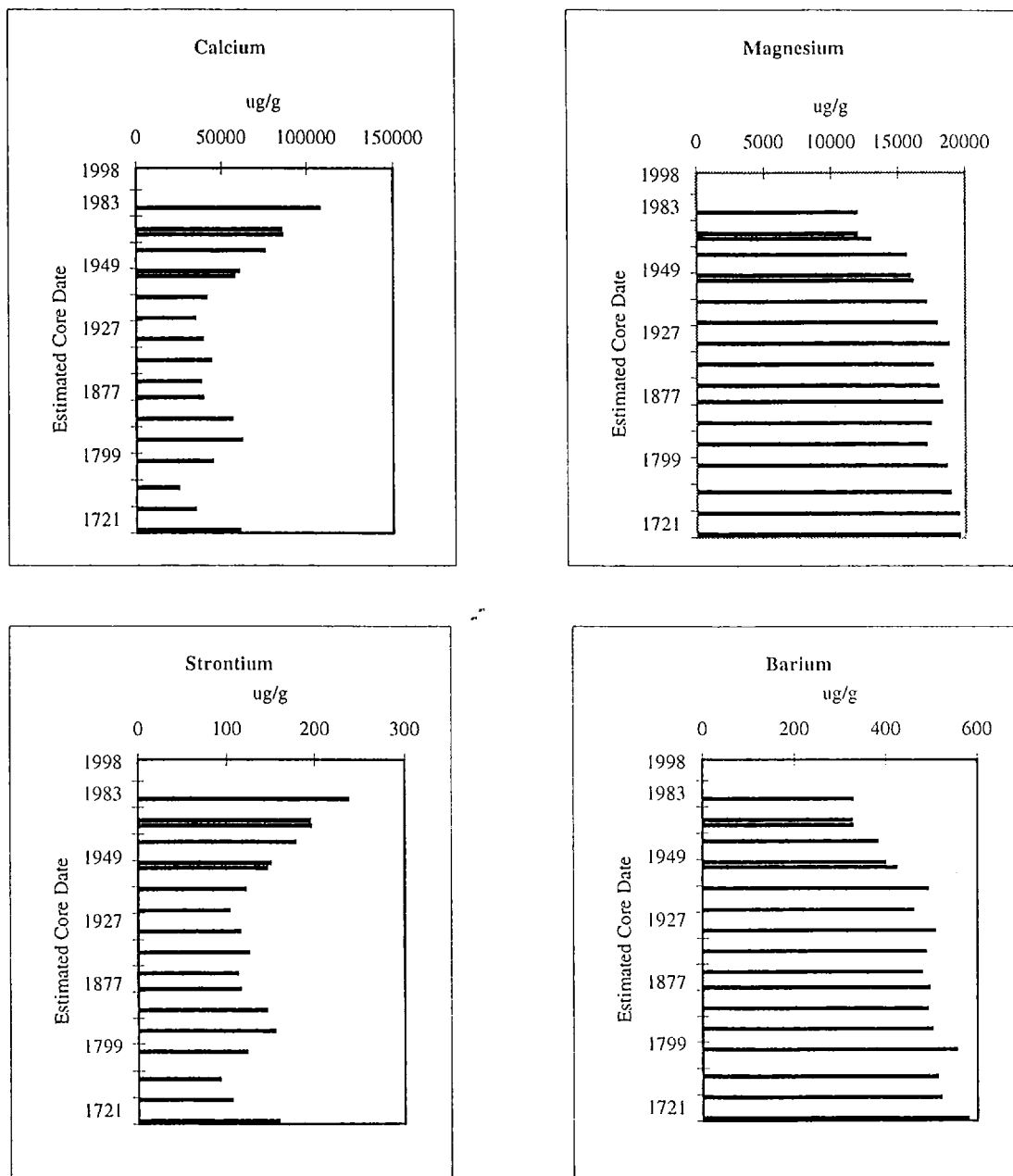
- Phytoplankton sampling over one year, including under various ice-cover conditions (i.e. heavy snow cover in January, clear ice in February/March) to confirm the dominance of colonial blue-green and green species in relation to diatoms and determine how the community structure changes over the seasons. This effort should be accompanied by nutrient sampling employing low-level detection analysis and should also include some analysis of trace elements in the water column and perhaps the major inflowing streams.
- Purple-sulphur bacteria should be sampled in surface sediments several times through the year, focusing on the ice-cover period. In addition to taxonomic identification and enumeration of species, group specific pigments may be analyzed as a means of determining the abundance of this group. Water just above the lake bottom should be sampled at the same time and analyzed for total sulphides, ammonia, and trace metals and a careful field survey of dissolved oxygen, pH and oxidation-reduction potential (ORP) done at various depths.

References

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- Wetzel, R.G. Limnology. Second Edition. Saunders College Publishing. New York. 767p.
- Wilson, S.E., B.F. Cumming and J.P. Smol. 1996. Assessing the reliability of salinity inference models from diatom assemblages: An examination of a 219 lake dataset from western North America. Can. J. Fish. Aq. Sci. 53: 1580-1594.

Figure 1. Windermere Lake Sediment Core

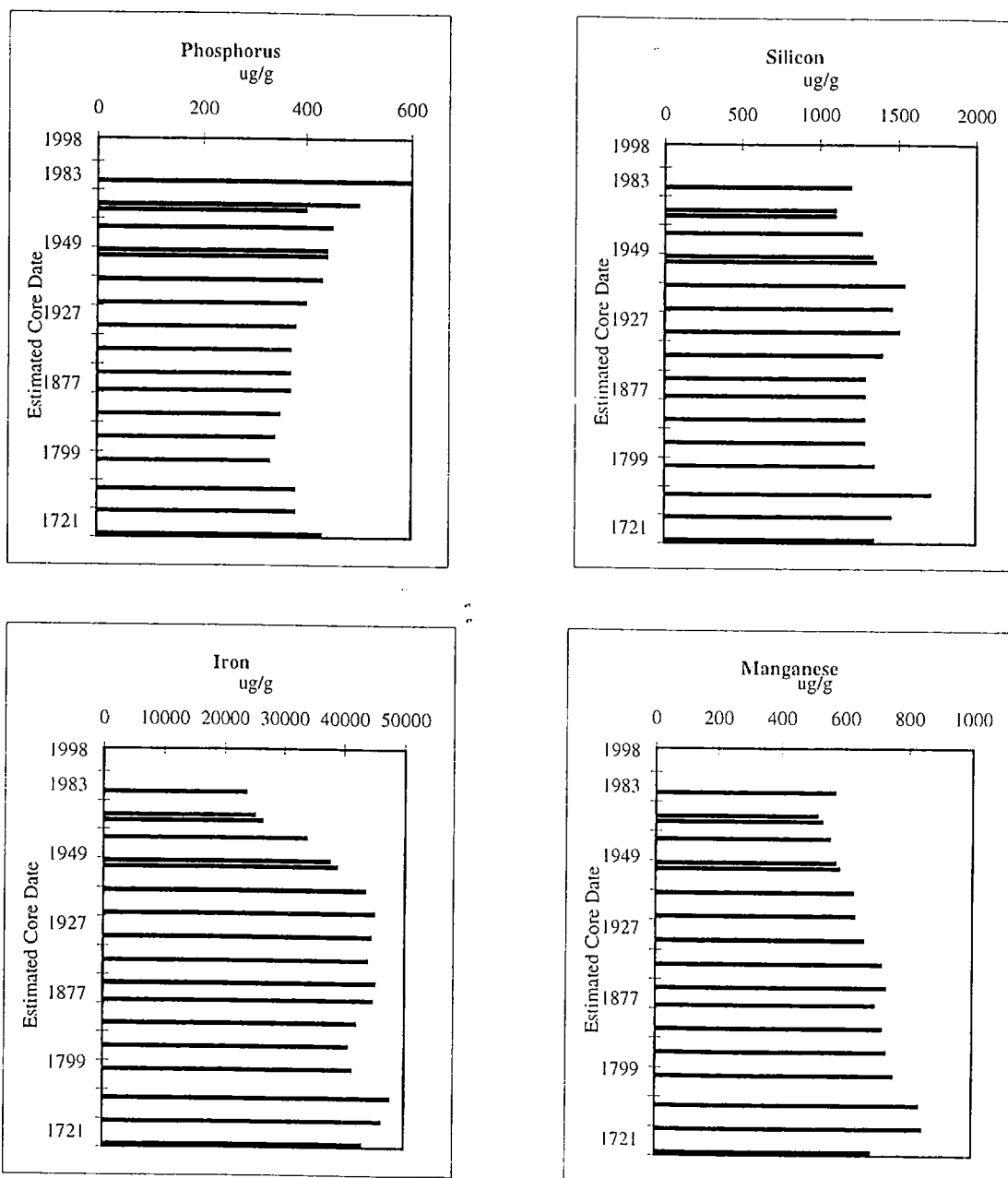
Alkaline-Earth Elements



Core approximately 35 cm total depth, sampled on July 23, 1998. All concentrations in $\mu\text{g/g}$ dry weight.

Figure 2. Windermere Lake Sediment Core

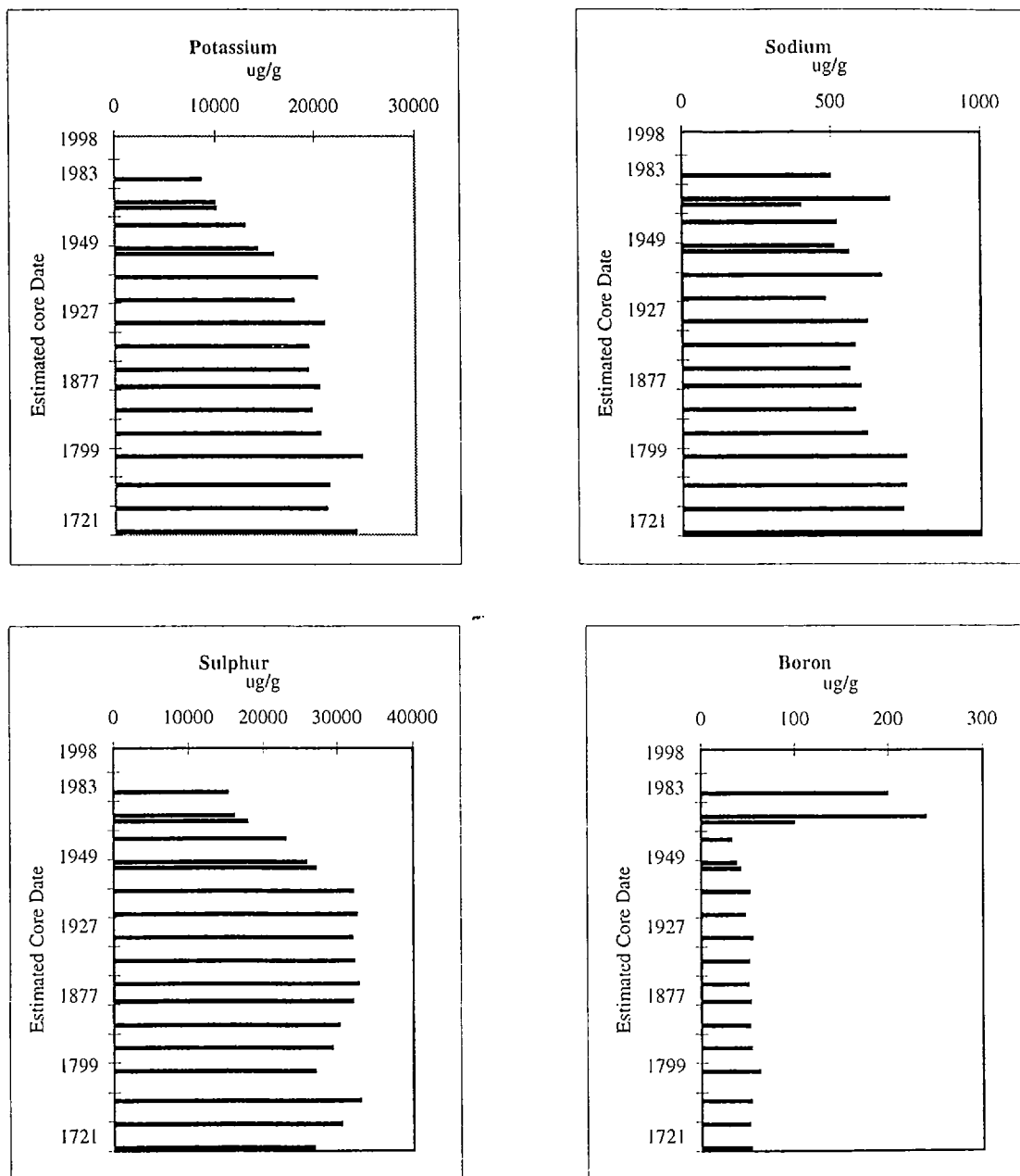
Nutrients and Binding Elements



Core approximately 35 cm total depth, sampled on July 23, 1998. All concentrations in $\mu\text{g/g}$ dry weight.

Figure 3. Windermere Lake Sediment Core

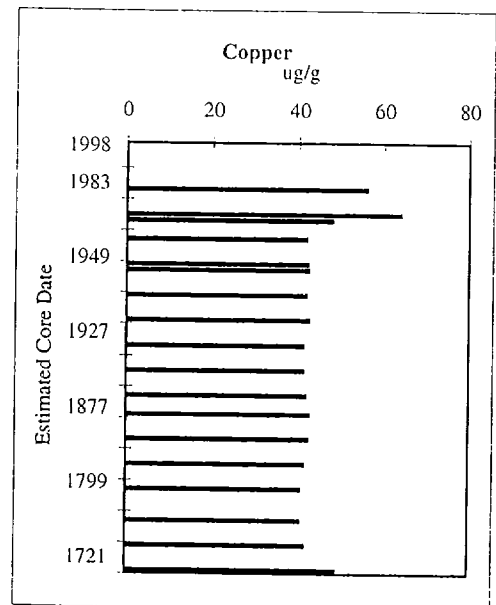
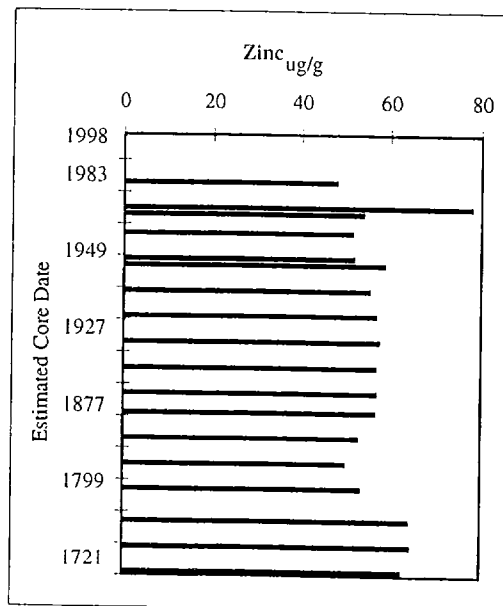
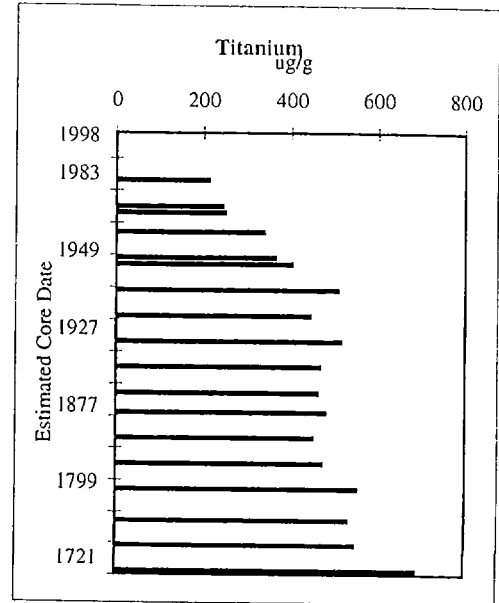
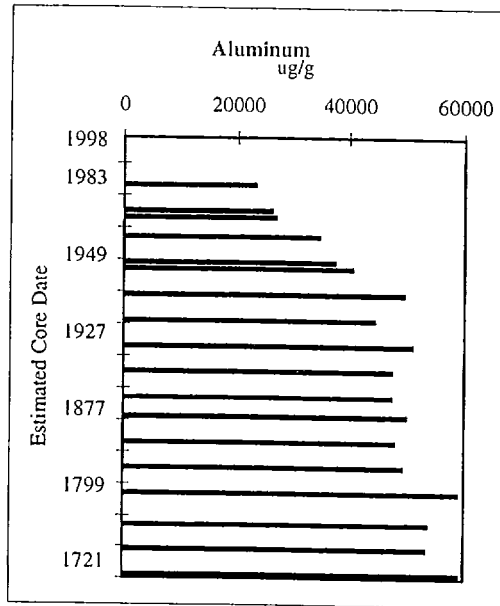
Alkali Metals, Sulphur and Boron



Core approximately 35 cm total depth, sampled on July 23, 1998. All concentrations in $\mu\text{g/g}$ dry weight.

Figure 4. Windermere Lake Sediment Core

Metals



Core approximately 35 cm total depth, sampled on July 23, 1998. All concentrations in $\mu\text{g/g}$ dry weight.

Table 1. Elemental Analysis of Windermere Lake Sediment Core

Estimated Date	Core Depth (cm)	Aluminum	Boron	Barium	Beryllium	Calcium	Chromium	Copper	Iron	Manganese	Magnessium	Potassium	Sodium	Phosphorus	Sulphur	Silicon	Strontium	Titanium	Vanadium	Zinc
1998	0																			
1995	5																			
1995	10																			
1989	15																			
1989	20																			
1983	25																			
1983	30																			
1976	35	23200	200	329	2	108000	28	56	23700	568	12000	8650	500	600	15200	1200	238	213	30	48
1976	40																			
1970	45																			
1970	50																			
1970	55	26200	240	327	2	85600	44	64	25100	513	12000	9980	700	500	16100	1100	195	243	30	78
1970	60	26900	100	328	2	86300	10	48	26400	528	13000	10100	400	400	17800	1100	196	251	30	54
1970	65																			
1958	70																			
1958	75	34700	33	383	2.6	75800	38	41.9	33800	552	15600	13000	520	450	23000	1270	179	339	38	51.6
1958	80																			
1949	85																			
1949	90																			
1949	95	37400	38	400	2.4	60600	39.5	42.2	37700	571	15900	14300	510	440	25800	1340	151	366	41	51.8
1949	100	40600	42	424	2.5	57800	42.7	42.5	38900	583	16100	15900	560	440	27100	1360	147	402	44	58.7
1942	105																			
1942	110																			
1942	115	49600	52	492	2.7	41600	48.5	41.8	43500	627	17100	20300	670	430	32100	1540	122	510	54	55.4
1934	120																			
1934	125																			
1934	130																			
1934	135																			
1934	140	44600	47	460	2.5	34600	44.1	42.6	45000	631	17900	17900	480	400	32600	1460	104	446	49	57
1927	145																			
1927	150																			
1927	155																			
1927	160	51100	55	508	2.8	39100	51	41.3	44500	660	18800	21000	620	380	31900	1510	116	517	57	57.6
1918	165																			
1918	170																			
1918	175																			
1918	180	47600	51	487	2.7	44100	49.6	41.3	43900	716	17600	19400	580	370	32200	1400	126	469	52	56.9
1903	185																			
1903	190																			
1903	195																			
1903	200	47500	50	479	2.7	38100	45.4	42	45200	730	18000	19300	560	370	32800	1290	113	463	51	57
1877	205																			
1877	210																			
1877	215	50100	53	495	2.8	39300	47.6	42.6	44800	694	18300	20500	600	370	32100	1290	116	481	54	56.9
1877	220																			
1851	225																			
1851	230	48000	52	491	2.9	56100	45.4	42.4	42100	716	17400	19700	580	350	30200	1290	146	452	52	53
1851	235																			
1851	240																			
1851	245																			
1825	250	49400	53	501	3.1	62100	45.5	41.5	40700	729	17100	20600	620	340	29300	1290	155	474	54	50
1825	255																			
1825	260																			
1825	265																			
1799	270	59000	62	555	3.2	44700	55.1	40.6	41400	750	18600	24700	750	330	27000	1350	123	554	64	53.5
1799	275																			
1799	280																			
1773	285																			
1773	290																			
1773	295																			
1773	300	53800	53	513	2.7	25000	53.6	40.6	47600	832	18900	21400	750	380	33000	1710	92.4	532	58	64.2
1747	305																			
1747	310																			
1747	315																			
1747	320	53400	51	521	2.8	34700	56	41.6	46200	843	19500	21200	740	380	30500	1460	106	548	59	64.5
1721	325																			
1721	330																			
1721	335																			
1721	340	59100	53	580	3.5	60900	53.9	49	43100	682	19600	24100	1000	430	26800	1350	160	689	66	62.6

Core approximately 35 cm total depth, sampled on July 23, 1998. All concentrations in µg/g dry weight. No results for upper 70 mm due to insufficient sample.

APPENDIX I

Paleolimnological Analysis of Windermere Lake, B.C.
Final Report (March 1999).

Dr. Brian Cumming, Assistant Professor
Queen's University
Kingston, Ontario

Paleolimnological analysis of Windermere Lake, B.C -- Final Report (March 1999).

Contract Number - P.O. 812000

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List of Figures and Appendices:

Fig. 1 Summary of analyses for Windermere Lake.

Fig. 2 Stratigraphic distribution of diatom taxa in the
Windermere Lake core.

Appendix A: Summary of ^{210}Pb and LOI data, and diatom analyses.

Appendix B: Summary of data used in calculating ^{210}Pb dates and
 ^{210}Pb output.

Appendix C: Summary of relative abundances of diatom taxa in
Windermere Lake.

BACKGROUND

Windermere Lake was cored on July 23, 1998 by Rick Nordin and Les MacDonald. The core was retrieved using a modified K-B corer (internal diameter ~ 6.35 cm) from the deep basin. On shore the core was sectioned into 0.5-cm intervals into 120-ml plastic containers. These samples were shipped on ice to Queen's University where they were stored in our coldroom at 4°C. Each container was weighed to determine the total wet weight of sediment then subsampled for ^{210}Pb , diatom and pigment analyses. Twenty intervals (every 2 cm) were subsampled for diatom and pigment analyses, and sixteen intervals for ^{210}Pb analysis. Subsamples for analysis of pigments were sent to Prof. Leavitt at University of Regina. Prepared samples for ^{210}Pb analysis (see below) were sent to MYCORE Ltd.

METHODS

^{210}Pb Dating and Percent Organic Matter

The wet weight of the sediment was determined for all the subsections of the core. Sixteen subsamples of wet sediment from each core were weighed and oven-dried (24 hr at 105°C) and reweighed to determine percent water and dry weight of the sediment. Samples that were submitted for ^{210}Pb analysis were ground to a fine dust by use of a pestle and redried overnight at 105°C. The weight of this dried sediment was recorded to four

decimal places after it was put in a tared plastic digestion tube for determination of ^{210}Pb activity that was shipped to MYCORE Ltd.

Percent organic matter for each of the 16 ^{210}Pb samples was determined using standard loss-on-ignition methods (Dean, 1974). A known quantity of dried sediment (recorded to four decimal places) was heated to 550°C for 2 hours. The difference between the original weight of the sediment and the weight of sediment remaining after ignition was used to estimate the percent of organic matter in each sediment sample.

^{210}Pb activities were estimated from determination of 209-Po and a tracer of known activity by alpha spectroscopy. Unsupported ^{210}Pb is calculated by subtracting supported ^{210}Pb (the baseline activity determined from bottom samples of the core) from the total activity at each level. The sediment chronology and sedimentation rates were calculated using the constant rate of supply (CRS) model (Appleby and Oldfield, 1978) from the estimates of ^{210}Pb activities and estimates of cumulative dry mass (Binford, 1990). See Appendix B for summaries of ^{210}Pb analyses by MYCORE (B-1), summary of ^{210}Pb calculations (B-1,2), and output from the CRS model (B-3).

Diatom Preparation and Enumeration

Slides for diatom analysis were prepared using standard techniques (Cumming, Wilson, Smol and Hall, 1995). Briefly, a small amount of wet sediment was suspended in a 50:50 (molar) mixture of sulfuric and nitric acid in a 20-ml glass vial for 24 hr. prior to being submersed at 70°C in a hot water bath for 5 hr. The remaining sediment material was settled for a period of 24 hr, at which time the acid above sample was removed. The sample was rinsed with distilled water and allowed to settle once again for 24 hrs. The procedure was repeated approx. 10 times until the sample was acid free (litmus test). The samples were settled onto coverslips in a series of four 100% dilutions, which when dry, were mounted onto glass slides using a high-resolution mounting media called Naphrax[®]. For each sample, at least 400 diatom taxa were enumerated under oil immersion at 1000X magnification using an objective with a numerical aperture of 1.3. These analyses were based on the references of Krammer and Lange-Bertalot (1986, 1988, 1991a,b), Patrick and Reimer (1966, 1975) and Cumming et al. (1995).

Cluster Analysis

Cluster analysis, which groups similar diatom assemblages, was run on the taxa represented in Fig. 2. Cluster analysis on the diatom assemblage composition was constrained to the depth of the core samples to provide an unbiased assessment of changes in diatom assemblages through time. A squared chord was used to determine similarity between samples in the cluster analysis (Fig.2). Zones were placed based on these analyses to represent distinct groups in diatom assemblages through time (dashed line on Fig. 2).

Environmental Reconstructions from diatom assemblages

Inferences of total phosphorus downcore were based on a total phosphorus model based on the 111 freshwater lakes from the 219 lakes sampled by Wilson, Cumming & Smol (1996). This model is based on estimates of taxa optima from weighted-averaging regression on non-transformed relative percentage data. The coefficient of determination (r^2) of this model is 0.66, and the jackknifed r^2 is 0.47. This model is superior to the earlier models developed by Reavie, Hall & Smol (1995) for several reasons including its better predictive ability and the larger number of samples which provide more analogs for downcore reconstructions.

The total phosphorus inferences (Fig. 1E) were critically assessed to determine: 1) if they tracked the main direction of variation in the diatom species assemblages (Fig. 1D); and 2) to assess if the assemblages encountered downcore are well represented in the modern-day samples (Fig. 1F). If the diatom-based phosphorus reconstruction matches the main direction of variation in the diatom assemblages downcore, then we can be fairly confident that the diatoms are tracking changes correlated to phosphorus. If the directions of variation do not match, then the diatom-inferred phosphorus reconstructions do not fully represent the changes, if any occurred, in diatom species composition downcore. Presumably, other environmental variables, or interactions between environmental variables, are contributing to the changes in diatom assemblages.

The main direction of variation in the diatom assemblages downcore in Windermere Lake was determined from the first axis scores from a principal components analysis (PCA) ordination using a co-variance matrix and non-transformed species abundance data. A PCA was chosen to represent the main direction of variation in this core since only minor changes in diatom assemblages occurred and thus a linear ordination technique would more effectively capture changes in this core than an approach based on unimodal techniques.

The reliability of the downcore total phosphorus inferences assumes that the diatom assemblages encountered downcore are well represented in our modern diatom assemblages. To determine if appropriate analogs existed for the core samples, we determined which samples in our present-day dataset of 111 lakes most resembled each of the downcore samples. This determination was based on a squared chord dissimilarity coefficient between all species found in the core samples. The best match between downcore and modern samples was compared with the distribution of best match between modern samples. Any downcore sample that was more dissimilar than 80% of the modern distribution were deemed to be a 'poor analog'. Similarly, any downcore sample that was more dissimilar than 95% of the modern distribution were deemed to have 'no analog' in our present-day dataset. If the downcore assemblages have good representation in modern samples, more confidence can be placed in the reconstruction. If modern analogs do not exist or are poor, then caution must be placed in

reconstructions from these downcore samples.

RESULTS AND DISCUSSION

²¹⁰Pb Profile, Sedimentation Rates and Organic Matter

The ²¹⁰Pb profile from Windermere Lake shows a slight flattening in activity in the top 2 cms of the core (Fig. 1A). The two most likely explanations for this pattern in the ²¹⁰Pb profile are: 1) an increase in sedimentation rate that could cause a dilution of the ²¹⁰Pb activity; or 2) slight sediment mixing. Small changes in the percent organic matter and diatom assemblages in the top few centimeters do not allow us to rule out slight mixing in the uppermost 2 cm of this core. Sedimentation rates in general are quite slow resulting in the short core representing a fairly long span of time, with estimated bottom dates of the early 1700s. The inferred increase in sedimentation rate from approximately 1920 to 1960 (Fig. 1B) corresponds to a few percent increase in percent organic matter (Fig. 1C, * note the different time scales). Increases in organic matter can be attributed to several factors including increased in-lake production of organic matter, increased inwash of organic matter, or decreases in the load of inorganic sediment to the lake. Historical information on the development of Windermere Lake may provide insight to these patterns.

Diatom Assemblage Changes and Analyses

Approximately 85 diatom taxa were encountered in the sediment core from Windermere Lake (Appendix C-1). The assemblages are dominated by a diversity of benthic taxa, such as *Fragilaria brevistriata*, *Fragilaria construens*, *Fragilaria pinnata*, *Navicula vitabunda*, and *Navicula* cf. *minuscula*. The latter two *Navicula* taxa only occur in the modern-day B.C. dataset at low abundances (6% or less) and thus not a lot is known about the ecological preferences of these taxa. The rest of the assemblage consist primarily of other benthic taxa in the genera *Amphora*, *Gomphonema*, *Navicula*, *Nitzschia*, and the ephiphytic taxa (live attached to macrophytes) in the genera *Achnanthes*. Planktonic taxa are very rare, with *Cyclotella gamma* being the only taxon to reach greater than 2%.

Cluster analysis suggests the changes in diatom assemblages through time can be divided into two primary zones (Fig. 2). The difference in zones A and B is the loss of *Nitzschia bacillum*, *Navicula diluviana*, and *Achnanthes rosenstockii* in Zone A, and the increase or appearance of other taxa such as *Navicula* cf. *minuscula*, *Cyclotella gamma*, *Navicula modica*, and *Navicula schaderi* in Zone A. The latter three taxa are not in the modern-day B.C. dataset, and thus little is known of their ecological requirements in this region.

The changes that have occurred in the diatom assemblages are quite minor and this is reflected in the small changes seen in the inferred total phosphorus (TP). TP has remained relatively stable since the 1700s, with a small increase in TP

since approximately 1950 (Fig. 1E). However, these latter changes are being largely driven by the increases in *Navicula* cf. *minuscule*, which we have poor analogs (Fig. 1F), and thus the TP estimates must be viewed as preliminary at this time.

PCA axis 1 scores (Fig. 1D) accounts for 66% of the variation in diatom taxa in this core. The correlation between the reconstructed TP and the main direction of variation in taxa (represented by PCA axis 1 scores) is 0.75 when all 20 points are considered, which suggests that the changes seen in the diatom assemblages may be related to changes in total phosphorus.

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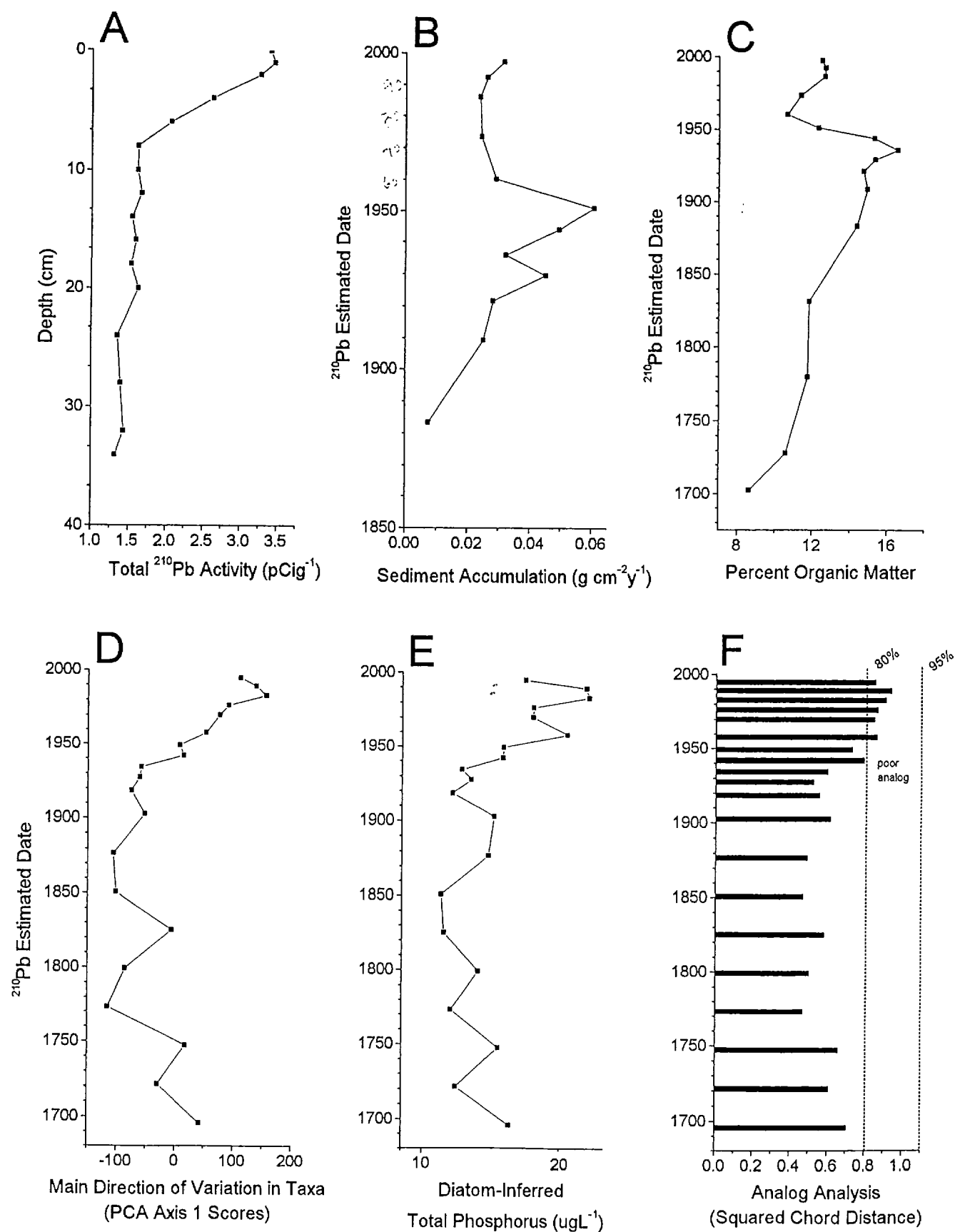
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Figure Captions

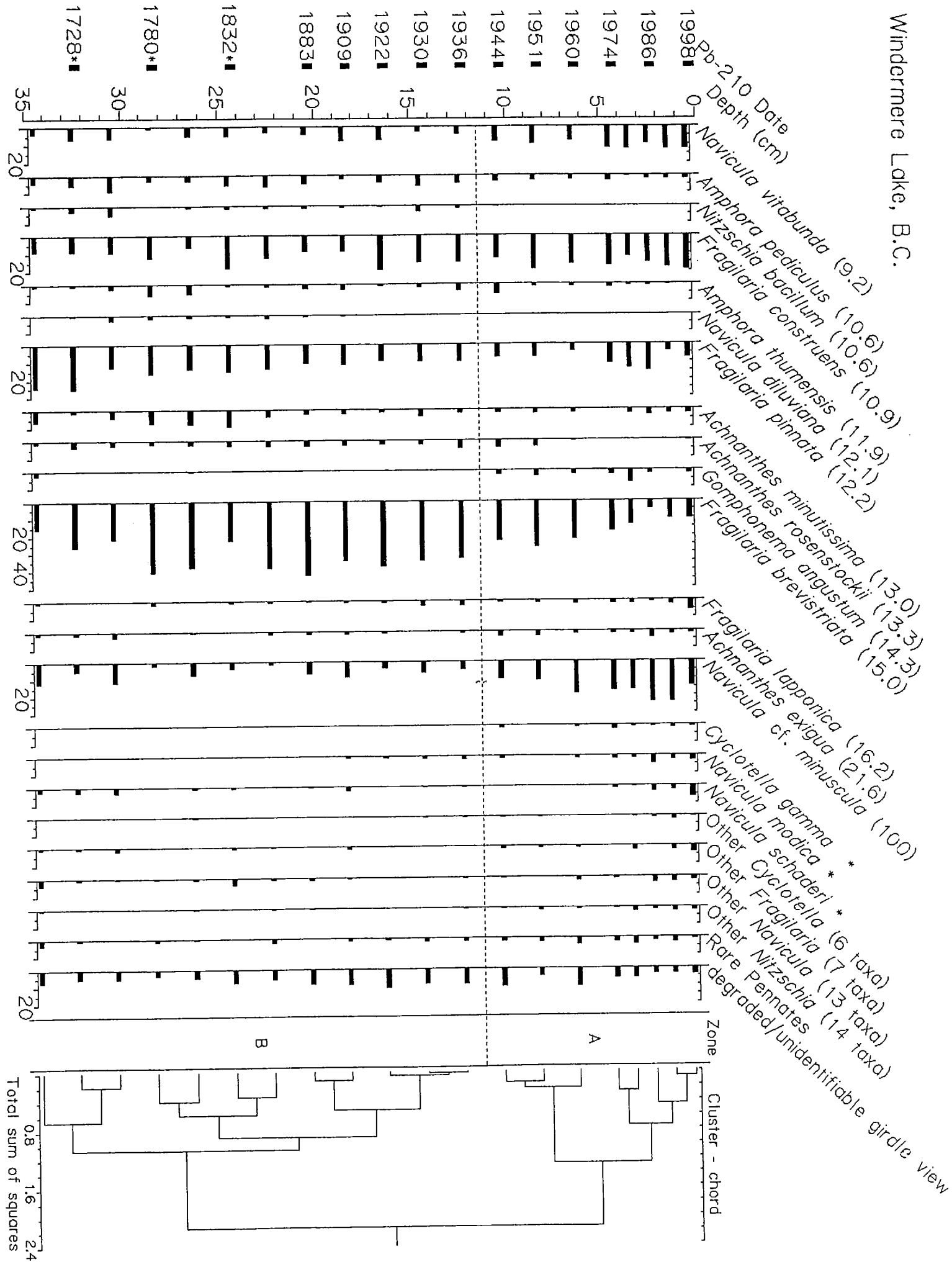
Figure 1. Summary diagram for the sediment core from Windermere lake showing: A) total ^{210}Pb activity from which the chronology of the core is based; B) the sediment accumulation rate; C) the change in the percent of organic matter in the core; D) the main direction of variation in the diatom assemblage data; E) diatom-based estimated late-summer total phosphorus; and F) analog analysis showing the dissimilarity between present-day and downcore samples (any sample that has a squared chord distance > 0.8 was determined to be a poor analog, whereas any sample with a squared chord distance greater than 1.1 was determined to have no analog in the modern dataset).

Figure 2. Diatom stratigraphy of the diatom taxa that were present in at least ~2% relative abundance in the sediment core from Windermere Lake, B.C. (see Appendix C for a complete list of taxa and the relative percentage data). The diatom taxa are arranged in order of increasing late-summer total phosphorus (TP) optima which is indicated in parentheses for those taxa with known optima. Those taxa without known optima are marked with an asterisk. The dotted line separates the stratigraphy into two zones that were identified by a cluster analysis on the diatom assemblage composition that was constrained to the depth of the core samples. ^{210}Pb dates with an asterisk are based on extrapolation of the sedimentation rates of the bottom two dates with unsupported ^{210}Pb activity.

Windermere Lake, British Columbia



Windermere Lake, B.C.



Pb210 and LOI summary

INTTOP (cm)	INTBOT (cm)	Pb210AC (pCi/g)	LOI(550C %organic)	Estimated AD-DATE	SED RATE (g/cm2/yr)
0	0.5	3.4156	12.51	1997.6	0.0314
1	1.5	3.4618	12.70	1992.5	0.0262
2	2.5	3.2746	12.65	1986.3	0.0238
4	4.5	2.6336	11.37	1973.6	0.0242
6	6.5	2.0683	10.65	1960.3	0.0290
8	8.5	1.6233	12.31	1951.1	0.0606
10	10.5	1.6210	15.30	1944.1	0.0493
12	12.5	1.6700	16.56	1936.1	0.0320
14	14.5	1.5467	15.36	1929.7	0.0451
16	16.5	1.5912	14.72	1921.8	0.0280
18	18.5	1.5404	14.92	1909.4	0.0249
20	20.5	1.6284	14.37	1883.5	0.0074
24	24.5	1.3529	11.82	1831.7	
28	28.5	1.3965	11.76	1780.0	
32	32.5	1.4338	10.58	1728.2	
34	34.5	1.3170	8.62	1702.3	

Diatom analyses summary

Depth (c ₀) TOP	Depth (c ₁) BOTTOM	Estimated AD-DATE	log TP	TP(μg L ⁻¹)	LogSalinit	salinity mg L ⁻¹	PCA AX1	11 lakes minimum sq. chord
0.5	1	1995.1	1.223	16.71	1.949	88.92	110	0.8496
1.5	2	1989.4	1.358	22.80	1.945	88.10	137	0.9345
2.5	3	1983.1	1.364	23.12	1.91	81.28	155	0.9039
3.5	4	1976.8	1.242	17.46	1.966	92.47	90	0.8628
4.5	5	1970.3	1.241	17.42	1.939	86.90	75	0.8446
6.5	7	1958.0	1.316	20.70	1.969	93.11	51	0.8609
8.5	9	1949.3	1.175	14.96	1.971	93.54	7	0.7276
10.5	11	1942.1	1.174	14.93	1.976	94.62	13	0.7895
12.5	13	1934.5	1.084	12.13	1.983	96.16	-59	0.5985
14.5	15	1927.7	1.104	12.71	2.006	101.39	-62	0.5229
16.5	17	1918.7	1.064	11.59	1.974	94.19	-76	0.5544
18.5	19	1902.9	1.155	14.29	1.985	96.61	-53	0.6127
20.5	21	1877.0	1.143	13.90	2.026	106.17	-106	0.4899
22.5	23	1851.1	1.04	10.96	2.014	103.28	-102	0.4675
24.5	25	1825.3	1.046	11.12	2.003	100.69	-7	0.579
26.5	27	1799.4	1.121	13.21	2.033	107.89	-86	0.4996
28.5	29	1773.5	1.061	11.51	2.029	106.91	-116	0.4681
30.5	31	1747.6	1.165	14.62	1.978	95.06	17	0.6568
32.5	33	1721.7	1.072	11.80	1.952	89.54	-30	0.608
34.5	35	1695.9	1.189	15.45	1.993	98.40	42	0.7037

SUMMARY PB210 ANALYSES BY MYCORE - WINDERMERE

Sample Number	Disk #	Section of Cor	Sample Weight used	209 Po Counts	210 Po Counts	210 Po Meas	210 Po (Bq/g)	Precision 1 STD (%)
		Top	Bottom					
Windermere Lake								
17	791	0	0.5	1255	1836	398	0.098	5.5
18	792	1	1.5	1358	3311	807	0.106	3.9
19	793	2	2.5	1304	4928	1113	0.100	3.3
20	794	4	4.5	1392	6628	1318	0.084	3.0
21	795	6	6.5	1298	5755	855	0.066	3.7
22	796	8	8.5	1403	7051	907	0.052	3.5
23	797	10	10.5	1321	5595	663	0.052	4.1
24	798	12	12.5	1208	12653	1427	0.054	2.8
25	799	14	14.5	1215	6137	635	0.050	4.2
26	800	16	16.5	1337	5474	651	0.053	4.1
27	801	18	18.5	1345	5129	600	0.051	4.3
28	802	20	20.5	1295	6239	758	0.054	3.8
29	803	24	24.5	1406	5094	582	0.045	4.4
30	804	28	28.5	1431	4036	460	0.046	4.9
31	805	32	32.5	1446	2160	258	0.047	6.6
32	806	34	34.5	1495	6185	698	0.044	4.0

SUMMARY PB210 CALCULATIONS FOR DETERMINING DATES - WINDERMERE

Back calculate to coring (KPL)									
Section of Core		Date Po Sample		Date of coring		Date of coring		Time since coring (days)	
Top	Bottom								
Windermere Lake									
0	0.5	98	12	23	98	7	23	153	0.125
1	1.5	98	12	23	98	7	23	153	0.126
2	2.5	98	12	23	98	7	23	153	0.120
4	4.5	98	12	23	98	7	23	153	0.096
6	6.5	98	12	23	98	7	23	153	0.076
8	8.5	98	12	23	98	7	23	153	0.059
10	10.5	98	12	23	98	7	23	153	0.059
12	12.5	98	12	23	98	7	23	153	0.051
14	14.5	98	12	23	98	7	23	153	0.056
16	16.5	98	12	23	98	7	23	153	0.056
18	18.5	98	12	23	98	7	23	153	0.056
20	20.5	98	12	23	98	7	23	153	0.059
24	24.5	98	12	23	98	7	23	153	0.049
28	28.5	98	12	23	98	7	23	153	0.051
32	32.5	98	12	23	98	7	23	153	0.052
34	34.5	98	12	23	98	7	23	153	0.048

CALCULATIONS FOR INPUT INTO BINFORD PROGRAM

BINFORD FILE INPUTS FOR CALCULATION OF DATES AND SEDIMENTATION RATES

Windermere-Pb210

Windermere

C1

16

0.0509

Back calculated to coring

		Pb-210		Std dev		Pb210		Std dev		Rho		INTTOP		INTBOT		Pb210		Pb210		Rho		OM		CUMTOP		CUMBOT		std	
		activity		activity		activity		activity		activity		activity		activity		activity		activity		activity		proportion		activity		activity		activity	
		(Bq/g)		(Bq/g)		(pCi/g-1)		(pCi/g-1)		(g cm-3)		(cm)		(cm)		(pCi/g-1)		(pCi/g-1)		(g cm-3)				(g cm-2)		(g cm-2)		(pCi/g-1)	
		activity		activity		activity		activity		activity		activity		activity		activity		activity		activity		activity		activity		activity		activity	
		(Bq/g)		(Bq/g)		(pCi/g-1)		(pCi/g-1)		(g cm-3)		(cm)		(cm)		(pCi/g-1)		(pCi/g-1)		(g cm-3)		proportion		(g cm-2)		(g cm-2)		(pCi/g-1)	
0	0.5	0.1264	0.0071	3.4156	0.1918	0.1256	0.1643	0.0000	0.125	0.0000	0.0628	0.1918	0.125	0.0000	0.0628	0.1918	0.125	0.0000	0.0628	0.1918	0.125	0.0000	0.0628	0.1918	0.125	0.0000	0.0628	0.1918	0.125
1	1.5	0.1281	0.0053	3.4618	0.1420	0.1643	0.1456	1.00	0.127	0.1455	0.2276	0.1420	0.127	0.1455	0.2276	0.1420	0.127	0.1455	0.2276	0.1420	0.127	0.1455	0.2276	0.1420	0.127	0.1455	0.2276	0.1420	0.127
2	2.5	0.1212	0.0041	3.2746	0.1121	0.1456	0.1585	2.00	0.127	0.3032	0.3760	0.1121	0.127	0.3032	0.3760	0.1121	0.127	0.3032	0.3760	0.1121	0.127	0.3032	0.3760	0.1121	0.127	0.3032	0.3760	0.1121	0.127
4	4.5	0.0974	0.0032	2.6336	0.0856	0.1585	0.1957	4.00	0.114	0.5894	0.6687	0.0856	0.114	0.5894	0.6687	0.0856	0.114	0.5894	0.6687	0.0856	0.114	0.5894	0.6687	0.0856	0.114	0.5894	0.6687	0.0856	0.114
6	6.5	0.0765	0.0030	2.0683	0.0806	0.1957	0.2187	6.00	0.106	0.9454	1.0432	0.0806	0.106	0.9454	1.0432	0.0806	0.106	0.9454	1.0432	0.0806	0.106	0.9454	1.0432	0.0806	0.106	0.9454	1.0432	0.0806	0.106
8	8.5	0.0601	0.0024	1.6233	0.0638	0.1615	0.2187	8.00	0.123	1.3374	1.4182	0.0638	0.123	1.3374	1.4182	0.0638	0.123	1.3374	1.4182	0.0638	0.123	1.3374	1.4182	0.0638	0.123	1.3374	1.4182	0.0638	0.123
10	10.5	0.0600	0.0018	1.6210	0.0486	0.1160	0.2187	10.00	0.153	1.6586	1.7680	0.0486	0.153	1.6586	1.7680	0.0486	0.153	1.6586	1.7680	0.0486	0.153	1.6586	1.7680	0.0486	0.153	1.6586	1.7680	0.0486	0.153
12	12.5	0.0618	0.0018	1.6700	0.0486	0.1160	0.2187	12.00	0.166	1.9876	2.0456	0.0486	0.166	1.9876	2.0456	0.0486	0.166	1.9876	2.0456	0.0486	0.166	1.9876	2.0456	0.0486	0.166	1.9876	2.0456	0.0486	0.166
14	14.5	0.0572	0.0025	1.5467	0.0677	0.1207	0.1599	14.00	0.154	2.2424	2.3028	0.0677	0.154	2.2424	2.3028	0.0677	0.154	2.2424	2.3028	0.0677	0.154	2.2424	2.3028	0.0677	0.154	2.2424	2.3028	0.0677	0.154
16	16.5	0.0589	0.0027	1.5912	0.0721	0.1599	0.1599	16.00	0.147	2.5130	2.5929	0.0721	0.147	2.5130	2.5929	0.0721	0.147	2.5130	2.5929	0.0721	0.147	2.5130	2.5929	0.0721	0.147	2.5130	2.5929	0.0721	0.147
18	18.5	0.0570	0.0027	1.5404	0.0729	0.1685	0.1685	18.00	0.149	2.8133	2.8976	0.0729	0.149	2.8133	2.8976	0.0729	0.149	2.8133	2.8976	0.0729	0.149	2.8133	2.8976	0.0729	0.149	2.8133	2.8976	0.0729	0.149
20	20.5	0.0603	0.0025	1.6284	0.0673	0.1786	0.1786	20.00	0.144	3.1241	3.2135	0.0673	0.144	3.1241	3.2135	0.0673	0.144	3.1241	3.2135	0.0673	0.144	3.1241	3.2135	0.0673	0.144	3.1241	3.2135	0.0673	0.144
24	24.5	0.0501	0.0025	1.3529	0.0665	0.1951	0.1951	24.00	0.118	3.8423	3.9399	0.0665	0.118	3.8423	3.9399	0.0665	0.118	3.8423	3.9399	0.0665	0.118	3.8423	3.9399	0.0665	0.118	3.8423	3.9399	0.0665	0.118
28	28.5	0.0517	0.0029	1.3965	0.0779	0.2088	0.2088	28.00	0.106	4.6394	4.7438	0.0779	0.106	4.6394	4.7438	0.0779	0.106	4.6394	4.7438	0.0779	0.106	4.6394	4.7438	0.0779	0.106	4.6394	4.7438	0.0779	0.106
32	32.5	0.0530	0.0040	1.4338	0.1073	0.2424	0.2424	32.00	0.106	5.5627	5.6839	0.1073	0.106	5.5627	5.6839	0.1073	0.106	5.5627	5.6839	0.1073	0.106	5.5627	5.6839	0.1073	0.106	5.5627	5.6839	0.1073	0.106
34	34.5	0.0487	0.0023	1.3170	0.0609	0.3402	0.3402	34.00	0.086	6.2944	6.4645	0.0609	0.086	6.2944	6.4645	0.0609	0.086	6.2944	6.4645	0.0609	0.086	6.2944	6.4645	0.0609	0.086	6.2944	6.4645	0.0609	0.086

avg 1.375035 = supported Pb210
stds 0.050898 1.476831

YOU ARE ANALYZING CORE C1

FROM LAKE Windermere

THE DATA ARE:

INTTOP	INTBOT	PB210ACT	UNSUPACT	RHO	PERCORG	CUMMASST	CUMMASSB	SDACT
0.0	0.5	3.41560	2.04060	0.12560	0.120	0.0000	0.0628	0.1918
1.0	1.5	3.46180	2.08680	0.16430	0.120	0.1455	0.2276	0.1420
2.0	2.5	3.27460	1.89950	0.14560	0.120	0.3032	0.3760	0.1121
4.0	4.5	2.63360	1.25860	0.15850	0.110	0.5894	0.6687	0.0856
6.0	6.5	2.06830	0.69330	0.19570	0.100	0.9454	1.0432	0.0806
8.0	8.5	1.62330	0.24830	0.16150	0.120	1.3374	1.4182	0.0638
10.0	10.5	1.62100	0.24600	0.21870	0.150	1.6586	1.7680	0.0724
12.0	12.5	1.67000	0.29500	0.11600	0.160	1.9876	2.0456	0.0486
14.0	14.5	1.54670	0.17160	0.12070	0.150	2.2424	2.3028	0.0677
16.0	16.5	1.59120	0.21610	0.15990	0.140	2.5130	2.5929	0.0721
18.0	18.5	1.54040	0.16540	0.16850	0.140	2.8133	2.8976	0.0729
20.0	20.5	1.62840	0.25340	0.17860	0.140	3.1241	3.2135	0.0673
24.0	24.5	1.35290	0.00000	0.19510	0.110	3.8423	3.9399	0.0665
28.0	28.5	1.39650	0.00000	0.20880	0.110	4.6394	4.7438	0.0779
32.0	32.5	1.43380	0.00000	0.24240	0.100	5.5627	5.6839	0.1073
34.0	34.5	1.31700	0.00000	0.34020	0.080	6.2944	6.4645	0.0609

STANDARD DEVIATION OF SUPPORTED PB-210 = 0.0509

Pb-210 dates for Lake Windermere

core C1

INTTOP	INTBOT	MIDINT	TTOP	SDTTOP	TBOT	SDTBOT	SEDRATE	SDSEDRT	SUMTOP
0.0	0.5	0.2	0.00	1.05	2.00	1.07	0.0314	0.0100	2.1214
1.0	1.5	1.2	4.49	1.11	7.62	1.15	0.0262	0.0081	1.8445
2.0	2.5	2.2	10.73	1.22	13.80	1.28	0.0238	0.0074	1.5188
4.0	4.5	4.2	23.31	1.56	26.59	1.66	0.0242	0.0084	1.0266
6.0	6.5	6.2	36.60	2.12	39.99	2.28	0.0290	0.0120	0.6786
8.0	8.5	8.2	46.86	2.71	48.19	2.78	0.0606	0.0255	0.4931
10.0	10.5	10.2	53.33	3.15	55.54	3.26	0.0493	0.0243	0.4031
12.0	12.5	12.2	61.62	3.83	63.44	4.02	0.0320	0.0165	0.3113
14.0	14.5	14.2	68.18	4.59	69.51	4.73	0.0451	0.0271	0.2539
16.0	16.5	16.2	75.37	5.56	78.22	5.95	0.0280	0.0201	0.2029
18.0	18.5	18.2	87.51	7.72	90.91	8.35	0.0249	0.0219	0.1390
20.0	20.5	20.2	108.96	14.16	121.22	20.13	0.0074	0.0123	0.0713

Execution terminated : 0

C:\pb210>

APPENDIX II

Analyses of Fossil Pigments from Windermere Lake
March 30, 1999

Dr. Peter Leavitt, Associate Professor
University of Regina
Regina, Saskatchewan

From: Peter Leavitt
To: Gordon Kan, Les McDonald
Subject: Final Report for Fossil Pigment Analyses
Date: 30 March 1999

Title: Analyses of fossil pigments from Windermere Lake

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LIST OF FIGURES AND APPENDICES

- Fig 1. Fossil pigment concentrations (nmoles g⁻¹ organic matter) in short core from Windermere Lake. Pigment names and taxonomic affinities indicated for each profile.
- Appendix 1. Organic-matter specific concentrations (nmoles pigment/g organic matter) of fossil pigments from Windermere Lake.

BACKGROUND

Windermere Lake was cored in July 1998 by Rick Nordin and Les MacDonald using standard gravity corer procedures. The core was taken from the deepest point in the basin of Windermere Lake. At lakeside, cores were sectioned into 0.5-cm intervals into whirlpak bags. Samples were shipped on ice to Dr. Brian Cumming, Department of Biology, Queen's University, Kingston, Ontario. Samples were stored at 4°C until further analysis. In Fall 1998, Dr. Cumming forwarded subsamples of sediment to the University of Regina where they were stored at -12°C until analysis. Aliquots of each sample were obtained for analysis of organic content (as weight loss on ignition at 500°C for 1 hr) and phototrophic pigments (by liquid chromatography).

METHODS

Sediment Chronology

At the time of report preparation, no information was available concerning the procedures used for determination of sediment age. Consequently, all stratigraphic patterns are reported relative to burial depth (in cm). Normally, ^{210}Pb activity is used to estimate core chronology. Inquiries about sediment chronology should be forwarded to either: Dr. Gordon Kan, Environment Canada, Pacific Environmental Science Centre, 2645 Dollarton Hwy, North Vancouver, BC, V7H 1V2, or; Dr. Brian Cumming, Dept Biology, Queen's University, Kingston, ON, K7L 3N6.

Pigment Analyses

Sediments from each sample were freeze-dried under a hard vacuum (0.1 Pa) for 48 hrs using a VirTis model 24Dx48 lyophilizer to remove excess water prior to extraction of lipid-soluble carotenoids, chlorophylls (Chls) and their derivatives. Dried sediments were stored at -80°C in the dark under an inert nitrogen gas atmosphere following drying to

prevent pigment oxidation.

Carotenoids, chlorophylls (Chls) and their derivatives were extracted (18 h, 4°C, dark, under N₂) from freeze-dried sediments using a standard mixture of acetone:methanol:water (80:15:5, by vol.; Leavitt et al. 1989). Sediment residues were exhaustively extracted with three aliquots of solvent mixture, extracts filtered (0.22 µm Acropore membrane), and solvents evaporated in the dark using N₂ gas. Dried extracts were stored at -20°C under N₂ in the dark until pigment analysis. Just prior to quantification, pigments were brought to room temperature in the dark and dissolved in a precisely known volume of injection solvent (acetone:ion-pairing reagent:methanol; 70:25:5, by volume) containing 3.2 mg litre⁻¹ Sudan II (Sigma Chemical Co.). This chromatographic dye is an internal standard that has carotenoid-like absorption characteristics ($\lambda_{\text{max}} = 485, 442.5$ nm in acetone), runs at a unique position on the chromatogram, and is used to correct for dilution, injection, and chromatographic errors (Leavitt and Findlay 1994). Ion-pairing reagent (IPR) consists of 0.75 g tetrabutyl ammonium acetate and 7.7 g ammonium acetate in 100 ml distilled water.

Concentrations of fossil pigments are quantified using a Hewlett Packard (HP) model 1040 HPLC system with HP model 1040 photo-diode array detector and model 1046A fluorescence detector following the reversed-phase liquid chromatography procedure of Leavitt et al. (1989). Briefly, analytical separation was achieved by isocratic delivery (1.5 ml min⁻¹; 21,000 kPa) of a mobile phase A (10% IPR in methanol) for 1.5 min, a linear succession to 100% solution B (27% acetone in methanol) over 7 min, and isocratic hold for 12.5 min. Rainin C-18 columns (10 cm, 5 µm particles) were re-equilibrated by continued isocratic delivery for 3 min, a linear return to 100% A over 3 min, and isocratic supply for a final 4 min.

Pigments isolated from sediments were compared to those from unialgal cultures of known pigment composition (Leavitt et al. 1989; Leavitt and Findlay 1994) and, for some compounds, with authentic standards provided by the US Environmental Protection Agency

(US-EPA). Spectral characteristics and chromatographic mobility were used to identify pigments from sediments (Leavitt et al. 1989). Analysis of fossil pigments was restricted to carotenoids characteristic of cryptophytes (alloxanthin), mainly diatoms (diatoxanthin), diatoms with chrysophytes and some dinoflagellates (fucoxanthin), diatoms and dinoflagellates (diadinoxanthin), chlorophytes and cyanobacteria (lutein-zeaxanthin), cyanobacteria (echinenone), colonial cyanobacteria (myxoxanthophyll), colonial cyanobacteria from the group Nostocales (canthaxanthin) and N_2 -fixing cyanobacteria (aphanizophyll), as well as the major a, b, and c-phorbins. Chlorophyll (Chl) b and pheopigment derivatives (mainly pheophytin b) were used to distinguish green algae from cyanobacteria, whose carotenoid zeaxanthin was not separated from the chlorophyte pigment lutein on our HPLC system. Similarly, chromatographic peaks from aphanizophyll (Aphanizomenon), oscillaxanthin (Oscillatoriaceae), and 4-keto-myxoxanthophyll (Anabaena) were incompletely resolved and were reported as aphanizophyll. Compounds tentatively identified as originating from purple sulphur phototrophic bacteria (okenone) were also noted in Windermere Lake sediments. Unfortunately, insufficient material was provided for more complete characterization, and interpretation of these compounds is regarded as tentative. Further details of pigment distribution among algal groups are provided by Davies (1976) and Goodwin (1980).

Pigment concentrations were expressed as nmoles pigment g^{-1} organic matter, as recommended by Leavitt (1993) and Leavitt and Findlay (1994). Organic matter contents of freeze-dried sediments were determined as weight loss on ignition for 1 hr at 500°C (Leavitt and Findlay 1994). Comparison of 20 years of phytoplankton data with annual fossil records has demonstrated that organic matter-specific concentrations are linearly correlated to algal biomass for a wide variety of fossil carotenoids, particularly for algae that are abundant during the ice-free season (Leavitt and Findlay 1994).

All profiles were interpreted following recommendations by Leavitt (1993), Leavitt and Findlay (1994), and Leavitt et al. (1994). Copies of these publications are available on request. In brief, organic-matter specific pigment concentration are interpreted as indices of

the past biomass of the main algal groups in the lakes, including, but not limited to total algae, diatoms, cryptophytes, greens, cyanobacteria, and colonial (surface bloom-forming) cyanobacteria. Species-level information is not available from the analysis of fossil pigments. Instead, analyses summarized past algal community change and lake production in the functional units most important to lake managers (as above). Data are most useful for documenting historical changes in algal abundance within individual lakes, rather than differences in production among lakes (see Leavitt 1993 for further information).

In this report, we will provide brief analyses of organic-matter specific fossil concentration as a metric of lake production. However, as we have demonstrated elsewhere (Leavitt 1993, Leavitt and Findlay 1994), absolute concentrations of pigments can be strongly influenced by the morphometry of the lake (depth, light penetration, deepwater oxygen content). Consequently, stratigraphic changes are most reliably interpreted as changes in individual pigment abundance relative to the historical maximum or minimum exhibited for that compound, rather than as a function of other compounds (e.g., alloxanthin interpreted relative to the history of alloxanthin alone).

RESULTS and DISCUSSION

Windermere Lake

Sediments of Windermere Lake were characterized by relatively low organic matter content (~15% of dry weight) throughout the 35 cm core (Fig. 1). Low organic matter content may reflect unproductive lake conditions (Leavitt 1993), or may result from dilution of autochthonous organic matter (i.e., originating from within the lake) by inorganic matter from the lake catchment (Rowan et al. 1992). In general, the organic matter content of Windermere Lake sediments increased steadily from a minimum of ~12% at the bottom of the core to a maximum of ~20% at 10 cm depth, before declining sharply to stable values of 13-14% in the uppermost 8 cm of sediments.

Organic-matter specific concentrations of most carotenoids were intermediate to minimum values recorded in unproductive alpine lakes and high concentrations characteristic of eutrophic systems (e.g., Leavitt and Findlay 1994, Vinebrooke et al. 1998). Pigment preservation was good throughout the core, with high molecular ratios of undegraded Chl *a*:pheophytin *a* (>2:1) at most depths, moderate concentrations of labile pigments (Chl *a*, fucoxanthin), and the presence of okenone from phototrophic bacteria, organisms characteristic of environments with low oxygen contents and optimal pigment preservation. However, unlike some lakes, extremely labile carotenoids and chlorophylls (e.g., Chl *c* from diatoms, chrysophytes, dinoflagellates) were not well preserved. Together, these observations suggest that while some pigment degradation occurred during deposition, fossil stratigraphies were reliable and could be used to infer historical changes in the production of Windermere Lake.

Analysis of fossil pigments suggested that total algal production has increased only modestly through the period of time encompassed by the Windermere sediment core (Fig. 1). Concentrations of ubiquitous pigments (β -carotene, Chl *a*, pheophytin *a*) were stable between 10 and 35 cm burial depth, increased slightly between 5 and 10 cm depth, and exhibited core-wide maxima in the uppermost 3 cm of sediments. The observation that both chemically-stable β -carotene and easily-degraded Chl *a* exhibited similar stratigraphic patterns suggests that changes in fossil pigment concentrations reflected historical variations in algal abundance rather than artifacts of selective pigment preservation or deposition.

Fossil pigment analyses suggested that the original algal communities were composed of similar abundances of diatoms and chlorophytes (green algae), with lesser amounts of cryptophytes and bloom-forming cyanobacteria (Fig. 1). Although pigments from colonial cyanobacteria were present throughout the core (e.g., myxoxanthophyll, echinenone; canthaxanthin not shown), concentrations in deep sediments (>15 cm) were only 20% those recorded for eukaryotic algae such as diatoms (diatoxanthin) and chlorophytes (lutein-zeaxanthin).

Concentrations of most indicator pigments increased modestly over the course of the sediment core. For example, pigments from green algae (lutein-zeaxanthin, Chl b, pheophytin b), cyanobacteria (echinenone, myxoxanthophyll, canthaxanthin) and cryptophytes (alloxanthin) increased two-fold to maxima in the uppermost 3 cm of the Windermere Lake core. Similarly, fucoxanthin from diatoms, chrysophytes and dinoflagellates increased more than 500% above background levels in the uppermost 10 cm of the core, whereas diatoxanthin from diatoms was inferred to have declined ~25% over the same period. Because sedimentary fucoxanthin content increased, whereas that of diatoxanthin declined slightly, we interpret that recent algal communities have experienced a recent increase in the relative abundance of either chrysophytes or dinoflagellates at the expense of diatoms, possibly in the form of deepwater blooms (cf. Leavitt et al. 1989).

Concentrations of algal and bacterial pigments increased to peak values in the most recently deposited sediments (upper 3 cm; Fig. 1). In particular, pigments from colonial cyanobacteria (myxoxanthophyll, canthaxanthin) and green algae (lutein-zeaxanthin, Chl b, pheophytin b) increased two-fold in the surface 3 cm, whereas smaller increases were recorded for other indicator compounds. Because these pigments are among the most reliable of the algal indicators (Leavitt 1993, Leavitt and Findlay 1994), these patterns suggest a recent increase in the abundance of bloom-forming taxa. However, at this juncture it is uncertain whether inferred increases algal abundance have occurred in surface waters (where they are seen by the public) or at depth in the form of deepwater populations. Further, despite the relative stability of the indicator compounds, nonselective post-depositional degradation of pigments is common in many lakes and may have also contributed to the observed patterns in the most recently-deposited sediments (Leavitt 1993).

Analysis of fossil okenone from purple sulphur bacteria suggested that the stratification regime of Windermere Lake has varied in the recent past. For example, concentrations of okenone increased five- to 10-fold in the uppermost 10 cm of sediment. Because these phototrophic bacteria require light for photosynthesis, yet are fatally poisoned by molecular oxygen, the presence of their pigments throughout the core suggests that light

has always penetrated into anoxic bottom waters. Most commonly, this condition occurs when a lake is strongly or permanently stratified (meromictic), relatively transparent and when bottom waters are completely anoxic. However, sharp increases in the deposition of okeneone in the most recent sediments suggest either that light penetration has recently improved, or that the lake has become more strongly stratified and that deepwater anoxia has become more profound. Presently, we cannot distinguish among these mechanisms. Regardless of the cause, we note that concentrations of okeneone were always low when compared with values recorded from strongly stratified sites (e.g., Leavitt et al. 1989), suggesting that Windermere Lake exhibited only seasonal stratification.

(sediment artifacts
anoxia during ice
cover)

In summary, analysis of sedimentary pigments from Windermere Lake suggested that original fossil assemblages were mainly composed of eukaryotic algae, particularly diatoms, cryptophytes, and green algae. Bloom-forming cyanobacteria were present, but relatively uncommon. Total algal production is inferred to have increased slightly from minimum values recorded deep in the core to maxima in the uppermost 3 cm of sediment. Concentrations of many indicator pigments, particularly those of colonial cyanobacteria, also increased ~two-fold to peaks in the uppermost sediments, although it is unclear whether this pattern arises from increased production of deepwater blooms, surface blooms or nonselective degradation of pigments following algal deposition to the lake bottom. Analysis of fossil carotenoids from anaerobic sulphur bacteria suggested that the stratification regime of Windermere Lake has changed recently, either due to improved light penetration into anoxic waters, or increased chemical stratification leading to more profound oxygen deficits in bottom waters.

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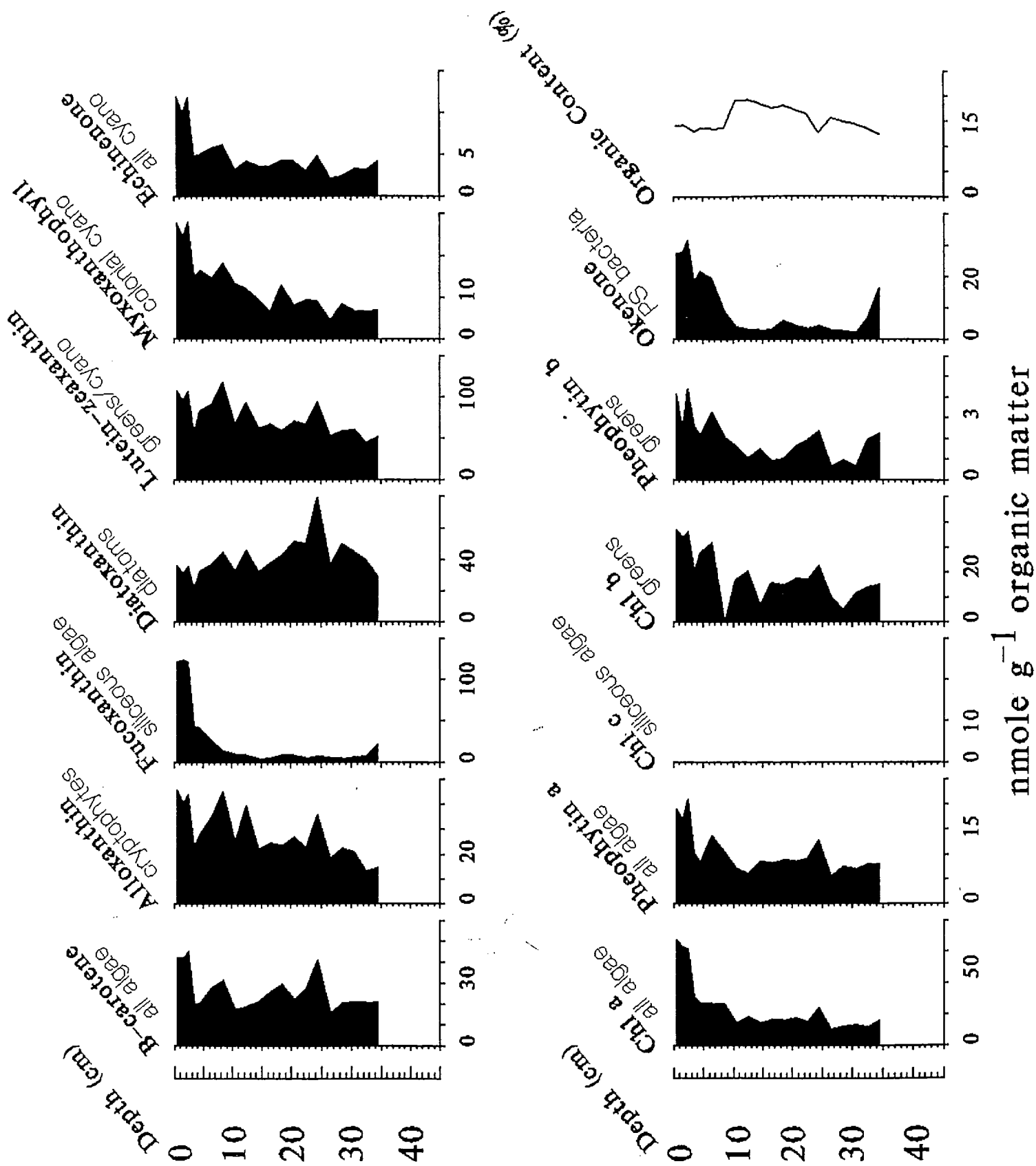
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APPENDICES

Windermere Lake



Windermere Lake pigment concentrations.

All pigment concentrations in units of nmoles pigment/gram organic matter. Missing values (.) are distinguished from sample without pigment (0.000). Column headings represent:

Lake - character variable identifying site

Depth - burial depth in cm

Chlorophyll c - pigment from diatoms, chrysophytes and dinoflagellates (10x actual concentration)

Chlorophyll c isomer - pigment from diatoms, chrysophytes and dinoflagellates (10 x actual concentration)

Fucoxanthin - carotenoid derived from diatoms, chrysophytes and some dinoflagellates.

"Aphanizophyll"- groups of carotenoids produced by nitrogen-fixing cyanobacteria. Includes aphanizophyll, 4-keto-myxoxanthophyll, and oscillaxanthin

Diadinoxanthin - carotenoid from diatoms and some dinoflagellates.

Myxoxanthophyll - carotenoid from colonial cyanobacteria

Alloxanthin - carotenoid from cryptophytes

Diatoxanthin - carotenoids from diatoms and some chrysophytes

Lutein-zeaxanthin - carotenoids from green algae (Lutein) and cyanobacteria (zeaxanthin)

Canthaxanthin - carotenoid from colonial cyanobacteria, predominantly bloom-forming Nostocales.

Chlorophyll b - pigment from green algae

Carotenoid X - unknown origin, possibly phototrophic bacterial pigment

Okenone - carotenoid from phototrophic purple sulphur bacteria.

Chlorophyll a - pigment from all algae

Echinonone - carotenoid from all cyanobacteria

Isorenieratene - carotenoid from green sulphur bacteria (tentative)

Phaeophytin b - derivative of Chl b

Phaeophytin a - derivative of Chl a

β -carotene - carotenoid from all algae

Organic content - % weight loss on ignition at 550°C (1 hr)

Total Chl C - sum of Chl c isomers

"WIND	"	0.500,	0.000,	0.000,	121.118,	8.258,	0.000,	27.525,	45.672,	36.217,	107.292,	11.384,	12.071,	36.784,
27.575,	84.609,	11.876,	0.000,	0.000,	4.112,	18.911,	42.208,	14.167,	0.000					
"WIND	"	1.500,	0.000,	0.000,	123.476,	7.391,	0.000,	24.520,	40.740,	31.302,	95.660,	10.132,	5.338,	33.921,
27.652,	78.701,	10.044,	0.000,	0.000,	2.456,	17.022,	41.744,	14.404,	0.000					
"WIND	"	2.500,	0.000,	0.000,	120.199,	7.787,	0.000,	27.734,	44.130,	35.184,	106.291,	11.366,	12.149,	35.931,
31.356,	76.428,	11.813,	0.000,	0.000,	4.367,	20.845,	44.855,	13.852,	0.000					
"WIND	"	3.500,	0.000,	0.000,	43.737,	4.156,	0.000,	14.909,	23.044,	21.685,	56.872,	6.509,	7.873,	20.190,
18.496,	38.664,	4.809,	0.000,	0.000,	2.513,	10.159,	19.631,	12.886,	0.000					
"WIND	"	4.500,	28.236,	0.000,	40.864,	6.693,	11.871,	16.359,	28.550,	32.582,	82.305,	8.832,	5.198,	27.710,

21.613,	33.838,	4.973,	0.000,	2.154,	8.600,	20.948,	13.682,	28.236											
"WIND 19.264,	" 6.500, 33.062,	0.000, 0.000, 5.746,	0.000, 0.000, 0.000,	26.409, 3.221, 13.591,	14.097, 28.134, 13.385,	0.000, 0.000, 0.000	14.362, 13.385,	34.975, 0.000	37.298,	91.202,	8.234,	7.049,	31.830,						
"WIND 8.911,	" 8.500, 32.315,	0.000, 0.000, 6.138,	0.000, 0.000, 0.000,	13.749, 23.022, 10.543,	18.356, 31.400, 13.860,	44.921, 0.000	17.961, 13.860,	44.540,	116.161,	7.966,	6.283,	0.000,							
"WIND 3.852,	" 10.500, 17.739,	0.000, 0.000, 3.126,	0.000, 0.000, 0.000,	9.882, 12.072, 7.033,	8.640, 16.919, 19.113,	25.286, 0.000	13.180, 19.113,	32.475,	66.732,	5.114,	6.718,	17.301,							
"WIND 2.974,	" 12.500, 22.806,	0.000, 0.000, 4.091,	0.000, 0.000, 0.000,	8.449, 19.158, 6.114,	26.577, 18.813, 19.257,	39.224, 0.000	11.884, 19.257,	45.684,	91.586,	5.524,	8.690,	20.504,							
"WIND 2.761,	" 14.500, 17.765,	0.000, 0.000, 3.528,	0.000, 0.000, 0.000,	4.125, 8.287, 8.435,	0.000, 21.261, 18.375,	22.138, 0.000	9.305, 18.375,	32.154,	60.889,	3.612,	5.720,	7.106,							
"WIND 3.122,	" 16.500, 20.493,	0.000, 0.000, 3.526,	0.000, 0.000, 0.000,	6.007, 9.107, 0.898,	0.000, 25.990, 17.478,	24.329, 0.000	6.550, 17.478,	37.424,	66.500,	3.912,	5.680,	15.595,							
"WIND 6.021,	" 18.500, 20.019,	0.000, 0.000, 4.253,	0.000, 0.000, 0.000,	9.375, 9.134, 1.009,	0.000, 29.622, 18.007,	23.483, 0.000	12.785, 18.007,	42.851,	57.731,	4.755,	2.920,	14.537,							
"WIND 4.128,	" 20.500, 21.553,	0.000, 0.000, 4.097,	0.000, 0.000, 0.000,	9.097, 8.311, 1.593,	0.000, 21.999, 16.944,	26.871, 0.000	8.028, 16.944,	51.069,	70.084,	5.094,	4.266,	17.451,							
"WIND 3.320,	" 22.500, 17.974,	0.000, 0.000, 2.946,	0.000, 0.000, 0.000,	5.900, 6.522, 1.877,	0.000, 27.393, 16.123,	22.590, 0.000	9.329, 16.123,	49.647,	65.299,	4.611,	5.991,	16.986,							
"WIND 4.096,	" 24.500, 29.820,	0.000, 0.000, 4.740,	0.000, 0.000, 0.000,	8.018, 10.240, 2.336,	0.000, 40.818, 12.616,	35.972, 0.000	8.961, 12.616,	79.043,	93.110,	5.848,	5.772,	22.649,							
"WIND 2.680,	" 26.500, 12.766,	0.000, 0.000, 2.137,	0.000, 0.000, 0.000,	5.082, 5.271, 0.606,	8.682, 15.400, 15.459,	18.543, 0.000	4.430, 15.459,	36.628,	52.120,	2.403,	1.029,	9.930,							
"WIND 2.428,	" 28.500, 15.090,	0.000, 0.000, 2.293,	0.000, 0.000, 0.000,	4.267, 6.144, 0.923,	5.068, 20.355, 14.745,	22.247, 0.000	8.282, 14.745,	49.776,	57.724,	3.097,	3.335,	5.009,							
"WIND 1.830,	" 30.500, 16.076,	0.000, 0.000, 3.146,	0.000, 0.000, 0.000,	6.131, 5.935, 0.634,	5.810, 20.907, 14.233,	20.689, 0.000	6.555, 14.233,	44.288,	59.182,	4.669,	2.811,	11.459,							
"WIND	" 32.500,	0.000,	0.000,	7.525,	2.039,	0.000,	6.477,	39.398,	44.098,	4.739,	2.126,	13.869,							

[illegible]

