# TIMELINE

OF

**EVENTS** 

Time	A at: it.					
Time	Activity					
Unknown	May 15, 2006					
(assumed to	Doug Erickson (DE) arrived on site to conduct routine					
be AM)	sampling and monitoring of water flows					
1224	A coll was made by DE Live III					
1224	A call was made by DE on his cell phone					
1330	A call was received by DE on his cell phone					
1330	DE meets BC Timber Sales, at Euclid Fuel					
	Area approximately 5 minutes west of accident location.					
Dohyoon	They spoke for approximately 15 minutes.					
Between 1345 and	DE's vehicle was observed by Bruce J. Donald (BJD) and					
	Bob Newcombe (BN) in the vicinity of the #1 Shaft Waste					
1400	Dump					
1400	BJD photograph of #1 Shaft Waste Dump shows DE's truck					
	at the accident site. Photo numbers 1006389 and 1006390					
	have time stamps of 11:04:17 and 11:04:21.					
	Review of the camera Kodak model DC5000 on May 23,					
	2006 reviews a difference of 2 hours and 55 minutes from					
	the current day actual time determined using Fred					
Dotuces	Hermann's watch.					
Between	DE arrived and entered the #1 Shaft Waste Dump Flow					
1345 and	Intake building (accident location)					
1400						
	No other record of outgoing telephone calls or visual contact					
	with DE until his discovery on Wednesday, May 17, 2006					
0812	May 16, 2006					
0012	DE's cell phone notes a missed call from					
1150	:ell phone					
1130	DE's cell phone notes a missed call from Pryzm					
	Environmental					
OCEO	May 17, 2006					
0659	called DE's cell phone					
0700 to 0746	A total of 3 additional missed calls from unknown persons					
~0800	called BN requesting a search for DE					
-09001	DNI					
~0800+	BN and discussed where DE should					
	be working and agreed to begin a search. BJD reminded					
	BN and hat BJD and BN had seen DE's truck at the #1					
-090011	Shaft Waste Dump area on Monday.					
~0800++	BN and with BJD's cell phone) leave the office in					
	separate vehicles to begin search. BN was going to upper					
	Mine area and to Arena CT-04 and work toward CT-05					
1	1 (4)					
0040 ::	and Sludge Pond.					
0843 time from BN's	and Sludge Pond.  BN called (cell to cell) said he had found DE's truck.  BN had looked in the building seeing DE					

cell phone	Bob was going to call ambulance, should go to				
	mine gate.				
0845 from	BN called 911 (confirmed by BC Ambulance Call Log)				
BN's cell	reported a possible sampling shack.				
phone	Full text of call in 911 way file #2006.05.17074537				
0848 from	BN called Teck Cominco's				
BN's cell	to respond to the #1 Shaft Waste Dump				
phone	area v-				
After 0848	BN entered the building				
before 1st	_				
ambulance					
arrival					
Between	and first ambulance arrived at accident site.				
0848 and	Kim Weitzel (KW) entered the building, started down the				
0858	ladder, paused to ask if gas was present.				
	She				
	told Snawn Currier (SC) something had happened to				
	his partner and SC entered the building descending				
	immediately, asked KW if she was OK and				
0850	Alan Collinson (AC), Acting Fire Chief receives a call for				
	assist from 911 dispatch of at the Sullivan Mine				
0859 from	places a 911 call				
BJD's cell	Advises 911 that we have 4 people down including 2				
phone	ambulance attendants.				
	Mentions possible gas and H2S.				
	Advised by dispatch not to let anyone into the shack.				
	Refer to 911 wav file 2006.05.17075907				
0902	meets on road between accident site and mine				
	gate advising him not to enter the building may have gas.				
	Refer to 911 wav file 2006.05.17075907				
0903	meets AC on road between accident site and mine				
	gate advising him not to enter the building without apparatus				
	- may be gas				
	Refer to 911 wav file 2006.05.17075907				
0905	AC arrives at accident site.				
	One BC Ambulance, One TC ambulance, TC pickup, TC				
	and private vehicle at scene.				
0915	Engine carrives at accident sight and prepared to enter				
	building with confined space gear.				
0917	Second ambulance – arrives on site establishes triage				
	area.				
0922	First patient, KW, removed from the building and taken to				
	triage.				
0923	Engine arrives on scene.				
0929	Second patient, Shawn Currier (SC) removed from building				
	and taken to triage.				

0933	Third patient, BN, removed from building and taken to triage			
0934	Third ambulance, arrives on scene.			
0936	Ambulance arrives on scene			
	and goes to triage area.			
0937	First victim, DE, removed from building			
	by			
0944	First ambulance, leaves for hospital with KW.			
0945	AC notifies Kimberley RCMP to attend scene.			
0947	Fourth ambulance, arrives on scene.			
0948	Second ambulance, leaves for hospital with SC.			
0950	Third ambulance ????, leaves for hospital with BN.			
0955	Fifth ambulance, ?????. arrives on scene.			
0955-1007	Fourth and Fifth Ambulance leave scene.			
1007	RCMP arrives on scene.			
1020	Engines and return to station.			
1020	Chief Inspector of Mines, Fred Hermann (FH), notified of			
	accident.			
1020	Regional Director, Ricci Berdusco, (RB), and manager			
	mining Richard Booth notified of accident by FH.			
1040	Richard Booth calls Terry Paterson (TP) and TP is directed			
400-	to proceed to scene secure the site and control access.			
~1025	arrives on scene.			
1031	BJD calls MEMPR			
~1035	Helicopter landed just south of the accident scene.			
~1042	Coroner Gerry McIntyre (GM) arrives on scene.			
~1050	FH advised there are four fatalities.			
1109	Teck Cominco calls in Emergency Response Team (ERT)			
1120	from Trail Smelter.			
~1210	TP on scene and secure site.			
~1210	Safety perimeter moved to command center site 200 yards			
1247	up from main gate access.  Phil Pascuzzi (PP), and Garry MacDonald (GMD), arrives on			
1 2m T 1	scene.			
1425	AC and confined space team return to scene, conduct air			
20	tests, establish safety perimeter and to ensure public was			
	not at risk.			
	Remained on scene until 1623 conducting perimeter gas			
	checks			
1500	FH and RB arrived on scene FH assumes control of site.			
1500	Teck Cominco senior staff arrives at office location.			
1620	arrive with gas testing			
	equipment.			
1623	AC and confined space team returned to station.			
~1700	DE removed from site.			
~1800	ERT arrives on site. Conducts perimeter gas checks.			
	guo official.			

2035	ERT stand down for the day.
~2130	Building locked, site secured, security on site.

# Air tests taken by Kimberley Fire Department Using ISC Model TMX-412

Date	Time	Location	H2S	СО	Оху	LEL
17/05/06	1425	Safe Zone #2	.3 ppm	BDL	21.1%	0
	1427 100 M away		1 ppm	5 ppm	21.1%	N/R
		from building				
	1429	30M	1 ppm	5 ppm	21.1%	N/R
	1431	25M downwind	0 ppm	5 ppm	21.1%	N/R
		from the				
		building				
	1433	8M perimeter of	0 ppm	5 ppm	21.1%	N/R
		building				
	1434	4M perimeter	0 ppm	5 ppm	21.1%	N/R
	1436	NW side 3M	0 ppm	5 ppm	18.0%	N/R
	1436	Downwind side	0 ppm	5 ppm	18.0%	N/R
		(SW)				
	1437	Tester inside	0 ppm	0 ppm	7.0%	0
		the door (in				
		breathing zone)			<u></u>	
	1440	Returned to Safe	Zone #2	· · · · · · · · · · · · · · · · · · ·		
	1452	Established			:	
		safe zone #3	}			
'		10M				
	1500	Perimeter test	4 ppm	0	21.1%	0
1510 Con	tinuous p	erimeter monitorin	g no change	es in readin		
	1528	Reading from	0	0	0.7%	5%
·		inside of the				
		building 4 feet				
		down from the				
		platform				
1633 retui	rned to th	e Fire Station Base	9			

<sup>\*\*</sup>N/R denotes Not Recorded

D--- F-644

# Air tests obtained by and Using BW Tester

Date	Time	Location	H2S	CO	Оху	LEL
17/05/06	1625	Loc # 1 10M perimeter	0	0	20.9%	0
	1640	Loc # 2 At the door ledge	0	0	3.4% Using pump with probe	4%
	1650	Loc # 2 At the door ledge	0	0	9.5% Using pump with probe	0%
	1725	Loc #1	0	0	20.9%	0%
	1726	Loc #3	6 ppm	0	20.9%	0%
	1730	Loc #2 At the door ledge	0	0	7.25 Using pump with probe	5%
1726 read	ding witn ero	essed by Ricci Ber	dusco it fla	shed on s	creen and ret	urned
	1811	LOC # 1	0	0	20.9%	0
	1820	Loc # 4 by the front bumper of pick up truck very close to the shed	0	0	19.9%	0
	1935	Vent Pipe approx. 30M south of Shed	3 ppm	0	20.9%	0
	2010	3900 portal manhole	0	0	20.9%	0
	2020	Pump Station 961/962	0	0	20.9%	0
	2032	Gerry Sorensen Intake	0	0	20.9%	0
May 18/06	0810	Loc # 5 @ incident site approximately 80feet away from the shed 2 <sup>nd</sup> safety zone established	0	0	20.9%	0
0820 BW 1	ester ca	librated				
	0915	Loc #5	0	0	20.9%	0
	0927	Loc #5	0	0	20.9%	0
	0928	Loc#2	0	0	15.7%	0
	0945	Loc #1	0	0	20.9%	0
	0955	Loc#1	0	0	20.4%	0
)950 SO2	- 10 pur	nps less than 0.5p	om at 20 p	umps 0.50	mag	

Down 6 -644

0956	002 > 300	Oppm above the we	eir in the s	shed.		
0956 L	.EL varied	between 6% and 4	% Oxy 0.0	6% H2S 0	ppm	
	1047	3900 portal	0	0	20.9%	0
	1206	Pump Station 961/962 50 feet east of shed	0	0	20.9%	0
	1213	In side of the shed	0	0	20.9%	0
	1300	Gerry Sorensen way intake 15' in front of the door	0	0	20.9%	0
	1301	Gerry Sorensen way intake @ the door	0	0	20.9%	0
	1302	Gerry Sorensen way intake inside first door	0	0	20.9%	0
	1303	Gerry Sorensen way intake 12' inside first and second door	0	0	20.9%	0
	1512	Incident site Loc #5	0	0	20.9%	0
	1513	Loc#4	0	0	20.9%	0
May 19/06	0110	Perimeter readings outside of shed within 5'	0	0	20.9%	0
	0150	Perimeter readings outside of shed within 5'	0	0	20.9%	0

Tests	s taken b	elow by ERT using only 2 senso	j handheld	l analog sa n and I FI	ampling do	evice with
May 20/06	1040	3900 Portal Outside			21.0%	0
	1045	3900 Portal			21.0%	0
	1046	3900 Portal inside manhole			21.0%	0
	1103	Pump Station 961/962 outside			21.0%	0
	1105	Pump Station 961/962 Inside			21.0%	0
	1108	Pump Station 961/962 Inside Lower floor			21.0%	0
	1112	Pump Station 961/962 sump inlet			21.0%	0
	1115	Pump Station 961/962 Sump north of shed		·	18.5% Probe at water level	0
	1130	Grit Chamber Pump Station			21.0%	0
	1135	Grit Chamber emergency shut off valve out side of door			21.0%	0
	1136	Grit Chamber emergency shut off valve In side of Room		,	21.0%	0
	1140	Gerry Sorensen Way Intake Outside			21.0%	0
	1141	Gerry Sorensen Way Intake Inside of the first door			21.0%	0
	1142	Gerry Sorensen Way Intake Inside 2 <sup>nd</sup> room			21.0%	0

ALL TIMES ARE MOUNTAIN STANDARD TIME (MST)

Sample	May 17 Wed	STANDARD TIN May 18 Thu	May 19 Fri	May 20 Sat
Location	1			_
Pipe In-flow	1730			
(TC)	1730			
1 Ft. above	1730			
1	1730			
Weir (TC)	4700			
1 ft. below	1730			
grade				
plywood floor				
(TC)				
Opening of	1730			
Outlet Pipe				
(TC)				
Upstream		~0930		
Pipe (TC)				
Bottom of Pit		~0930		
(TC)				
3900 Portal		1040		
(TC)				•
961/962	-	1206		
Pump Station			+	
(TC)				
Gerry		1300		
Sorenson		1000		
(TC)				
			~0137	
Sample 1A			10107	
(MEMPR)			  -	
At Pipe			~0137	
Sample 1B			~013/	
(MEMPR)			]	
At Pipe			0407	
Sample 1C			~0137	
(TC)				
at Pipe	<u> </u>			
Sample 2A			~0137	
(MEMPR)				
1 Ft. below				
entry level	,			
Sample 2B			~0137	
(MEMPR)				
1 Ft. below				
entry level				
Sample 2C			~0137	
(TC)				
1 Ft. below				•
I I FL. DEIOW	1	1	1	1

ontry loyal				
entry level Sample 3A			~0137	
(MEMPR)			~013/	
Inside Shed				
(Breathing				
Zone) 5 ft		\ 		
above entry		,		
level				
Sample 3B			~0137	
(MEMPR)			013/	
Inside Shed				
(Breathing				
Zone) 5 ft				
above entry		,		
level				
Sample 3C			~0137	
(TC)			010.	
Inside Shed				
(Breathing				
Zone) 5 ft				
abové entry	<u> </u>	·		
level				
Sample 4			0215	
(MEMPR)				
In Fresh Air				
approx. 300		[		
yds. from				
incident shed				
uphill and				
upwind			<del></del>	
Sample 1				1046
(Rescan for				
MEMPR) at		]		
Portal 3900				
inside bldg. at				
ground elev.				A dette 1
				Additional
		]		Samples taken
				by Danner recylle
				Rescanresults
				expected May
				23, 2006

All samples were gathered by the ERT (Emergency Response Team) lead by from Trail Operations. Air samples are bag (grab) samples of the atmosphere. Both MEMPR and Teck Cominco had representatives witnessing the sampling.



Ministry of Health

British Columbia Ambulance Service Kamloops Communication Centre

#### **MEMORANDUM**

Prepared by: Doug Garland

Date: May 19, 2006

File:

Re: Sullivan Mine - Kimberley 2006.05.17 - updated 1645 19 MAY 2006

Below is a chronology of events for the above incident – times (PST) taken from Freedom Recording Application unless otherwise noted

Time	Event	Details
0745	Call Received in Kamloops Centre	<ul> <li>Tadanack Boulevard, the upper mine, at the toll(?) of the #1 shaft waste dump, Sullivan Mine, Kimberley</li> <li>Only one main entrance, will have a man there to direct the ambulance</li> <li>Man may have ne is in a sampling shack</li> <li>Patient is will be able to access,</li> <li>Not an operating mine – no mine rescue</li> <li>Will be able to access the patient – it's a confined area that is the only problem</li> <li>ear old male</li> </ul>
0746		<ul> <li>dispatched to the call — location reviewed with male crew member on the phone</li> </ul>
0747		<ul> <li>assigned the call within the CAD software (CAD time)</li> </ul>
0748		* 10-8 to call (data head)
0748	Kimberley Fire Dept	<ul> <li>contacted – possible call at the Sullivan Mine, upper mine, use the main entrance, year old male has – Tadanack Blvd</li> <li>Want you to be advised should you choose to respond</li> <li>Fire asks if we have contacted Search and Rescue, no</li> <li>EMD indicates to Fire the Kimberley Fire does not do First Response but the (Fire Dept) do rescue</li> </ul>
0756	Kimberley Fire Dept	calls back for more detailed location information for call – they are up on the road and don't see anybody
0757	Kimberley Fire Dept	<ul> <li>A/Charge talks to FD – instructed to go to the main entrance – upper mine at the Sullivan Mine</li> <li>A/ Charge gives information to Fire Department of upper Sullivan Mine main entrance</li> </ul>
0758	Kimberley Fire Dept	<ul> <li>A/ Charge give additional information - #1 waste shaft –</li> <li>FD informs Cominko truck has arrived – have not seen the ambulance but they took off before we did</li> <li>EMD attempts to contact – by radio</li> </ul>
0759	Mine Calls	<ul> <li>ambulance crews are in the hole and have in the hole they are both</li> <li>must be gas in there as they have .n the hole</li> </ul>

		we need more people here as there is no one here  Mine staff says that they have no idea what type of gas might be H2S  Both attendants are  There is no mine rescue – I am here by myself  Fire Dept is on the way  mine staff instructed by dispatch to not let them go in unless they have the proper apparatus  can see them in the hole  one lady (1 <sup>st</sup> attendant) asked if there was gas
		now on scene 4 people down in the hole 2 <sup>nd</sup> ambulance attendant went into the hole and asked "are you OK are you OK" and then
		<ul> <li>fire department on scene - they</li> <li>EMD gets exact directions to the mine gate - go past the Hospital, turn left at the Town Site Grocery, carry straight on meet a the main gate</li> <li>FD Chief is there, radioed down to the Fire Hall as they did not have any of their breathing equipment with them</li> <li>Car coming from Cranbrook and another car coming from Kimberley</li> </ul>
0800	Dave Brookes (Supt)	<ul><li>paged</li><li>paged</li><li>paged</li></ul>
0801		both crew members call in
0802	Kimbaylay Eira	calls in – able to clear in about 5 minutes
0002	Kimberley Fire  Dave Brooks (Supt)	<ul> <li>Called by A/ Charge – advised possibly H2S – 2 ambulance attendants down – please advise your crews</li> <li>Calls in – updated on the incident year old male went , the bystander has 2 crew members went in, sounds like H2S, may be up to 4 patients, both crew members have</li> </ul>
		<ul> <li>Paged</li> </ul>
		<ul> <li>Paged</li> <li>assigned the call within the CAD software (CAD time)</li> </ul>
0803		both crew members call in
0804	<u>, , , , , , , , , , , , , , , , , , , </u>	assigned the call to Sullivan Mine
0805	Cranbrook Hosp	<ul> <li>Acting Charge notified Cranbrook hospital of 4 patients</li> <li>Up to 4 patients in probably dealing with H2S – just giving notification, not started transporting yet</li> <li>Dispatch ticket marked as on route (data head)</li> </ul>
0806		<ul> <li>Marked on the dispatch ticket as being 10-7 (this is indicated on the dispatch ticket that it was manually time stamped by the EMD)</li> </ul>

		<ul> <li>10-8 – no info on the data head – EMD gives the call over the air to the crew – advises crew to stay well clear of danger</li> </ul>
0807		<ul> <li>calls dispatch to clarify directions</li> </ul>
0808	Kimberley Fire	<ul> <li>declares a confined space entry – ambulance attendants have entered the space – have activated their confined space crew</li> </ul>
		no response from ambulance crew
		<ul> <li>no response from anyone in the hole</li> </ul>
	470	<ul> <li>breathing apparatus ETA 6 minutes</li> </ul>
0809	PAACC	<ul> <li>PAACC calls Kamloops offering assistance with staging resources in the Kimberley area</li> </ul>
0011		= 10-7 scene (datahead)
0811		<ul> <li>On scene – 1<sup>st</sup> two ambulance attendants are down</li> <li>is on scene – you stay well clear - were (Dispatch)</li> <li>not in touch with – hey might be</li> </ul>
		<ul> <li>is also been responding, keep me updated</li> </ul>
		<ul> <li>The hospital has been notified</li> </ul>
		<ul> <li>EMD advises crew to make sure the do not get</li> </ul>
0040		themselves in danger
0816		Calls in update – may be 4 down there – our two crew
		and two more – the fire department just arrived with
		apparatus – going to patch you to the charge so that we continuity
		<ul> <li>EMD patches to Charge</li> </ul>
		Sent ½ way to Cranbrook
0817	PAACC	<ul> <li>Kelowna repositioned to pick up DBrooks</li> </ul>
		4 down there 2 of our crew and 2 others confined space
		team on site confirmed H2S exposure with Fire
		Dept – fire department does not know concentrations –
		cautioned by A/ Charge to be wary of their clothes
		as sometime they have the H2S on their clothes –
		member says that but is thankful for the reminder
		<ul> <li>Calls in for update – still on route – possibly 4 patients down the shaft – couple of those may be our crew</li> </ul>
		members – advised by EMD to stay well clear – is
	•	on scene, and have been paged in
0818		Sent into Kimberley for coverage
0819		<ul> <li>Paged to call disp – fly to Kimberley</li> </ul>
		<ul> <li>10-8 sent right into Kimberley</li> </ul>
0820		way to Cranbrook for coverage
0822	Charge	Mine site calls back with update – reporting fire
	Ü	and BCAS on scene - Fire has brought their engine with
		their breathing apparatus – also another group has
		arrived the confined space team
0824	Cranbrook Hosp	Updates hospital that there are 4 confirmed patients with
	•	H2S - now entering the mine with breathing apparatus
		and mine rescue to try get them out
		<ul> <li>1<sup>st</sup> person been down for about 1 hour</li> </ul>
		a ather for about 1/ hour

		<ul> <li>we have 2 ambulance crew that have gone down – have been down for about 20 minutes</li> </ul>
0825	Disp	<ul> <li>updates DBrooks – mine rescue they have amassed some sort of mine rescue team an they have fire there going in with the proper apparatus</li> </ul>
0826	Kimberley	<ul> <li>Discusses location directions with Charge</li> <li>Called to respond to Mine site for MCI</li> </ul>
0828	Invermere	Called to respond to Kimberley
0830		<ul> <li>Sent right in to Mine site</li> <li>10-7 @ Mine site</li> </ul>
0832	-	<ul> <li>states he needs to know where these guys are at</li> <li>updated two have has been pulled from the tunnel —         ot sure who it is — is transporting – the</li> <li>other is on the way up – EMD asked him to call DBrooks</li> </ul>
0833	Kimberley Fire	<ul> <li>sent to Kimberley</li> <li>called to update – 2 people out confirmed two other people in the hole as well - 2 other people still in there - 3<sup>rd</sup> individual removed</li> <li>updated that – is also responding now</li> <li>Sent right into Cranbrook</li> </ul>
0834	Cranbrook Hosp	Charge updated 2 pts on route both
0835		<ul> <li>Calls in update of 3 confirmed</li> <li>10-8</li> </ul>
0837		<ul> <li>member located by disp to move spare unit to airport to transport ALS airevac crew</li> <li>Sent right into Kimberley</li> </ul>
0839	:1	Charge informs \ regarding Kimberley incident
0840	PAACC	<ul> <li>PAACC calls _ will be in Kamloops in 10 minutes</li> <li>P.Hecher can go to AirSpan Helicopters AVF – going to</li> </ul>
0844		PHecher notified to respond to AirSpan
0845		
0846		Ticket marked by EMD as transporting (CAD time)
0849		<ul> <li>Ticket marked by EMD as 10-7 scene (CAD time)</li> <li>Transporting to Cranbrook</li> <li>Ticket marked by EMD as transporting (CAD time)</li> </ul>
	DGarland	<ul> <li>Asking for unit to be placed at the Cranbrook airport for airevac crew</li> <li>Suggested calling direct for CISD</li> </ul>
0850		Transporting (data head)
)851		<ul> <li>Transporting (data ricad)</li> <li>Transporting 1 patient – till on scene – confirms with EMD that there are no other patients needing to be transported</li> </ul>
0852		Trauma support calls back
0854		<ul> <li>10-7 at mine</li> </ul>
ሰደናል		- calla EAAD calcina for times informed at in

<u></u>		crew member – asking for help at hospital
0857		calls in asking where they are to be positioned
	A/Charge	notified RCMP
0858		ு sent ⊂ ംo Cranbrook Hospital
0900	A/Charge	a calls "place on hold
		<ul> <li>lifts off in Invermere heading for Sullivan Mine Site</li> </ul>
0901		<ul> <li>calls for update – will attend at the hospital</li> </ul>
0902		<ul> <li>calls back – Charge asks that they have CIS at Cranbrook Hosp ASAP</li> </ul>
0905	Cranbrook Hosp	<ul> <li>called by Charge updated situation and ETA</li> </ul>
		<ul> <li>retrieved both ambulances returning to Cranbrook</li> </ul>
		<ul> <li>ETA to Cranbrook Hosp 5 minutes – other 2 units are</li> </ul>
0910	DBrooks	about 10 minutes from Hosp
0911		boards in Kamloops for Cranbrook
	Kimberley Fire	<ul> <li>Charge calls to provide CISD contact information</li> <li>10-7 Cranbrook Hospital</li> </ul>
0912		out 5 minutes from the hosp ( nas split their
		crew in two different cars both using the same call sign)  10-7 hospital (data head)
	PAACC	<ul> <li>Calls with updated ETA for into Cranbook in 45</li> </ul>
		minutes, close to 10
0913		<ul> <li>Update – both second patients ETA 5 min to hospital</li> </ul>
0914	LHarder	<ul> <li>Calls in for update from Charge</li> </ul>
0915		= 10-7 – Station
		<ul> <li>2<sup>nd</sup> half of 10-7 Cranbrook Hosp</li> </ul>
0917		<ul> <li>10-7 Cranbrook Hosp</li> </ul>
0000		Ticket marked by EMD as 10-7 (CAD time)
0926		<ul> <li>Discusses situation with Charge – Charge updates and</li> </ul>
0928		offers whatever help they need
0920		<ul> <li>Calls in from Mine Site – one on scene will carry onto Cranbrook Hosp</li> </ul>
0933	Kimberley Fire	<ul> <li>Charge calls to inform that a number of firefighters are in</li> </ul>
0000	ramboney i ne	Cranbrook and arrangements need to be made to get
		them back to Kimberley
0935		Approximate time at Cranbrook
0937		= 10-7 station
0937		cleared for return to Cranbrook
0945	Victoria Disp	Called asking for names of deceased crew
0948	CUPE 873	CUPE 873 office notified of deceased crew members
		Calls in with update and confirmation of 1 pronounced
	***	crew member
0953	RCMP	<ul> <li>Asking for delay of notification of NOK until BCAS</li> </ul>
	· · · · · · · · · · · · · · · · · · ·	management can arrive
0958		□ Clear station
0959	RCMP	<ul> <li>Called back to inform there will be a staging area at the</li> </ul>
		Hosp – will try to delay the notification of NOK until
		BCAS management can be on site
1000	DBrooks	<ul> <li>Approximate time touch down in Cranbrook</li> </ul>

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1006	<ul> <li>Charge calls equesting ETA for Cranbrook – Counselor on route ETA 1 hour</li> </ul>
1012	<ul> <li>Calls in update that both BCAS crew members are deceased</li> </ul>
1013	<ul> <li>Confirmed that they have left the airport with DBrooks should be at hosp in 10 minutes</li> </ul>
1029	<ul> <li>10-7 stn</li> <li>EMD places crew on city coverage</li> </ul>
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## **ADDITIONAL**

# WITNESS STATEMENTS

Fire Fighter statement for .

On Wednesday morning May 17,2006 I was directed by acting Fire Chief Al Collinson to respond to at the end of Tadanac Blvd. I along with firefighters nd. unded with Eng Al Collinson took off a head in the command unit. When we reached the address we stopped to look over the bank to see if there was an ambulance down on the lower mine road as the direction the fire chief received from the dispatch were quite confusing. We were looking to see if there was some one in the creek. The chief had continued up the mine hill towards the dam to see if that were the incident was located. He came into contact with some one from tech cominco and said the incident was at the #1 mine dump. He then returned to the lower mine parking lot and had both nd go with him in the command unit while I returned to the station to activate the search and rescue crew for the rescue of As I reached the station I received a call on the radio for me to contact the city confined space crew and return to the mine #1 dumpsite with a crew and Eng I asked firefighter J do a call out for the fire department and to . to get the confined space crew, which he did. I along with contact the -Fire fighter hen responded to the mine site.

Enroute we were informed by the Chief that we would require SCBA when we arrived at the scene .ıd <sup>–</sup> donned their protective equipment and rescue harnesses. Both and I completed the safety checks. Another ambulance arrived and was setting up a triage area. Once and prepared to enter the building with ontrolling the rescue lines. pulled the first paramedic (female) up and I took her over to the triage area where ambulance attendants were located. The second victim .nd (male) attendant was pulled up and we took him to the triage area. Our second team arrived just as we had removed this person and as low on air so long with pulled out the third victim bnr and we took him to the triage area. The fourth victim was removed and and we then moved him to an area next to triage. Once we had all the victims removed we had the crews move to a rehab area behind the Command unit. and I moved ambulances and stretchers around for the ambulance crews and ssisted n the ambulance to the Cranbrook Regional Hospital.

This statement is accurate to the best of my knowledge.

Respectfully submitted

#### **MEMO**

То:	1	File:
From:		

Date:

Thursday, May 18, 2006

Subject:

May 17<sup>th</sup> confined space rescue

as requested, I will summarize my recollection of yesterday's events.

I initially heard the call via the City radio outside my office and immediately went to the fire hall to see if my assistance was needed. The engine was parked in the street facing east when I got on. You were officer, was driving and ...nd myself were in the back. On the way up we were informed that the rescue will be a confined space and that we were to get our BA's on.

Upon arrival at the scene, and I took our BA's off to get our harnesses on when we discovered that there was only one harness on engine elected to put the harness on and who was already on scene, suggested the he go into the chamber, with his BA on the evaluate the situation. Chief Collinson agrees and I go with to act as safety for him. I had my BA back on at this point. Once in the chamber, indicated that there was 4 victims that he could see and that they were

By this time had his harness on and he and acted as the rescue tean in the bottom of the chamber, both tied off on life lines. I acted as safety for both of the guys inside the building on the platform above.

With this crew we managed to get the female paramedic out first.

If an out of air so wapped places with and he and worked to get the next paramedic out. Basically the two guys below would lift the victim to me and I would pull them to the door at which time the crew out side would transport to the paramedics rescue station.

The ambulance crew had set up a staging area about 30 metres away from the building.

For the next two victims and (These guys arrived on Engine from Marysville Station) were the rescue team in the hole. I changed bottles between the first two victims and the second victims. ssisted me as the safety person on the platform above the chamber for the next two victims. Bob Newcombe was the third victim brought out and finally Doug Erikson was brought out. Doug was

The ambulance attendants were to three of the victims during and after the rescue. Ambulances were arriving and taking the victims to the Cranbrook Hospital. and iraveled in the ambulances to Cranbrook to assist with

From a timeline point of view, Doug

Weather was hot (+25 C or so) and clear at the time of the rescue.

Other people on site at the early stages were (Kimberley Fire department), and another paramedic, and the Cominco More ambulance attendants arrived after the rescue to transport and the RCMP showed up ( and another constable).

Although we didn't follow our normal procedures for confined space of setting up tripods etc., I felt that we were very safe at all times. All of the rescuers were tied off and I was acting as the "extractor" for our guys should something happen. We encouraged our guys to keep talking, ensured that all the guys had good BA seals and were very careful in the tight quarters.

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Statement of observations and actions for on May 17<sup>th</sup>, 2006.

during the Sullivan Mine Deaths

Role in incident: Firefighter, Kimberley Fire Department.

I am unsure of time of arrival on scene.

We staged our engine (engine # , within 100m est. of the incident scene. My crew partner , and I were advised by our truck officer, , that our role was to don SAR (supplied air respirator) and assist in the rescue of an number of victims, possibly four. Approaching the scene, I observed 2 BC Ambulance, a generic van labeled "ambulance", Kimberley Fire Dept. engine , and command vehicle, as well as a Tech Cominco pick-up and a Nissan Frontier pick-up. At this time two BCAS personnel, and the onsite first aid person appeared to be establishing a triage/treatment area at a distance of roughly 35 feet from the door to the shed. The four KFD entry personnel on were in the shed with SCBA and coveralls. Two of the four were visible within one to two feet of the door, and the two others were inside the lower compartment. Outside the shed I witnessed KFD members assisting with the entry teams, managing ropes and waiting to assist with moving victims. The entry teams were harnessed and tethered.

My partner and I donned our harnesses and SAR with the assistance of driver) and (engine officer). As we completed donning our equipment the first victim was being removed. This person was the female BC Ambulance attendant. As I was not in my PPE (personal protective equipment), I was unable to assist in the carry of the first victim. During the carry of the first victim we completed the donning of our PPE.

and I were then instructed to aid in the lift and extraction of the remaining victims.

Upon entering the shed, the air was clear (no mist, steam or other visible airborne matter). The crew in the lower compartment were lifting the male BCAS attendant. The upper crew were pulling up this victim. My partner and I were assisting with this lift. We carried this victim approximately 10 feet from the shed until adequate personnel took him from us. The low air alarm on the SCBA of one of the crew in the lower compartment began to sound and that crew removed themselves. The crew in the upper shed then moved into the lower compartment to remove the Cominco employee.

and raised by the lower crew and and myself. Then carried to the triage by area by unknown personnel. (to me)

At this time the crew in the lower compartment has a low air alarm and removed themselves from the compartment. A back up team was established and and I then entered the lower compartment. Before climbing entirely down, a KFD member reminded us not to break the seal on our masks as the atmosphere was IDLH. (immediately dangerous to life and health)

Initial observations of the lower compartment were:

Culvert from the rear of the compartment, with water flowing into the compartment.

Estimated 12 inches of water pooled.

Water flowing out of the compartment at the front.

Clear water.

My partner, checked the victim and found him - We then notified the surface crew this will be a recovery so resources could be used to aid in the resuscitation of the victims removed. Removal of the victim was done with a rope and lift assist by myself and surface personnel. Our crew then confirmed no further victims before exiting the compartment.

Upon exit if the shed we then removed our PPE and attempted to aid the BCAS with victim care.

All victims were removed by BCAS, with the exception of the man

I then loaded equipment, marked the shed with fire-tape, and left the scene.

There was no sulfur smell at any time during this event.

This is a statement made to the best of my recollection on this day of May 18, 2006.

Regards,

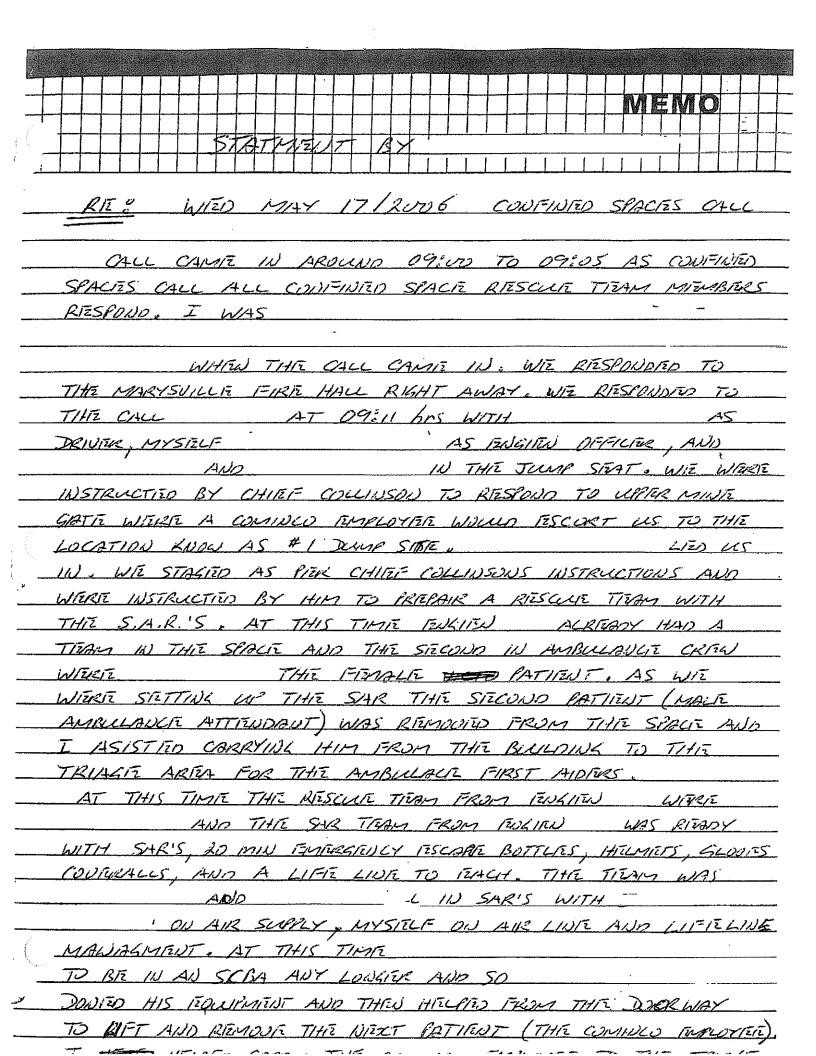
PRIAGE EQUIPMENT (AIR, SAR, ELEVATED

#### Confined Space Report by

While I was working on Wednesday, May 17, 2006, I also was on monitor for the Fire Hall. I heard a call for a . I started to respond then the call came in as a possible 4 person down. Command called for confined space response. As I responded I called to see where he was so he could bring engine the scene as it has confined space rescue equipment. I arrived at the Kimberley Fire Hall and proceded to the TECK property below the old entranceto the mine. I put on a safety harness and breathing apparatus. When I was given the go ahead, I entered the small building with and together we removed the female attendant. As we started to remove the male attendant, air started to run out. I tapped him on the shoulder and told him his bells were going off which is the signal to let us know when the oxygen supply is running out. exited the confined space and I proceded to male attendant off so we could get him out. Following that, my air was low, I I exited the confined space.

then assisted in loading the patients into the ambulance. When everyone was removed from the confined space, we returned to the Fire Hall. A short time later, I returned to the scene to do some air quality readings with some of the other fire men for the Mines

Inspector.



I WAS NOW IN CHARGIE OF THE AIR SUPPLY FOR THE
SAR'S ASWILL SIE AND LIFELINE MANAGEMENT . WITHIN MINUSTIES
HATE KIESCUE TIEAM REMOVED THE LAST PATIENT (THE
CONTRACTER) AND I HELPED TO CARRIE HIM TO THE TRIBGE
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DID. I DID NOTICE THAT
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THE PATIENT INTO THE AMERICANCES FOR TRANSPORT WITH
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COLLINSON AND PHOTOGRAPHING THE SCIENTE.
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AND AL COLLINSON WIENT TO THE TOP OF THE HILL
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AND IZSTARLISHIN A SAFIE ZOWIE AT THIS TOP OF THIS HILL.
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ALFA SIDIZ AT THIE DOOKWAY AT WAIST HIGHT WIE GOT 10%
OXYGIED LIEVIELS. I THEN REPOCHED INTO THE SHIED 3' AND
THE OXYGEN LEVEL DROPPED TO 7.0 % WIE THEN RETURNEDED
BACK TO THE SPAFE ZONE , THE SECOND TEAM RETURNED TO
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THE 20 METTERS
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RCMP, AND CORRINDE THEN MONTED IN TO INSPECT THIS SCENT
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IN THIS IZNTRANCIE OF THIS SHIPD NITTER THIS LOCATION OF THIS
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FULL SCBA'S OU - SHORTLY AFTING THAT THE
AND WIFE HAUTE
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FIRST HALL . JUST PRIOR TO ITAVING AN AIR TESTING CREW
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UPPON RETURN TO THE HACK WIE ALL MET WITH A
CRITICAL INCIDIENT COUNCILEUR FOR ABOUT 1'2 hos.

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#### To Chief Allen Collinson;

The following is my statement of the events of the "confined space " rescue of May 17th, 2006. I received a page-out calling for a confined space team to the upper mine site. I arrived at the Marysville station moments later and responded with ind -.We arrived on scene to find command, engine ,2 ambulances and two civilian vehicles. We brought out the confined space gear and proceeded to set it up and myself donned harnesses as . and set up the air supply manifold. As we were doing this the first victim was being removed by the first teams to arrive. When we had our gear on and were ready to go, we assisted in moving the second victim to the paramedics.At this time team one members were running out of air.I entered the structure followed by -.I lowered myself into the space below where was trying to secure the third victim. I helped to secure and raise the victim to ...I then proceeded to secure the final victim.At this point,\* exited the lower space and came down to help.I managed to

The other members of the team above took him and exited. I then did a quick search of the space and f : which I tossed up to the entrance and I then climbed out and exited.

Upon leaving the building,I took off my mask and air lines and proceeded to assist paramedics with Shawn. and myself helped load Shawn into an ambulance.We were then asked to proceed to Cranbrook with the patient.We to the patient along with a paramedic until we reached the emergency room.We were there until arrived and drove us back to Kimberley.

Kimberley Fire Department

May 18, 2006

#### To Whom It May Concern:

#### Re: May 17, 2006 Statement of

- Approximately 900 hours, I drove Engine to the scene with and and the rear seats. On route to the scene, we were informed of four casualties in a confined space. This information was relayed to us through
- Upon arrival to the scene, myself, and proceeded to don our Self Contained Breathing Apparatus. Through this procedure, we were receiving direction from Allen Collinson, Kimberley Fire Dept. Fire Chief.
- ad myself proceeded to join another fire fighter, at the outside entrance of the Water quality Station. At this time, we performed a safety check procedure prior to the rescue.
- ad and I had the life lines in our hands.
- went part way down the ladder and confirmed that the bodies were there. At this point, bserved that the handrail was in the way. He ripped it out to aid in the rescue procedure.
- proceeded to the female paramedic. Id I assisted in the lift to the surface where there was Paramedics and Fire Fighters waiting for her.
- :ttempted to retrieve the second Paramedic, however his low air alarm sounded and he exited the building.
- whom was behind went down the ladder and the second Paramedic myseir and other rescuers assisted to take the body out to awaiting Paramedics and Fire Fighters.
- I proceeded to go down the ladder next. I a third casualty male. ssisted me in bringing up the third casualty to awaiting Paramedics and Fire Fighters.
- My air supply at this point was at around half a tank
- At this point, and . went down to retrieve the last casualty. I assisted with lift of this casualty to the awaiting Paramedics and Fire Fighters.

.

- Once the recoveries had taken place, I
- I assisted with loading the second and third casualties onto the ambulance stretchers.
- I was very confident with the entire recovery procedure of this situation. The communication between the team members involved was very professional and effective.

Sincerely

We arrived on scene with 4 men on board.

I went down to the building and started to set up air packs.

After setting up I put on a SCBA and switch out with

I went into the building and help to Dring out Last 2 people.

After all the people were out we started putting people into

ambulances. We then cleaned up and

# BRITISH Ministry of Health and Ministry Responsible for Seniors

# EMERGENCY HEALTH SERVICES COMMISSION OCCURRENCE REPORT

2 0,5 2+0,0,6 1,7 0,5 2+0,0,6 P-247558 CARNO.				PAGE	OF-
Flecher, Peter  Sullivan mine Response  0930 · Call Fram Dispatch to Attend to the  Sullivan mine in Kimberly as I/C. Dispatch  States Two mine people down to a crew members  down Possible H2S Gas. Dave Brooks  Flying back from Kamploops with Aireauce Team.  0950 · Lift off Invernere Airport to Kimberley  (Airspan 2061R). Talked to Pilot  wat that Gas states he had Training in the  ras due to Flying around fact wells.  1025 · Landed at Sullivan mine 2 Ramp and  other person on Scene one code 4 Talked to  Cpl , 2 Ambulanes People and one other  person on soute to Hoop in Cranbrook CPR in Progress  Cpl States all Its had Passed away, Dispatch Notified  Tasked  Dank  Verents Site  Weatherd by to Ramp.	(DD / MM / YYYY)  12   0   5   2+ 0   0   6   1   7   9   5   2+ 0   0   6	PESPONSE NO. P = 2475.58	OPERATOR NO.	<u> </u>	<b></b>
Hecker, Peter  Sullivan mine Response  0930 - Call from Dispatch to Attend to the  Sullivan mine in Kimberley as I/C. Dispatch  States Two mine people down to 2 craw members  down . Possible H2S Gas. Dave Brooks  Flying back from Kamploops with Avravac Team.  0950 - Lift off Inverner Airport to Kimberley  (Airspun 2061R). Talked to Pilot  aut H2S Gas states he had Training in the  ras due to flying around Gas wells.  1025 - landed at Sullivan mine 2 Remit and  other person on Scene one code & Talked to  Cpt . 2 Ambulance leoph and one other  person on route to Hosp in Cranbrook CPR in Progress  Cpt States all Its had Passed away, Dispatch Notified  Task  Bank X Landons Site  Walked up to Remit	EA	EMPLOYEE NO.	CAR NO.		
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	walk	ed up to Kemi.			
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	ÖRE(S)		1 2 1 2		



# EMERGENCY HEALTH SERVICE COMMISSION OCCURRENCE REPOR

DATE (DO / MM / YYYY)			PAGE	OF
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TTENDANT	EMPLOYEE NO.	CAR NO.	<del></del> .	
Hecher	EMPLOYEE NO.	-	÷	
E: (cont)				
10:30 - Lift off for cran	boule 11.0			· · · · · · · · · · · · · · · · · · ·
10:40 - Landed at Compression	11-120			
10:40 - Landed at Crambrook To Admin building Saw	Hellpad, at 1	osp. Wa	1 ked	<u></u>
Talked to	for about 1	min (u	rent	
inside and Saw	saul he	told me		
that Kim Westzel is Dec	c0 $c$ $c$ 0 $c$ 1 $c$ 2 $c$ 3 $c$ 4 $c$ 5 $c$ 4 $c$ 5 $c$ 4 $c$ 5 $c$ 4 $c$ 5 $c$ 6 $c$ 7 $c$ 7 $c$ 9		<u> </u>	
heartene on Should a	1 0 0			
colled a	for Disputch to	re lec se		
		12130		
went to the ER end to	alked to Do	3		
the Stated all they had		· · · · ·		
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Foundation ferom were on	11.	them know	<u> </u>	_
call talked to	The way wer	t buck		
with Airever team	. Dove Brod	ks and	ecl	_
Dave asked me to Ta	and			
People.	and Get some	info E	OM	_
	-10-			_
Company, Bob Carled	writes for Sullive	n Closer		_
- I wanted	and Person	to Die)		
[ Caller	I. Then hung u	y the P	Lune	
of the form	ret Eve Dept	told th	en	
(S)	DAYE (D	0/MM/YYYY) L[6,5 20	06	1
			10 10	1

# BRITISH Ministry of Health and Ministry Responsible for Seniors

# EMERGENCY HEALTH SERVICE COMMISSION OCCURRENCE REPORT

DATE (DD / MM / YYYY)	PAGE OF
DATE (DD / MM / YYYY)  212   O   5   210   0   6   17   0   5   21:010   6   EMPLOYEE NO.	1 3 17
SHIVER EMPLOYEE NO. CAR NO.	**************************************
TTENDANT	
Feche	
E: (Cont)	
Where to go. Mest someone driving out in Pick	
with coming markings told possible	cup_
gas not to go in.	<u> </u>
OAL CScene	
ON Scene found Ambulance running with lights one He found Baga on top of tathe yelled into hole No Answer Backed and Ex	
with lights one He found Baga on top of fath	diler E
yelled into hole No Answer Backed away. Fre	Deat
on Scene he stopped to fire men that wanted to	
them on Road state acrued on Scene mure person	
them on Roud Stated Has Gas not to go in Fire	meet
States 6 full 1 1111 1 10 go in Fire	<u> </u>
States 6 foot hole, little building 6x6? Jump bug	
and no one considered sil	+
The think the terms of the term	
a tiretichter was all a	
one at a time kim lie to I be it	our
Currier 2nd Bob 3rd Dong 4th out en	aun
one recieved	iery
on 4th Person (Don	<u>-5)</u>
World .	
DATE (DD / MM / YYYY)   2,12,10,15,12,10	- 4
	3 10 10



### Ministry of Health and Ministry Responsible for Senlors

# EMERGENCY HEALTH SERVICE COMMISSION OCCURRENCE REPOF

		PAGE OF
2,2 0,5 2,0,0,6 1,7 0,5 2,0,0,6	YY) RESPONSE NO. P- 247558	OPEHATOR NO.
	EMPLOYEE NO.	CAR NO.
Hecher	EMPLOYES NO.	-
E: (Cont)		
	Kimberley.	
Disputch Called		12 CC) Jan
- met Person at Scene	? told them a	2 mine
and 2 Ambulance staff	Nonted In In	Passin ent
up trace area with stars.	Kurn out 1st	veseue ser
Shown out next	(they solt) ,	· <del></del>
order they came out they we	ere transported (	Sharran Campaga
Orpue.	. +	· ·
	***	<u>.</u>
Cur	kimberly, Talked	to
in building Jump has a	sel on on the	
welted for the lest with	Scha ropes har	nes Fire went
	١.5	in lake?
Turned lights off in 1st Am	brilence,	
- Drove buck		
on Kim weitzel		-
Duspatch ca	cet hom	e
more worker down 2 farament	lies went after th	om c= 0
he was going up there At	Scene 3 people	
·		<del>*************************************</del>
		Lice souls
PREISTAT ORCH	DAT	tire people = (00/MM/YYYY) 12 0,5 2,0,0,6
		12 40 0 0

# BRITISH Ministry of Health and Ministry Responsible for Seniors

# EMERGENCY HEALTH SERVICES COMMISSION OCCURRENCE RÉPORT

			PAGE 5	of 7
TE (DD / MM / YYYY)  -12 0 5 20 0 6 17 0 5 20 0 6	PESPONSE NO. 247558	ОРЕНАТОЯ НО.	<u> </u>	
iyek	EMPLOYEE NO.	CAR NO.		· · · · · · · · · · · · · · · · · · ·
TENDANT He che	EMPLOYEE NO			
(cont)			· · · · · · · · · · · · · · · · · · ·	
	. Fire brough	+ 4+h	0ersur	<u> </u>
out d	?on_		<u> </u>	
Drave (car)canbrook)	-	•	42	<b>.</b>
cranbrook ist car that left.				
	ar Cranbros	k caller	1 at	
904 to mine in Kim	berley 3rd car in	1 3 Pts 01	1 Sce	ne
	Carled Dispatch	nform	ed the	m
crew down and one ook	er Pt.	•		ŀ
idrove +2 firefig	htvs	on 3 nox	Yersur	ι?
-At Stop Sign /2 was to co	inbroat	S		
- Called on	crenbrook at	0609		
to Sta, 10 mm at sta Dupa	tala Phoned Cr	م <u>در رمن</u> و	7	
Kimberly coutie. 10 min or	t of Crapbrool	Pispat	-ch	
Called them 4 Pts at the n	one to	mine a	4	
- UTOU, GOT There as 1"	Ambialence 1.20 c	ecuric	الماللم	
1st Pt (kim) Pulled in got o	ut drove	one Am	bulene	e
back (3rd pt ?)				
Cralle	el on c	ranbrook	9:00	
TO SAN UNKnown (	aged again In	n later	Sard	
Stn Croweage 10 min at Str	Phoned, head	to kumb	erley	
2055- cover At field,	A N 1	TE (UD / MM / YYYY)	<u>~e</u>	
AE(S) for Whele	- OA	LA: 0 ,5   6	200	6
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2403 Rev. 08/11/10

### Ministry of Health and Ministry Responsible for Seniors

# EMERGENCY HEALTH SERVIC COMMISSION OCCURRENCE REPO

		PAGE OF
DATE (DD / MM / YYYY)  2,2 0,5 2,0,6 1,7 0,5 2,6 0,6 6	Y) RESPONSE NO. P-247558	OPERATOR NO.
ATTENDANT	EMPLOYEE NO.	GAR NO.
Hecher	ÉMPLOYEĘ NO.	
RE(Cont)		····
	r as they acq	
Scene Prove 1/2	very then S	undeted with
2 Newly ?		
1330 - and I	at Crenbral	Sto boat
13:00? - called kumberley str to get Dave Brooks up here	triked to	+ states
robout what he a send -	tourd a	t on Radio
Long Garland C	Lalling live of	o corwer
Cosked A I could	or hipatch coul	ed and
for kimberley Str. He would	of England	is coverage
Str. W/c office locked unable	- 40 Course L	ts to Cranbrook
Dispatch couled and asked	all come	- Called
Storte Couled and asked it	T Could	open office.
Str to help crew. 1520. let	+ for Kimberley	1 To kimboly
- at Airpor	to 1605 at ki	7 Lagred
alled D. i.	only 2 people	the stay san
wheel Doug Earland to fex he reported a crew member of Cover night Shift Started	its to kimber	es do
overage one car covered	as drung up	to kumberle
overage One Cox covered	working on	Getting
father covered	until Sinda	Am.
	DATE	2 10,5120016
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Ministry of Health and Ministry Responsible for Seniors

# EMERGENCY HEALTH SERVICE: COMMISSION OCCURRENCE REPORT

		PAGE OF 7
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TENDAN' Hecher	EMPLOYEE NO.	
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at Str in Inverser	Drove Back our	· Zune Source
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Kimoz 1	SGENR NOTES
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10:25 - Landed Kimberlay millocode 4. Talked to CPL one other person on to route	2 Ambalance People and
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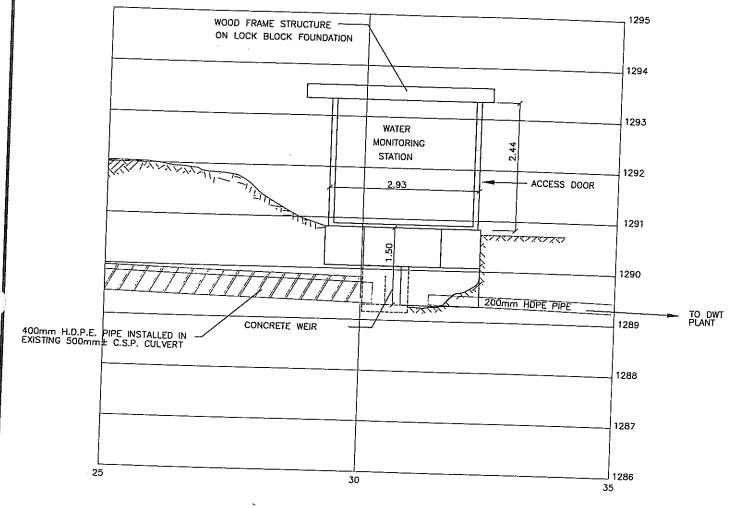
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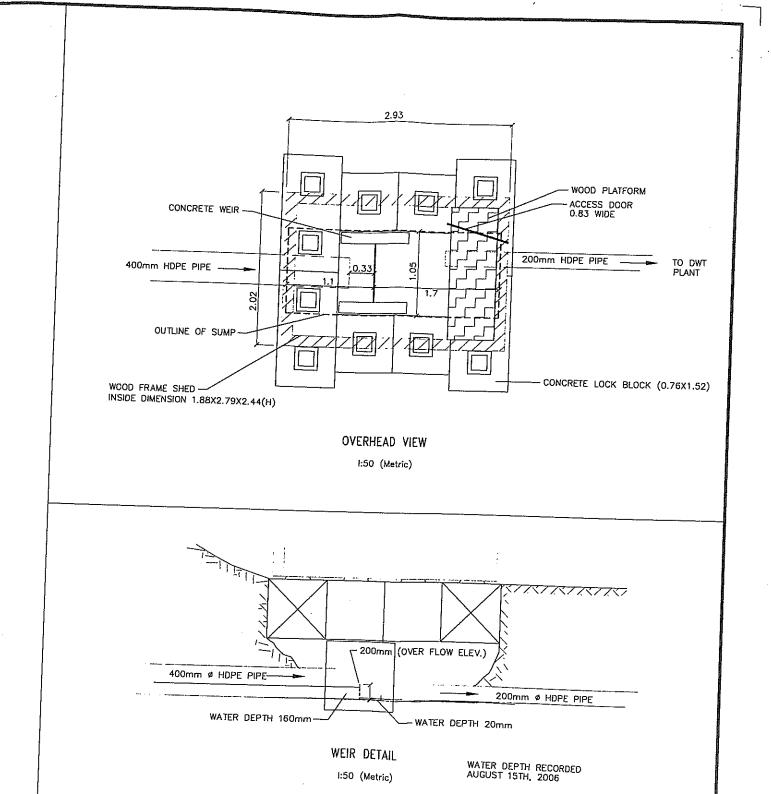
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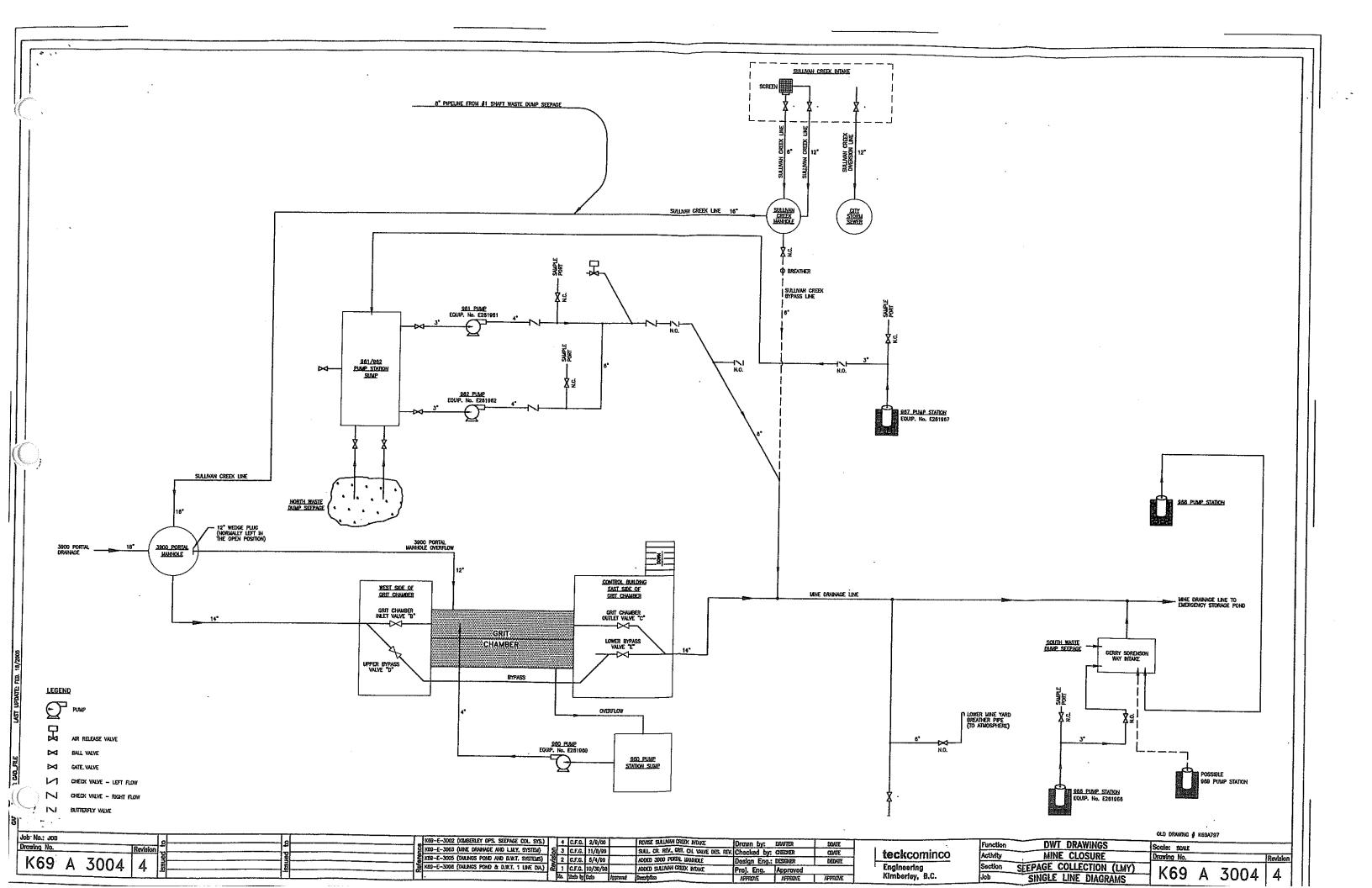
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### The University of British Columbia

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# Sullivan Mine gas samples: Isotope Tracing: A Preliminary Study

June 14, 2006

### 1.0 Background

Four samples were delivered to UBC-CERM3 for isotope tracing, in connection with the Sullivan Mine accident on May 17, 2006. The samples were collected from the water sampling location where the accident occurred. The samples and their composition (as analyzed by Cantest) are shown in Table 1.

#### Analysis of Components in Gas

CLIENT SAMPLE IDENTIFICATION:	Downstream Pipe	Upstream	Middle	Bottom		
		1			DETECTION	UNITS
CANTEST ID:	605180109	605180112	605180113	605180114	PTIMIT I	}
Methane	10.7	12.0	11.4	9.5	2	mL/cu. m
Carbon Monoxida	<	<	<	<	<b> </b>	mL/cu, m
Carbon Dioxide	7.0	7.0	7.0	5.0	1 1	% by vol.
Oxygen	2	2	2	3	1	% by vol.
Nitrogen	91	91	90	89	1 1	% by vol.
Nitrous Oxide	13.3	12.5	12.8	12.9	0.3	mL/cu, m
Trichiorotrifluoroethane	<	<	<	<	0.1	mL/cu. m
Chlorodifiuoromethane	<	<	<	<	0.05	mL/cu. m
Dichlorodifluoromethane	<	<	<	<	0.1	mL/cu, m

mL/cu, m = mL/cubic meter or ppm (v/v)

< = Less than detection limit

% by vol. = percent by valume

Table 1: Gas samples collected from Sullivan Mine site and their Composition (as analyzed by Cantest)

It can be seen that carbon dioxide is a major component of the gas samples. Identifying the source of this gas is believed to provide an important piece of information in the analysis of the situation. The sampling shed (where the accident occurred) is located at the toe of the west dump (WD1) which is more than 50 years old. Considering the history of the dump, it has been postulated that carbon dioxide may have originated from organic material (known to be buried in the dump). An alternative source for the carbon dioxide is inorganic carbonates which are reportedly present inside the dump and which act as buffering capacity for Acid Rock Drainage. To distinguish between these two possible sources, an analysis for carbon-13 was performed.

### 2.0 Theoretical basis

Carbon has two naturally-occurring stable isotopes:  $^{12}$ C (98.89%) and  $^{13}$ C (1.11%). The ratio of these isotopes ( $\delta^{13}$ C) is reported in ‰ relative to a standard sample that is commonly the standard VPDB (Vienna Pee Dee Belemnite).

$$\delta^{13}C = \frac{{}^{13}C / {}^{12}C_{sample} - {}^{13}C / {}^{12}C_{standard}}{{}^{13}C / {}^{12}C_{standard}} \times \frac{1000}{1}$$

The  $\delta^{13}$ C of the atmosphere is about -7‰, similar to the standard sample.

Typical plants, soil organic matter, soil  $CO_2$ , marine organic matter, and fossil fuels tend to have  $\delta^{13}C$  values in the range from -20 to -30%. This depletion of  $^{13}C$  occurs because the process of photosynthesis—on land or in the sea—preferentially uses the lighter carbon from the available  $CO_2$  pool. The carbon fixed by plants will therefore have a  $\delta^{13}C$  value that is less than that of the source  $CO_2$ . This carbon shift is generally in the range of -20 to -30%.

Trumbore & Druffel, (1995) have given the isotopic range of different sources of carbon as shown in Figure 1.

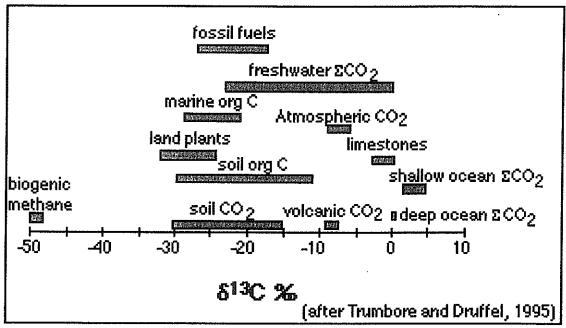


Figure 1.  $^{13}\text{C}/^{12}\text{C}$  ratios are expressed relative to a standard as  $\delta^{13}\text{C}$  values,

Note that limestones show the least shift (with  $\delta^{13}$ C ranging from 0 to -4) while organic carbon, on the contrary, is distinguished by a comparatively large shift (from -15 to -30, in most cases). This bimodal distribution of carbon isotopic ratios provides a valuable tool to distinguish between organic and inorganic sources of carbon. (Kump, 1991, Kwon & Schnoor, 1994, Trumbore & Truffel, 1995).

### 3.0 Isotope data and interpretation

The four gas samples were shipped to the Hatch Isotope Laboratory in Ottawa for isotope analysis of carbon dioxide. Oxygen-18 and carbon-13 were analyzed in the samples. The data are presented in Table 2.

OurLabID	Sample ID	$\delta^{13}$ C	δ <sup>13</sup> C Comment	δ <sup>18</sup> Ο
C-20993	S#605180114	-8.58	0	-10.47
C-20994	S#605180112	-8.61	0	-10.38
C-20995	S#605180113	-8.87	0	-10.52
C-20996	S#605180109	-8.61	0	-9.91
C-20997	S#605180109 QC Duplicate	-8.28	0	-9.71

Table 2. Oxygen and carbon stable isotope values of CO<sub>2</sub> from Sullivan Mine gas samples.

Note that all samples demonstrate relatively the same isotopic signature. The downstream sample (# 09) is slightly heavier (i.e. richer in both O-18 and C-13), but the difference is insignificant. The carbon shift is too low to attribute the source to organic matter. The shift is close to that of carbonates or freshwater CO<sub>2</sub>. While an organic source for the gas can be ruled out, a clear link to waste dump carbonate cannot be established with the present data. Isotopic analysis of carbonates present in the waste dump is necessary to confirm this correlation. Isotopic values of CO<sub>2</sub> in local fresh waters are also needed for the final analysis, since a portion of the CO<sub>2</sub> may be linked to recharge waters.

Similarly, the oxygen-18 values of the samples can be linked to the source of the carbon dioxide. In general, the oxygen isotope composition of organic material is determined by the isotopic composition of the source soil and water. Significant enrichment in  $\delta$  <sup>18</sup>O is commonly observed in most organic material. The main enrichment takes place in leaf water due to transpiration.

At first glance, the oxygen-18 of the samples appears too low to speculate on organic sources. The ratio is similar to that of ground waters. A comprehensive stable isotope study including ground waters and waste rocks is necessary to establish a clear link between isotopic values of the analyzed rocks and the source material.

Considering that the sampled carbon dioxide may also be carried by mine effluent, different water bodies (contributing to the waste dump recharge) and local precipitations should as well be analyzed to find their characteristic O-18 and Deuterium values. These isotopic signatures help (in combination with chemical signatures) to find the source of effluents in mining environments (Ghomshei & Allen, 2000a, 2000b, Lepitre et al, 2000).

### Respectfully submitted:

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Email: mory@interchange.ubc.ca

Adjunct Professor of Mining Engineering

### 4. References

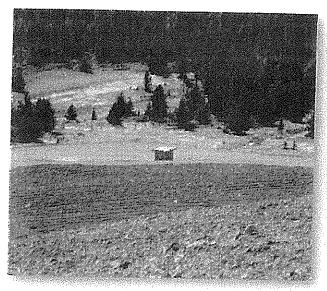
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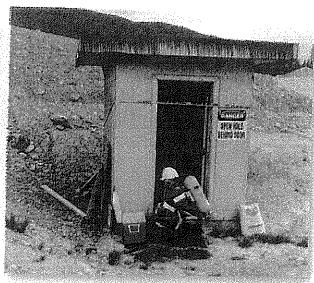


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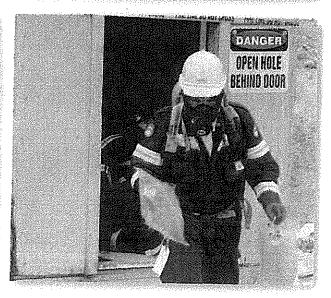
# Sullivan Mine Incident Investigation May 15 to 20, 2006

Independent Air and Water Quality Sampling and Measurements









August 2006



## EXECUTIVE SUMMARY



## **Executive Summary**

The investigation into the four tragic fatalities at the decommissioned Sullivan Mine on May 15 and 17, 2006 showed that the incident was caused by oxygen-depleted air displacing ambient air inside a shed. Oxygen-depleted air produces no warning symptoms, such as dizziness. The oxygen concentration inside the shed was approximately 2%. At oxygen concentrations below 6%, two breaths can cause a person to lose of consciousness, with death following within minutes (EIGA, 2003).

The shed where the incidents took place was constructed over a waste rock leachate collection sump at the toe of a topsoil covered waste rock pile. The sump was constructed down-gradient from a culvert and a drainage pipe extending into the base of the waste rock pile. The oxygen depletion was caused by sulphide oxidation reactions within the covered waste rock pile. The topsoil cover was placed on the waste rock pile by Teck Cominco in an effort to curb acid rock drainage (ARD) by impeding oxygen transfer to the sulphide-containing waste rock.

The acid rock neutralization reactions produced carbon dioxide, which partially replaced the consumed oxygen. The cooler and denser oxygen-depleted air inside the waste rock pile flowed down-gradient through the culvert and into the shed. Increasing daytime air temperatures, falling barometric pressure, air composition differences and local air pressure differences provided the driving force for the down-gradient air flow.

The widely published accounts of the Sullivan tragedy have helped to increase awareness of the very real hazards associated with working in and around confined spaces. Future efforts should be focused on strengthening awareness and enforcement of section 3.4 of the *Health, Safety and Reclamation Code for Mines in British Columbia* (BC MEM, 2003). The existing regulations provide clear guidelines for appropriate working procedures.

However, identification of confined spaces may not be sufficient to prevent future incidents. Whether or not it flows into a confined space, any culvert or large pipe with an appreciable outflow of oxygen-depleted air constitutes a hazard. An unwitting person looking into a pipe, culvert, or other vent that is discharging oxygen-depleted air could succumb to asphyxiation in a few seconds. Therefore, special attention must be paid to any openings that allow for air transfer between a covered reactive waste rock pile and the ambient air. A safe perimeter around the openings should be fenced off and warnings should be clearly posted. Established procedures for working in and around confined spaces should be followed when accessing such sites.

## TABLE OF CONTENTS



# Sullivan Mine Incident Investigation

## **TABLE OF CONTENTS**

Exec	cutive S	Summaryi		
Table	e of Co	ontentsii		
	List	of Appendicesii		
	Liet	of Figures		
	Lint	of Figures		
	List	of Tablesiii		
	List	of Platesiii		
1.	Ove	rview and Scope of Investigation1–1		
2.	Field Sampling Methods2–1			
	2.1	Air Quality2-1		
	2.2	Water Quality2-2		
	2.3	Microbial Activity2-3		
3.	Results3–1			
	3.1	Air Quality3–1		
	3.2	Water Quality		
	3.3	Water Quality3-5		
	3.3	Microbial Activity3-8		
4.	Analysis and Discussion4–1			
	4.1	Causes of Incident 4-1		
	4.2	Prevention of Future Incidents		
Refere	ences	R–1		

### LIST OF APPENDICES

- Appendix A CANTEST Air Quality Analysis Report (Rescan, May 20, 2006)
- Appendix B CANTEST Air Quality Analysis Report (Teck Cominco, May 18, 2006)
- Appendix C ALS Air Quality Analysis Report (Ministry of Energy and Mines, May 18, 2006)
- Appendix D ALS Water Quality Analysis Report (Rescan, May 20, 2006)
- Appendix E MicroMed Microbial Analysis Report (Rescan, May 20, 2006)

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LIS	ST OF FIGURES	
Fig	ure	Pag
4-1	Hourly Barometric Pressure and Temperature Recorded at Cranbrook Airport (May 14 to 21, 2006)	_
LIS	T OF TABLES	
Tab	le	Page
3-1	Air Samples Collected by Rescan on May 20, 2006 at 8:00 AM: Summary of Analytical Results (CANTEST)	_
3-2	Air Samples Collected by Teck Cominco on May 18, 2006 at 9:00 AM: Summary of Analytical Results (CANTEST)	3–3
3-3	Air Samples Collected by Ministry of Energy and Mines on May 19, 2006 at 1:07 AM: Summary of Analytical Results (ALS)	3–4
3-4	Sump Water: Analytical Results	
3-5	Results of Microbial Activity Analysis	
LIST	OF PLATES	
Plate		Page
1-1	The shed at the decommissioned Sullivan Mine where the asphyxiation deaths of four people occurred on May 14 and 16, 2006.	·
1-2	The shed where the fatalities occurr ed situated at the toe of the reclaimed waste rock pile.	
1-3	A PVC pipe at the bottom of the sump inside the shed; the pipe is installed in a culvert (not visible) that drains waste rock seepage to the sump. The pipe was the conduit for exhausting the oxygen-depleted air.	
1-4	End view of the partially sanded PVC pipe at the bottom of the sump; the culvert leading into the bottom of the waste rock pile surrounds the pipe.	
1-5	The outflow pipe at the bottom of the sump.	
2-1	Teck Cominco HazMat team members collecting air quality samples from the shed at the Sullivan Mine (8:00 AM, May 20, 2006)	
2-2	Teck Cominco HazMat team members collecting water samples from the shed at the Sullivan Mine on May 20, 2006	2_2

## 1. OVERVIEW AND SCOPE OF INVESTIGATION



## 1. Overview and Scope of Investigation

On Monday, May 15, 2006, Douglas Erickson

routine visit to an enclosed sump at the decommissioned Sullivan Mine in Kimberley, B.C. Two days later, on May 17, 2006, Teck Cominco employee Robert Newcombe

site while attempting to rescue Erickson. Paramedics Kim Weitzel

subsequently succumbed at the site while attempting to rescue Newcombe and Erickson. Later in the afternoon of May 17, 2006, the four victims were removed from the shed by fire fighters from the Kimberly Fire Department, who were equipped with self-contained breathing apparatus (SCBA).

The British Columbia Ministry of Energy and Mines (BC MEM) and Teck Cominco promptly launched an investigation into the tragic incidents. The initial investigation focused on collecting data that would help determine the cause of the incidents. Teck Cominco collected air samples from the scene of the incidents on the morning of May 18, 2006. Another set of air samples were collected at the site by BC MEM on the night of May 18, 2006. All sampling was carried out by Teck Cominco's volunteer hazardous materials (HazMat) team.

Rescan Environmental Services Ltd. (Rescan) were asked by BC MEM to conduct a site inspection on May 18, 19 and 20, 2006 and to collect an independent set of air and water quality samples from the site of the incidents. The samples were intended to supplement those collected by Teck Cominco and BC MEM on May 18, 2006.

The fatal incidents took place in a shed (2.41 m high, 2.02 m wide, and 2.02 m long) constructed over a waste rock seepage collection sump at the toe of a soil-covered waste rock pile (Plate 1-1 and 1-2). The sump was 1.68 m below the floor level of the shack. The waste rock pile had been covered by topsoil in early 2006 by Teck Cominco as part of the reclamation efforts at the decommissioned Sullivan mine. The inflow to the sump consisted of a culvert with a partially buried PVC pipe installed inside it (Plate 1-3 and 1-4). Before the topsoil cover had been placed on the waste rock pile the culvert had drained water from a French drain constructed along the perimeter of the waste rock pile. The outflow of the sump drained through another PVC pipe to a catchment sump for delivery to the St. Mary's water treatment plant (Plate 1-5).

The objectives of the air and water quality sampling were to determine the composition of the air inside the shed and the seepage collection sump and to characterize the seepage water running into the sump. In addition, samples for microbial activity were collected.

This report describes the field sampling methods and analytical results for the collected air and water quality samples. The chemical and physical processes that likely caused the oxygen depletion at the scene of the incidents are discussed and recommendations for preventing similar fatalities are presented.

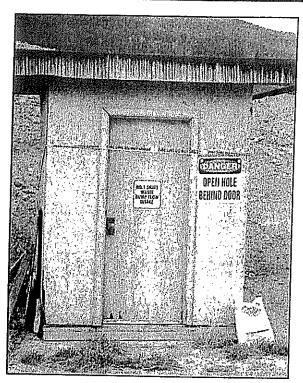


Plate 1-1 The shed at the decommissioned Sullivan Mine where the asphyxiation deaths of four people occurred on May 14 and 16, 2006.



Plate 1-2 The shed where the fatalities occurred, situated at the toe of the reclaimed waste rock pile.

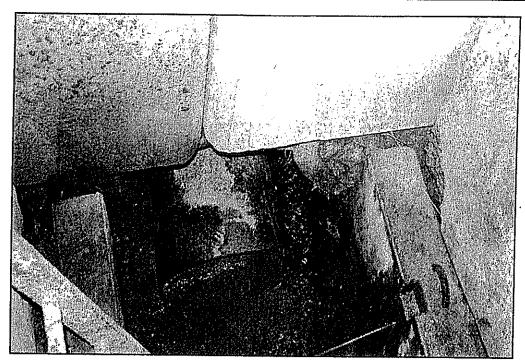


Plate 1-3 A PVC pipe at the bottom of the sump inside the shed. The pipe, which was the conduit for exhausting the oxygen-depleted air, is installed in a culvert (not pictured) that drains waste rock seepage to the sump.

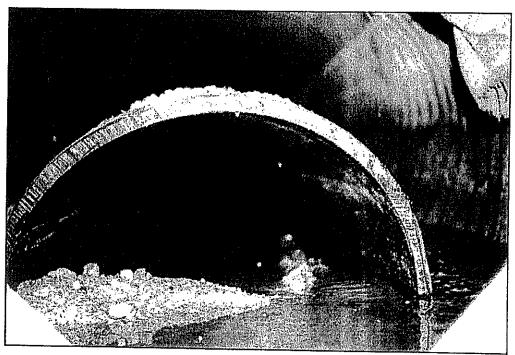


Plate 1-4 End view of the partially sanded PVC pipe at the bottom of the sump; the culvert leading into the bottom of the waste rock pile surrounds the pipe.

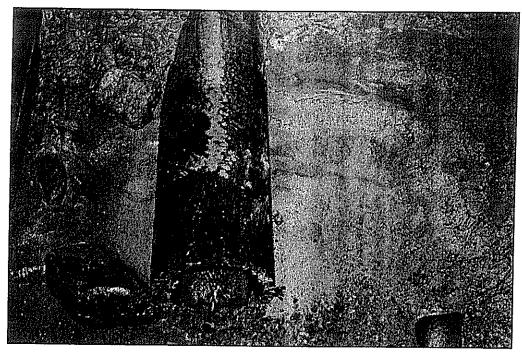


Plate 1-5 The outflow pipe at the bottom of the sump.

### 2. FIELD SAMPLING METHODS



## 2. Field Sampling Methods

### 2.1 Air Quality

CANTEST Laboratories of Vancouver, B.C. supplied Rescan with 1.0 and 5.0 L air quality sample bags (Tedlar bags). The 1.0 L bags were new while the 5.0 L bags had been used previously and certifiably cleaned by the CANTEST. The sample bags, along with a hand pump and sample tubing, were delivered to Rescan by courier on May 19, 2006. The air quality samples were treated as legal samples; the required chain of custody forms and seals were supplied by CANTEST.

The field sampling was carried out by volunteers from the Teck Cominco HazMat team who were each equipped with SCBA (Plate 2-1).

Rescan personnel provided the HazMat team members with detailed instructions on air sampling procedures. The air quality sample apparatus consisted of a 50 cm stainless steel tube (with an inside diameter of approximately 5 mm) attached to Teflon tubing and a rubber bulb with a one-way valve (hand pump). Sample bags were labelled by Rescan personnel and given to the HazMat team members. The 1.0 and 5.0 L sample bags were filled to approximately 80% capacity according to sample protocol. The Tedlar bags were then handed back to Rescan personnel, who had witnessed the sampling from a distance. The sample bags were sealed and signed according to legal sampling protocol. The air samples were placed in 34 L plastic coolers, which were legally sealed, and the appropriate chain of custody forms were filled out.

Air quality samples were collected at five locations in the following order:

- 1. Near the ceiling of the shed: The door to the shed was opened just enough to allow the HazMat team members to fit the stainless steel sample tube over the top of the door to collect an air sample near the ceiling of the shed (Plate 2-1).
- 2. **Inside the pipe draining into the sump:** HazMat team members climbed into the sump wearing SCBA and collected the air sample.
- 3. At grade inside the sump.
- 4. At the floor level inside the shed.
- 5. At the Sullivan Mine 3900 level portal sump shed.

One sample from each sampling location was provided to Teck Cominco for independent analysis.

In addition to air quality samples, the HazMat team measured the velocity of the air flowing from the pipe to the sump. On-site measurements of air temperature were made using a YSI 30 temperature sensor.



Plate 2-1 Teck Cominco HazMat team members collecting air quality samples from the shed at the Sullivan Mine (8:00 AM, May 20, 2006)

### 2.2 Water Quality

On-site measurements of temperature, conductivity, pH and dissolved oxygen of the seepage water in the sump were completed following the air quality sampling. The measurements were taken by Teck Cominco's HazMat team members using the following calibrated, digital, handheld instruments:

- a HANNA Instruments HI 9143 dissolved oxygen meter was used to measure dissolved oxygen and temperature;
- a YSI 30 conductivity and temperature sensor was used to measure conductivity and temperature; and
- an OAKTON waterproof pH tester was used to measure pH.

Detailed instructions on how to operate the instruments were provided by Rescan personnel.

Water quality sample bottles were supplied by ALS Environmental of Vancouver, B.C. The water quality samples were treated as legal samples and the required chain of custody forms and seals were provided by ALS Environmental.

Teck Cominco's HazMat team collected the samples as instructed by Rescan personnel (Plate 2-2). Samples were labelled and given to the HazMat team member who proceeded to

collect the water quality sample. One duplicate sample was collected from the seepage water draining from the culvert inside the sump between 8:20 and 8:25 AM on May 20, 2006.



Plate 2-2 Teck Cominco HazMat team members collecting water samples from the shed at the Sullivan Mine on May 20, 2006

After sampling had been completed the samples were handed back to Rescan personnel who added the required preservative, sealed the samples and placed them in coolers containing ice packs. The water quality samples were transported with the air quality samples to Rescan's Vancouver laboratory on May 20, 2006 where they were placed in secure storage. On May 23, 2006, the samples were delivered by Rescan personnel to ALS Environmental Laboratories for analysis, after the required chain of custody forms were completed. A detailed list of parameters included in the analysis and analytical results are presented in the following section.

### 2.3 Microbial Activity

Sample bottles for microbial activity samples were supplied by IG MicroMed Environmental in Richmond, B.C. The samples were not treated as legal samples. One duplicate sample was collected from the seepage water draining from the culvert inside the sump. The samples were analyzed by direct microscopic count and by plate count.

The samples were collected by HazMat team members and transported along with the water quality and air quality samples to Rescan's Vancouver office on May 20, 2006. On May 23, 2006, the samples were transported to IG MicroMed's laboratory in Richmond, B.C.



## 3. Results

## 3.1 Air Quality

The results of the on-site measurements on May 20, 2006 at 8:15 AM were:

- ambient air temperature: 14.1° C;
- · air temperature inside the culvert: 8.9° C; and
- velocity of air flowing from the pipe to the sump: 90 ft/min (0.46 m/s).

HazMat team members reported that the velocity of the air flowing from the pipe on May 18, 2006, at 9:00 PM was 168 ft/min (0.85 m/s).

The results of the air quality analysis of the samples collected by Rescan on May 20, 2006, are summarized in Table 3-1. Analytical test methods and results are presented in detailed in the air quality analysis data report from CANTEST included in Appendix A.

Table 3-2 shows a summary of the air quality analysis results for the samples collected by Teck Cominco at 9:00 AM on May 18, 2006. A detailed data report from CANTEST is included in Appendix B. Table 3-2 shows a summary of the air quality results for samples collected by BC MEM on the night of May 18, 2006. The samples were analyzed by ALS in Edmonton; a data report is provided in Appendix C.

The analysis of the air samples collected by Rescan on May 20, 2006 showed that the air inside the pipe and near the bottom of the sump was almost completely depleted of oxygen, with concentrations of 2% (Table 3-1). The concentration of nitrogen was between 90 and 91%, which is 11 to 12% higher than normal ambient nitrogen concentrations. Carbon dioxide concentrations were measured at 7%, approximately 200 times higher than ambient concentrations. Nitrous oxide concentrations were relatively low, at 12.4 and 12.7 ppm. Total hydrocarbon concentrations were equal to methane concentrations at 10.8 and 8.1 ppm upstream in the culvert and at grade in the sump, respectively. Carbon monoxide, sulphur dioxide, hydrogen sulphide and volatile organic compound (VOC) concentrations were all below analytical detection limits.

The composition of the air samples collected by Rescan in the pipe and near the bottom of the pit were similar to the composition of samples collected by Teck Cominco and BC MEM on the morning and night of May 18, 2006 (Table 3-2 and 3-3): 2 to 4% oxygen, 90 to 92% nitrogen and approximately 6% carbon dioxide.

The first air quality samples, collected near the ceiling of the shed before the door had been fully opened (Plate 2-1), contained 12 to 13% oxygen, 83% nitrogen and 3% carbon dioxide. The last air quality sample, collected near the floor of the shed (samples 3A and 3B), contained near ambient concentrations of oxygen (20%) and nitrogen (78%), while carbon dioxide concentrations were approximately eight times higher than ambient concentrations. At that time

Table 3-1 Air Samples Collected by Rescan on May 20, 2006 at 8:00 AM: Summary of Analytical Results (CANTEST)

Sample ID		1A	2A	3A	3B	TOPA	TOPB	BAG1	BAG2	
Location	Units	Upstream in Pipe	Bottom of Sump	,			Along Ceiling of Shed		3900 Level Portal Pipe Discharge	
Oxygen (O <sub>2</sub> )	% (v/v)	2	2	19	20	13	12	20	20	20.946
Nitrogen (N <sub>2</sub> )	% (v/v)	91	90	79	78	83	83	78	79	78.084
Carbon dioxide (CO₂)	% (v/v)	7	7	0.257	0.284	3	3	0.0502	0.0532	0.035
Carbon monoxide (CO)	ppm (v/v)	<1	<1	<1	<1	<1	<1	<1	<1	2.000
Methane (CH₄)	ppm (v/v)	10.8	8.1	<2	<2	8.4	5.6	<2	<2	1.7
Nitrous Oxide (N₂O)	ppm (v/v)	12.4	12.7	0.98	0.95	6.2	6.7	0.4	0.4	<0.5
Total Hydrocarbon	ppm (v/v)	10.8	8.1	<2	<2	8.4	5.6	<2	<2	-0.5
Sulphur dioxide (SO₂)	ppm (v/v)	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	_
Hydrogen Sulphide (H₂S)	ppm (v/v)	<0.5	< 0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	_
Volatile Organic Compounds (VOCs) <sup>2</sup>	ppm (v/v)	All 39 species	included in th	e analysis	s were be	elow detec	tion limit in			_

indicates concentrations less than stated detection limit.

1: Source: CRC Handbook of Chemistry and Physics and NASA Earth Fact Sheet: http://nssdc.gsfc.nasa.gov/planetary/factsheet/earthfact.html.

2: See Appendix A for details.

Table 3-2
Air Samples Collected by Teck Cominco on May 18, 2006 at 9:00 AM:
Summary of Analytical Results (CANTEST)

Sample ID		S#1	S#2	S#3	S#4	S#5	
Location	Units	Bottom of Pit	Upstream in Pipe (at Bottom of Pit)	3900 Portal Sump	961/962	Gerry Soron	Typical Ambient Air <sup>1</sup>
Oxygen (O₂)	% (v/v)	4	3	20	20	20	20.946
Nitrogen (N₂)	% (v/v)	90	90	78	78	78	78.084
Carbon dioxide (CO₂)	% (v/v)	6	6	0.0611	0.0498	0.219	0.035
Carbon monoxide (CO)	ppm (v/v)	<1	<1	<1	<1	<1	-
Methane (CH₄)	ppm (v/v)	10.6	1680	2.2	<2	2.1	1.7
Nitrous Oxide (N₂O)	ppm (v/v)	11.5	11.6	0.4	0.4	1.3	<0.5
Total Hydrocarbon	ppm (v/v)	41.6	1720	23.7	15.9	9.2	-
Sulphur dioxide (SO₂)	ppm (v/v)	<0.1	<0.1	<0.1	<0.1	<0.1	_
Hydrogen Sulphide (H₂S)	ppm (v/v)	<0.5	<0.5	<0.5	<0.5	<0.5	-
Volatile Organic Compounds (VOCs) <sup>2</sup>	ppm (v/v)	All 39 spec	cies included in the and	alysis were belo			

1 . .

#### <u>Notes</u>

2: See Appendix B for details.

<sup>&</sup>lt;: indicates concentrations less than stated detection limit.

<sup>1:</sup> Source: CRC Handbook of Chemistry and Physics and NASA Earth Fact Sheet: http://nssdc.gsfc.nasa.gov/planetary/factsheet/earthfact.html.

Table 3-3 Air Samples Collected by Ministry of Energy and Mines on May 19, 2006 at 1:07 AM: Summary of Analytical Results (ALS)

Sample ID		1A/B	2A/B	3A/B	Clean Air 1/2	
Location	Units	Pipe Air	1 Foot Below Grade (in Pit)	5 Feet Above Grade (Inside Shed)	Fresh Air	Typical Ambient Air <sup>1</sup>
Oxygen (O <sub>2</sub> )	% (v/v)	2	2	18	21	20.946
Nitrogen (N <sub>2</sub> )	% (v/v)	92	91	80	78	78.084
Carbon dioxide (CO <sub>2</sub> )	% (v/v)	5.9	5.7	1	0.049	0.035
Carbon monoxide (CO)	ppm (v/v)	<1	<1	<1	<1	-
Methane (CH₄)	ppm (v/v)	n/a	n/a	n/a	n/a	1.7
Nitrous Oxide (N₂O)	ppm (v/v)	n/a	n/a	n/a	n/a	<0,5
Total Hydrocarbon	ppm (v/v)	n/a	n/a	n/a	n/a	
Sulphur dioxide (SO <sub>2</sub> )	ppm (v/v)	n/a	n/a	n/a	n/a	-
Hydrogen Sulphide (H₂S)	ppb (v/v)	<10	<10	<10	<10	-
Volatile Organic Compounds (VOCs) <sup>2</sup>	μg/m³	21	550	400	330	-

<sup>&</sup>lt;: Indicates concentrations less than stated detection limit.

<sup>1:</sup> Source: CRC Handbook of Chemistry and Physics and NASA Earth Fact Sheet: http://nssdc.gsfc.nasa.gov/planetary/factsheet/earthfact.html.

<sup>2:</sup> See Appendix C for details.

the door to the shed had been opened for approximately 30 minutes. A sample collected five feet above grade in the shed on the night of May 18, 2006 by BC MEM contained 18% oxygen, 80% nitrogen and 1% carbon dioxide.

The composition of the air quality samples collected by Rescan on May 20, 2006, from the sump of the 3900 level portal was nearly identical to the composition of ambient air with slightly elevated carbon dioxide concentrations. This result was comparable to a sample collected at the same location by Teck Cominco on the morning of May 18, 2006 (Table 3-2).

#### 3.2 Water Quality

The results of the on-site measurements of temperature, pH and dissolved oxygen of the waste rock seepage in the sump were:

temperature: 9.0 to 9.1°C

pH: 3.3

dissolved oxygen: 0.0 %

Results of the water quality analysis were supplied by ALS Environmental Laboratories on May 25, 2006. The results are summarized in Table 3-4. A copy of the original water quality data report is included in Appendix B.

Table 3-4
Sump Water: Analytical Results

Sample ID		1A	1B	Travel Blank
Date Sampled		5/20/2006	5/20/2006	n/a
Time Sampled	Units	8:20	8:25	n/a
Physical Tests				
Colour	CU	75.7	73.7	<5.0
Conductivity	uS/cm	16000	15900	<2.0
Total Dissolved Solids	mg/L	30800	34300	<10
Hardness	mg/L	8070	8030	<0.54
pН	рΗ	3.27	3.27	5.89
Total Suspended Solids	mg/L	21.1	44.4	<3.0
Turbidity	NTU	7.82	6.75	<0.10
Dissolved Anions				
Acidity (to pH 8.3)	mg/L	11300	11600	1.8
Alkalinity-Total	mg/L	<2.0	<2.0	<2.0
Bromide	mg/L	< 0.50	<0.50	<0.050
Chloride	mg/L	40.5	40.5	<0.50
Fluoride	mg/L	18.5	20.8	<0.020
Sulphate	mg/L	19500	18900	<0.50

(continued)

Table 3-4
Sump Water: Analytical Results (continued)

Sample ID		1A	1B	Travel Blank
Date Sampled		5/20/2006	5/20/2006	n/a
Time Sampled	Units	8:20	8:25	n/a
Nutrients	····			
Ammonia Nitrogen	mg/L	6.26	5.05	<0.0050
Total Kjeldahl Nitrogen	mg/L	6.05	6.24	<0.050
Nitrate Nitrogen	mg/L	31.4	31.2	<0.0050
Nitrite Nitrogen	mg/L	0.046	0.030	<0.0010
Total Nitrogen	mg/L	43.2	44.6	< 0.050
Total Phosphate	mg/L	0.0633	0.0673	<0.0020
<u>Cyanides</u>				
Total Cyanide	mg/L	0.0013	<0.0010	n/a
Total Metals				
Total Aluminum	mg/L	843	875	<0.0010
Total Antimony	mg/L	<0.10	<0.10	< 0.00010
Total Arsenic	mg/L	<0.10	<0.10	<0.00010
Total Barium	mg/L	<0.050	< 0.050	< 0.000050
Total Beryllium	mg/L	<0.50	<0.50	< 0.00050
Total Bismuth	mg/L	<0.50	<0.50	< 0.00050
Total Boron	mg/L	<10	<10	<0.010
Total Cadmium	mg/L	2.98	3.09	<0.000050
Total Calcium	mg/L	442	442	<0.050
Total Chromium	mg/L	<0.50	<0.50	< 0.00050
Total Cobalt	mg/L	2.22	2.30	<0.00010
Total Copper	mg/L	1.83	1.87	< 0.00010
Total Iron	mg/L	122	121	< 0.030
Total Lead	mg/L	0.292	0.286	<0.000050
Total Lithium	mg/L	<5.0	<5.0	< 0.0050
Total Magnesium	mg/L	1670	1670	<0.10
Total Manganese	mg/L	673	712	<0.000050
Total Mercury	mg/L	0.000268	0.000167	<0.000050
Total Molybdenum	mg/L	<0.050	< 0.050	<0.000050
Total Nickel	mg/L	4.25	4.44	<0.00050
Total Phosphorus	mg/L	<3.0	<3.0	<0.30
Total Potassium	mg/L	30	30	<2.0
Total Selenium	mg/L	<1.0	<1.0	<0.0010
Total Silicon	mg/L	33.6	33.5	< 0.050
Total Silver	mg/L	<0.010	<0.010	<0.000010
Total Sodium	mg/L	35	35	<2.0

(continued)

Table 3-4
Sump Water: Analytical Results (continued)

Sample ID		1A	1B	Travel Blank
Date Sampled		5/20/2006	5/20/2006	n/a
Time Sampled	Units	8:20	8:25	n/a
Total Metals (cont'd)			<b></b>	
Total Strontium	mg/L	1.39	1.44	<0.00010
Total Thallium	mg/L	<0.10	<0.10	<0.00010
Total Tin	mg/L	<0.10	<0.10	<0.00010
Total Titanium	mg/L	<0.10	<0.10	<0.010
Total Uranium	mg/L	0.171	0.177	<0.000010
Total Vanadium	mg/L	<1.0	<1.0	<0.0010
Total Zinc	mg/L	4140	4320	<0.0010
Dissolved Metals				
Diss. Aluminum	mg/L	838	848	n/a
Diss. Antimony	mg/L	<0.10	<0.10	n/a
Diss. Arsenic	mg/L	<0.10	<0.10	n/a
Diss. Barium	mg/L	<0.050	< 0.050	n/a
Diss. Beryllium	mg/L	<0.50	<0.50	n/a
Diss. Bismuth	mg/L	<0.50	<0.50	n/a
Diss. Boron	mg/L	<10	<10	n/a
Diss. Cadmium	mg/L	3.05	2.88	n/a
Diss. Calcium	mg/L	440	441	n/a
Diss. Chromium	mg/L	< 0.50	< 0.50	n/a
Diss. Cobalt	mg/L	2.23	2.27	n/a
Diss. Copper	mg/L	1.94	1.86	n/a
Diss. Iron	mg/L	120	121	n/a
Diss. Lead	mg/L	0.292	0.273	n/a
Diss. Lithium	mg/L	<5.0	<5.0	n/a
Diss. Magnesium	mg/L	1690	1680	n/a
Diss. Manganese	mg/L	676	688	n/a
Diss. Mercury	mg/L	0.000235	0.000205	n/a
Diss. Molybdenum	mg/L	< 0.050	<0.050	n/a
Diss. Nickel	mg/L	4.48	4.27	n/a
Diss. Phosphorus	mg/L	<3.0	<3.0	n/a
Diss. Potassium	mg/L	30	30	n/a
Diss. Selenium	mg/L	<1.0	<1.0	n/a
Diss. Silicon	mg/L	33.4	33.5	n/a
Diss. Silver	mg/L	<0.010	<0.010	n/a
Diss. Sodium	mg/L	35	35	n/a
Diss. Strontium	mg/L	1.41	1.39	n/a

(continued)

Table 3-4
Sump Water: Analytical Results (completed)

Sample ID		1A	1B	Travel Blank
Date Sampled		5/20/2006	5/20/2006	n/a
Time Sampled	Units	8:20	8:25	n/a
Dissolved Metals (cont'd)				
Diss. Thallium	mg/L	<0.10	<0.10	n/a
Diss. Tin	mg/L	<0.10	<0.10	n/a
Diss. Titanium	mg/L	<0.10	<0.10	n/a
Diss. Uranium	mg/L	0.176	0.173	n/a
Diss. Vanadium	mg/L	<1.0	<1.0	n/a
Diss. Zinc	mg/L	4190	4200	n/a
Organic Parameters		7		
Total Organic Carbon	mg/L	19.2	18.7	n/a

Notes:

#### 3.3 Microbial Activity

Table 3-5 shows the results of the microbial analysis of the sump water collected at the site of the Sullivan incident. Appendix C shows a copy of the original data report from IG MicroMed Environmental.

Table 3-5
Results of Microbial Activity Analysis

Sample ID	Date/Time Collected	Heteorotrophic Plate Count (cfu/ml)	Sulphate Reducers	Hydrocarbon Degraders – Diesel Fuel (cfu/ml)
1A	May 20, 2006, 8:10 AM	7	Not seen microscopically	<50
1B	May 20, 2006, 8:10 AM	9	Not seen microscopically	<50

Note:

cfu: colony forming units.

Unfortunately, an analytical test for the bacteria *Thiobacillus ferrooxidans* was not available from IG MicroMed. *Thiobacillus ferrooxidans* is recognized as being responsible for the oxidation of iron and inorganic sulphur compounds in mine tailings where these compounds are abundant (Mousavi et al., 2006).

<sup>&</sup>lt;: Less than the detection limit indicated.

#### 4. ANALYSIS AND DISCUSSION



## 4. Analysis and Discussion

#### 4.1 Causes of Incident

The air quality results demonstrated that it was the oxygen-depleted atmosphere in the shed constructed over the bottom drain pipe that was the cause of the fatal incident at the Sullivan Mine. The oxygen content inside the shed and in the bottom of the pit was 2%, while carbon dioxide levels were measured at 7% (Table 3-1). Exposure to an atmosphere that contains less than 6% oxygen leads to immediate loss of consciousness without warning signs and subsequent death by asphyxiation (EIGA, 2003).

The measured concentrations of carbon dioxide of 7% would not have been immediately fatal but would have caused drowsiness, heavy breathing, dizziness and narcotic effects, had levels of oxygen been normal. Hydrogen sulphide, carbon monoxide and methane concentrations were all far below detrimental exposure limits (Doss and Person, 2002).

Confined spaces are potentially hazardous and should be treated with caution. Manure pits are the most common scene of asphyxiation fatalities in confined spaces. Manure pits contain high levels of decomposing organic material, which consume oxygen and produce noxious gasses such as hydrogen sulphide, ammonia and carbon monoxide. Methane gas can also be produced as a result of anaerobic bacterial decomposition of organic matter. While methane has relatively low toxicity, it is an explosive hazard at higher concentrations and can displace oxygen (NIOSH, 1990).

The microbial tests showed negligible heterotrophic bacterial activity, which indicate very low levels of aerobic decomposition of organic material (see Table 3-3). However, slightly elevated concentrations of methane gas suggest some anaerobic microbial activity (methanogens). Hydrocarbon degrading bacteria or sulphate reducing bacteria were not detected, which indicate that the oxygen depletion was not caused by degradation of possible hydrocarbon buried with the waste rock.

The oxygen depletion of the air at the sampling station where the incident took place was most likely caused by geochemical reactions within the waste rock pile connected to the sampling station. The waste rock produced by the Sullivan Mine contains sulphides, such as pyrite and pyritite, which oxidize and produce ARD through a series of chemical reactions. The water draining from the waste rock pile is typical of ARD: low pH (3.3), low (or no) dissolved oxygen, high levels of dissolved mineral and metal species including sulphate, aluminium, calcium, cadmium, cobalt, iron, magnesium, manganese, nickel and zinc (see Table 3-4). The low levels of nitrate and ammonia contained in the water are possibly remnants of explosives residues.

The strategy for curbing ARD is to prevent oxygen from entering the waste rock pile, thus reducing the rate of sulphide oxidation. This is typically done by submerging sulphide-bearing waste rock underwater or by attempting to reduce the ingress of air by placing a topsoil cover on the waste rock pile. Recent reclamation efforts by Teck Cominco included the placement of a thick layer (1 to 1.5 m) of topsoil over the waste rock pile. The topsoil cover extended past a

French drain along the base of the waste rock pile in an effort to seal off possible conduits from the drain to the waste rock pile.

Although a topsoil cover reduces oxygen transfer to a waste rock pile, it does not create a perfect seal. Air can diffuse through the soil and flow through cracks and openings in the cover, like culverts and drainage ditches. As oxygen enters the waste rock pile, pyrite or pyritite will oxidize through the chemical reactions described below (Stumm and Morgan, 1995):

Oxidation of pyrite:

$$FeS_2 + 3.5 O_2 + H_2O \rightarrow Fe^{2+} + 2 SO_4^{2-} + 2 H^+$$
 (1)

The resulting Fe<sup>2+</sup> in reaction (1) can be oxidized to Fe<sup>2+</sup> by:

$$2 \text{ Fe}^{2+} + 0.5 \text{ O}_2 + 2 \text{ H}^+ \rightarrow 2 \text{ Fe}^{3+} + \text{H}_2\text{O}$$
 (2)

The conversion of Fe(II) to Fe(III) can be greatly accelerated by the bacteria species *Thiobacillus* ferroxidans. These bacteria and several other species thought to be involved in pyrite oxidation are widespread in the environment. T. ferroxidans has been shown to increase the iron conversion reaction rate by a factor of hundreds to as much as one million times (Nordstrom and Southam, 1997; Bachelar-Nicolau and Johnson, 1998).

Combining reactions (1) and (2) gives:

$$2 \text{ FeS}_2 + 7.5 \text{ O}_2 + \text{H}_2\text{O} \rightarrow 2 \text{ Fe}^{3+} + 4 \text{ SO}_4^{2-} + 2 \text{ H}^+$$
 (3)

In a solution with pH>3, Fe(III) can precipitate from solution to produce additional acid (Janzen et al., 2000):

$$Fe^{3+} + 3 H_2O \rightarrow Fe(OH)_3 + 3 H^+$$
 (4)

Alternatively, Fe(III) can be consumed through further oxidation of sulphide minerals in acidic water by (Lee et al., 2003):

$$FeS_2 + 14 Fe^{3+} + 8 H_2O \rightarrow 15 Fe^{3+} + 2 SO_4^{2-} + 16 H^+$$
 (5)

Lee et al. (2003) showed that the acid generated by pyrite oxidation can dissolve available carbonates commonly found in waste rock to produce CO<sub>2</sub> gas by the following neutralizing reaction:

$$CO_3^{2-} + 2 H^+ \rightarrow H_2O + CO_2$$
 (6)

Combining reactions (1) and (6) yields:

$$FeS_2 + 3.5 O_2 + CO_3^{2-} \rightarrow Fe^{2+} + 2 SO_4^{2-} + CO_2$$
 (7)

In addition, combining reactions (3) and (6) yields Eq. (10):

$$2 \text{ FeS}_2 + 7.5 \text{ O}_2 + \text{CaCO}_3 \rightarrow 2 \text{ Fe}^{3+} + \text{Ca}^{2+} + 4 \text{ SO}_4^{2-} + \text{CO}_2$$
 (8)

According to Lee *et al* (2003), consumption of 1 mol of  $O_2$  by pyrite oxidation with carbonate neutralization may produce between 0.1 and 0.5 mol of  $CO_2$  and between 0.5 and 0.6 mol of sulphate. Reactions associated with microbial degradation of organic material produced 0.5 to 0.7 mol of  $CO_2$  for 1 mol of oxygen consumed. The composition of the air leaving the waste

rock pile at the site of the Sullivan incident was 2% O<sub>2</sub>, 7% CO<sub>2</sub> and 90% N<sub>2</sub>. Thus, approximately 0.4 mol CO<sub>2</sub> were produced for every 1 mol of O<sub>2</sub> consumed.

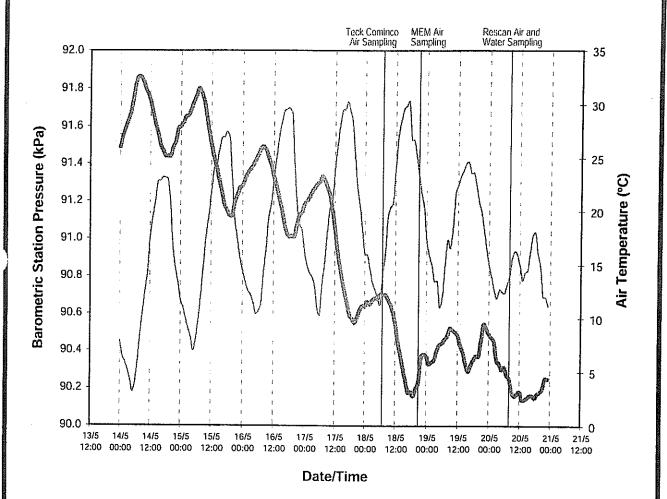
There are several gas transfer mechanism involved in the exchange of gasses between a waste rock pile and the surrounding environment including diffusion, convection and advection:

- As air enters the waste rock pile and sulphide is oxidized, an oxygen concentration gradient is established between the waste rock pile interior and the ambient air. This gradient provides a driving force for diffusion.
- The sulphide oxidation reaction is exothermic (heat-producing). Therefore, if the sulphide oxidation rate is sufficiently high, the temperature of the air inside the waste rock will increase and air may rise (convection) through the waste rock pile (Sraceka et al., 2004).
- At moderate or low sulphide oxidation rates, the production of heat within the waste rock pile may not be sufficient for convection to take place. As oxygen is replaced by heavier carbon dioxide, the density of the air increases and the air will tend to sink (gravity driven advection). If the temperature inside the waste rock pile is lower than the ambient air, the cooler air inside the waste rock pile will be denser than ambient air and will flow towards openings down-gradient. In addition, atmospheric pressure gradients may drive advection; blowing winds or rising and falling barometric pressure may increase gas transfer between the waste rock pile and the ambient air.

The mechanism that caused the oxygen-depleted air to flow from the waste rock pile to the sampling shed at the Sullivan Mine was likely density and temperature/pressure driven advection. The ambient temperature was measured at 14°C, while the air flowing from the waste rock pile was 8.9°C. The sulphide oxidation reactions would have depleted oxygen from air entering the waste rock pile, while the neutralization reactions would have added carbon dioxide to the air. Inside the pile, the oxygen-depleted air would have cooled and flowed along channels in the waste rock down-gradient toward openings such as the pipe located below the sampling shed.

Records of barometric pressure and air temperature are not available for Kimberly for May 2006. However, barometric pressure measured and recorded hourly at Cranbrook Airport, located approximately 30 km southeast of Kimberly was available. Figure 4-1 shows hourly barometric pressure and air temperature at Cranbrook Airport between May 14 and 21, 2006.

From May 14 to May 19, there were distinct diurnal fluctuations in temperature and barometric pressure, with an overall trend of decreasing barometric pressure and increasing temperature. Starting at approximately 8:00 AM, the air temperature would start to increase, causing the barometric pressure to drop. This trend would last until sunset – between 7:00 PM and 8:00 PM – when the trend would slowly reverse (decreasing air temperature and increasing pressure). This hypothesis is supported by the airflow measurements taken in the pipe from the rock pile. The flows in the pipe in the evening were approximately double the flows in the morning. The ambient temperature during the daytime hours was generally well above the temperature of the air flowing from the waste rock pile (9°C).



Hourly Temperature and Barometric Pressure

BRITISH at Cranbrook Airport, May 14 to 21, 2006

COLUMBIA

Temperature

Station Pressure (kPa)



#### 4.2 Prevention of Future Incidents

While identification of a confined space often is obvious with 20/20 hindsight, it is very unlikely that workers of any profession who have not completed confined space training and who are not regularly exposed to confined spaces would have recognized the danger. As environmental consultants, Rescan scientists and engineers collect thousands of water samples every year, mostly from groundwater wells and in open lakes, creeks and rivers, but also from open waste rock and tailings impoundments, culverts and ditches. Rescan staff is therefore included in the category of workers that would have been very unlikely to recognize the fatal danger that was present at the Sullivan site.

The widely published accounts of the Sullivan tragedy have helped to increase awareness of the very real hazards associated with working in and around confined spaces. Future efforts should be focused on strengthening awareness and enforcement of section 3.4 of the *Health, Safety and Reclamation Code for Mines in British Columbia* (BC MEM, 2003). For example, confined space training could be mandatory for any employee or contractor working at active or decommissioned mine sites. The existing regulations provide clear guidelines for appropriate working procedures.

However, identification of confined spaces may not be sufficient to prevent future incidents. Whether or not it flows into a confined space, any culvert or large pipe with an appreciable outflow of oxygen-depleted air constitutes a hazard. As has been stated, a second breath of air with less than 6% oxygen will cause immediate loss of consciousness and death (EIGA, 2003). An unwitting person looking into a pipe, culvert or other vent that is discharging oxygen-depleted air could succumb to asphyxiation in a few seconds. Therefore, special attention must be paid to any openings that allow for air transfer between a covered reactive waste rock pile and the ambient air. A safe perimeter around the openings should be fenced off and warnings should be clearly posted. Established procedures for working in and around confined spaces should be followed when accessing such sites.

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# APPENDIX A CANTEST AIR QUALITY ANALYSIS REPORT (RESCAN MAY 20, 2006)



#### Analysis Report

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Burgaby, B.C.

V5G 1K5

Analytical Services

REPORT ON:

Amended Report - Analysis of Gas Samples

REPORTED TO:

Rescan Environmental Services Ltd.

6th Fir-1111 West Hastings St

Vancouver, BC

V6E 2J3

Att'n: 4

CHAIN OF CUSTODY:

PROJECT NAME: PROJECT NUMBER:

196837 MOEM 246-0

**NUMBER OF SAMPLES: 8** 

REPORT DATE: May 26, 2006

DATE SUBMITTED: May 23, 2006

GROUP NUMBER: 70523009

**SAMPLE TYPE: Gas** 

NOTE: Results contained in this report refer only to the testing of samples as submitted. Other information is available on request.

#### **TEST METHODS:**

Carbon Monoxide in Gas - analysis was performed using procedures based on WCB Method 4200 using gas romatography with methanization and flame ionization detection.

Carbon Dioxide in Gas - analysis was performed using procedures based on WCB Method 4100 using gas chromatography with methanization and flame ionization detection.

Carbon Dioxide in Gas - analysis was performed using a gas chromatograph with thermal conductivity detection.

Methane in Gas - analysis was performed using gas chromatography with flame ionization detection.

Nitrogen in Gas - analysis was performed using a gas chromatograph with thermal conductivity detection.

Nitrous Oxide in Gas and Air - analysis was performed using procedures based on WCB Method 1101, involving analysis using gas chromatography with electron capture detection.

Oxygen in Gas - analysis was performed using a gas chromatograph with thermal conductivity detection.

Hydrogen sulphide in Air - analysis was based on WCB Method 0810. Hydrogen sulphide concentration is determined by drawing a known volume of air through a long-term colour indicating tube. The length of colour change in the tube is used to determine the concentration.

(Continued)

Rescan Environmental Services Ltd.

REPORT DATE:

May 26, 2006

GROUP NUMBER: 70523009

Sulphur Dioxide in Air - analysis was based on WCB Method 1360. Sulphur dioxide concentration was determined by drawing a known volume of air through a long-term detector tube. The length of colour change in the tube is used to determine the sulphur dioxide concentration.

Total Hydrocarbons in Gas - analysis was performed using procedures based on WCB Method 4700 using gas chromatography with flame ionization detection.

Volatile Organic Compounds in Gaseous Samples - analysis was performed using a modification of U.S. EPA Methods 624/8240/8260, involving injection of a gas sample into a Purge and Trap apparatus and analysis using GC/MS.

#### **COMMENTS:**

Amended Report: This report supersedes the electronic report issued on May 24, 2006. Samples 1B and 2B is omitted as requested by the client. No other chromatographable organic compounds were detected and identified in the submitted air samples.

#### **TEST RESULTS:**

(See following pages)

Rescan Environmental Services Ltd.

REPORT DATE:

May 26, 2006

GROUP NUMBER: 70523009

#### **Total Hydrocarbons in Gas**

CLIENT SAMPLE IDENTIFICATION:	SAMPLE DATE	CANTEST ID	Total Hydrocarbons
1A	May 20/06	605230026	10.8
2A	May 20/06		8.1
3A	May 20/06	605230033	< 2
3B	May 20/06	605230035	< 2
BAG1	May 20/06	605230036	
BAG2	May 20/06		< 2
TOPA	May 20/06	605230038	8.4
ТОРВ	May 20/06		5.6
DETECTION LIMIT UNITS			2 mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

Rescan Environmental Services Ltd.

**∂EPORT DATE:** 

May 26, 2006

GROUP NUMBER: 70523009

#### Compounds Determined using Detector Tubes in Gas

CLIENT SAMPLE IDENTIFICATION:	SAMPLE DATE	CANTEST ID	Sulphur Dioxide SO2	Hydrogen sulphide H2S
1A 2A 3B BAG1 BAG2 TOPA TOPB	May 20/06 May 20/06 May 20/06	605230038		V
DETECTION LIMIT	444		0,1 mL/cu. m	0.5 mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

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AEPORT DATE:

May 26, 2006

GROUP NUMBER: 70523009

#### Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	1A	2A	зА	3B	
DATE SAMPLED:	May 20/06	May 20/06	May 20/06	May 20/06	_ <u> </u>
CANTEST ID:	605230026	605230031	605230033	605230035	DETECTION
ANALYSIS DATE:	May 23/06	May 24/6	May 24/6	May 23/06	LIMIT
Benzene Bromodichloromethane Bromoform Bromomethane 2-Butanone Carbon Tetrachloride Chlorobenzene Chlorothane Chloroform Chloromethane Dibromochloromethane 1,2-Dibromoethane Dibromomethane Dichlorodifluoromethane 1,2-Dichlorobenzene 1,3-Dichlorobenzene 1,4-Dichlorothane 1,2-Dichloroethane 1,2-Dichloroethane 1,2-Dichloroethene 1,2-Dichloroethene 1,2-Dichloroethene 1,3-Dichloropropane 1,3-Dichloropropane 1,3-Dichloropropene 1,3-Dichloropropene 1,3-Dichloropropene 1,3-Dichloropropene 1,3-Dichloropropene 1,3-Dichloropropene 1,4-Methyl-2-pentanone 1,4-Methyl-2-pentanone 1,5-Methylene Chloride	<	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	0.01 0.01 0.02 0.08 0.5 0.01 0.01 0.04 0.03 0.04 0.01 0.01 0.02 0.02 0.02 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01

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REPORT DATE:

May 26, 2006

GROUP NUMBER: 70523009

#### Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	1A	<b>2</b> A	ЗА	3B	
DATE SAMPLED:	May 20/06	May 20/06	May 20/06	May 20/06	
CANTEST ID:	605230026	605230031	605230033	605230035	DETECTION
ANALYSIS DATE:	May 23/06	May 24/6	May 24/6	May 23/06	DETECTION LIMIT
Tetrachloroethene Toluene 1,1,1-Trichloroethane 1,1,2-Trichloroethane Trichloroethene Trichlorofluoromethane Vinyl Chloride Xylenes	V V V V V V	V	V V V V V V V V V V V V V V V V V V V	V V V V V V V	0.01 0.01 0.01 0.01 0.01 0.005 0.02 0.01

sults expressed as milligrams per cubic meter (mg/cu. m) = Less than detection limit

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**HEPORT DATE:** 

May 26, 2006

GROUP NUMBER: 70523009

#### Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	BAG1	BAG2	ТОРА	ТОРВ	
DATE SAMPLED:	May 20/06	May 20/06	May 20/06	May 20/06	
CANTEST ID:	605230036	605230037	605230038	605230039	DETECTION
ANALYSIS DATE:	May 23/06	May 23/06	May 23/06	May 23/06	LIMIT
Benzene Bromodichloromethane Bromoform	V V	< * * * * * * * * * * * * * * * * * * *	<b>V V</b>	V V	0.01 0.01
Bromomethane 2-Butanone	< <	< <	<	< < <	0.02 0.08 0.5
Carbon Tetrachloride Chlorobenzene	<	< <	< * * * * * * * * * * * * * * * * * * *	<b>V</b>	0.01 0.01
Chloroethane Chloroform	< "	< <	< <	< <	0.04 0.03
Chloromethane Dibromochloromethane	< <	< <	< <	<	0.04 0.01
1,2-Dibromoethane Dibromomethane Dichlorodifluoromethane	* <	<	<b>&lt;</b>	<b>&lt;</b>	0.01 0.02
1,2-Dichlorobenzene 1,3-Dichlorobenzene	< < < < < < < < < < < < < < < < < < <	* *	V / 1	· · · · · · · · · · · · · · · · · · ·	0.02 0.01
1,4-Dichlorobenzene 1,1-Dichloroethane	~ <			* <b>\</b> * \* \*.	0.01 ⊝0.01 ∌∄ 0.01
1,2-Dichloroethane 1,1-Dichloroethene	< <	<b>* *</b>		V W 1	0.04 0.04 0.01
cis-1,2-Dichloroethene trans-1,2-Dichloroethene	or:	. 14. <b>4</b>	<b>* * * * * * * * * *</b>	< < <	0.01 (c)
1,2-Dichloropropane cis-1,3-Dichloropropene	< <	< <	<b>&lt;</b> *	< <	0.01 0 } 0.01
rans-1,3-Dichloropropene Ethylbenzene 2-Hexanone	\$1 \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	< <	<b>V</b>	<	0,01 % 0.01 0.01
z-nexanone 4-Methyl-2-pentanone Methylene Chloride	< <	< <	<b>V</b> (1) (2) (1) (2) (2) (2) (2) (2) (2) (2) (2) (2) (2	< <	0.5 0.2
Styrene 1,1,2,2-Tetrachloroethane	<	< <		< <	0.6 0.01 0.02

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Rescan Environmental Services Ltd.

PORT DATE:

May 26, 2006

GROUP NUMBER: 70523009

#### Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	BAG1	BAG2	TOPA	ТОРВ	
DATE SAMPLED:	May 20/06	May 20/06	May 20/06	May 20/06	
CANTEST ID:	605230036	605230037	605230038	605230039	DETECTION
ANALYSIS DATE:	May 23/06	May 23/06	May 23/06	May 23/06	DETECTION
Tetrachloroethene Toluene 1,1,1-Trichloroethane 1,1,2-Trichloroethane Trichloroethene Trichlorofluoromethane Vinyl Chloride Xylenes	V V V V V V V V V V V V V V V V V V V	V V V V V V V V V V V V V V V V V V V	V V V V V V V V V V V V V V V V V V V	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	0.01 0.01 0.01 0.01 0.01 0.005 0.02 0.01

ults expressed as milligrams per cubic meter (mg/cu. m) . Less than detection limit  $\,$ 

Rescan Environmental Services Ltd.

REPORT DATE:

May 26, 2006

GROUP NUMBER: 70523009

#### Analysis of Components in Gas

CLIENT SAMPLE IDENTIFICATION:	1A	2A	зА	3B		·
DATE SAMPLED:	May 20/06	May 20/06	May 20/06	May 20/06	DETECTION	UNITS
CANTEST ID:	605230026	605230031	605230033		LIMIT	UNITS
Methane Carbon Monoxide Carbon Dioxide Carbon Dioxide Carbon Dioxide Oxygen Nitrogen Nitrous Oxide	7 7 2	2	2570 - 19	- 20,	20 1 1 1	mL/cu. m mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

% by vol. = percent by volume

Rescan Environmental Services Ltd.

.EPORT DATE:

May 26, 2006

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#### Analysis of Components in Gas

CLIENT SAMPLE IDENTIFICATION:	BAG1	BAG2	ТОРА	ТОРВ		
DATE SAMPLED:	May 20/06	May 20/06	May 20/06	May 20/06	DETECTION	LIMITO
CANTEST ID:	605230036	605230037	605230038	605230039	LIMIT	UNITS
Methane Carbon Monoxide Carbon Dioxide Carbon Dioxide Oxygen Nitrogen Nitrous Oxide	- 20 78	-	8.4 < 3 13 83 6.2	< 3 12 83	20 1 1	% by vol.

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

% by vol. = percent by volume

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Revision:00

Revision Date: April 19, 2005 Document Control Number: 60-F-028B

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APPENDIX B CANTEST AIR QUALITY ANALYSIS REPORT (TECK COMINCO, MAY 18, 2006)



#### **Analysis Report**



REPORT ON:

Analysis of Gas Samples

REPORTED TO:

Teck Cominco Ltd.

Kimberley Operations

Bag 2000 Kimberley, BC V1A 3E1

Att'n: Mr. Bruce Donald

CHAIN OF CUSTODY:

197319

NUMBER OF SAMPLES: 5

REPORT DATE: May 20, 2006

DATE SUBMITTED: May 20, 2006

**GROUP NUMBER: 70520001** 

SAMPLE TYPE: Gas

NOTE: Results contained in this report refer only to the testing of samples as submitted. Other information is available on request.

**TEST METHODS:** 

Carbon Monoxide in Gas - analysis was performed using procedures based on WCB Method 4200 using gas chromatography with methanization and flame ionization detection.

Carbon Dioxide in Gas - analysis was performed using procedures based on WCB Method 4100 using gas chromatography with methanization and flame ionization detection.

Carbon Dioxide in Gas - analysis was performed using a gas chromatograph with thermal conductivity detection.

Halogenated Hydrocarbons and Anaesthetic Agents - analysis was performed using procedures described in CSA Z180.1-00. Analysis was performed using gas chromatography with electron capture detection. Target compounds are "Freons", chlorinated-fluorinated ethane and methane compounds, and anaesthetic agents. CANTEST Method reference 67-A-007.

Methane in Gas - analysis was performed using gas chromatography with flame ionization detection.

Nitrogen in Gas - analysis was performed using a gas chromatograph with thermal conductivity detection.

Nitrous Oxide in Gas and Air - analysis was performed using procedures based on WCB Method 1101, involving analysis using gas chromatography with electron capture detection.

Oxygen in Gas - analysis was performed using a gas chromatograph with thermal conductivity detection.

(Continued)

CANTEST LTD.

CANTEST LTD.

Professional Analytical Services

4606 Canada Way Burnaby, B.C. V5G 1K5

FAX: 604 731 2386

TEL: 604 734 7276

1 800 665 8566

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001



Hydrogen sulphide in Air - analysis was based on WCB Method 0810. Hydrogen sulphide concentration is determined by drawing a known volume of air through a long-term colour indicating tube. The length of colour change in the tube is used to determine the concentration.

Sulphur Dioxide in Air - analysis was based on WCB Method 1360. Sulphur dioxide concentration was determined by drawing a known volume of air through a long-term detector tube. The length of colour change in the tube is used to determine the sulphur dioxide concentration.

Total Hydrocarbons in Gas - analysis was performed using procedures based on WCB Method 4700 using gas chromatography with flame ionization detection.

Volatile Organic Compounds in Gaseous Samples - analysis was performed using a modification of U.S. EPA Methods 624/8240/8260, involving injection of a gas sample into a Purge and Trap apparatus and analysis using GC/MS.

#### COMMENTS:

No other chromatographable organic compounds were detected and identified in the submitted air samples.

#### **TEST RESULTS:**

(See following pages)

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

CANTEST

#### Total Hydrocarbons in Gas

CLIENT SAMPLE IDENTIFICATION;	SAMPLE DATE	CANTEST ID	Total Hydrocarbon
S#1 Loc'n 1 Bottom of Pit	May 18/06	605200001	41.6
S#2 Loc'n 1 Upstream Pipe	May 18/06	605200002	1720
S#3 Loc'n 2 3900		605200003	23.7
S#4 Loc'n 3 961/962		605200004	15.9
S#5 Loc'n 4 Gerry Soron		605200005	9.2
DETECTION LIMIT			2
UNITS			mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v)

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

## CANTEST

#### Compounds Determined using Detector Tubes in Gas

CLIENT SAMPLE IDENTIFICATION:	SAMPLE DATE	CANTEST ID	Sulphur Dioxide SO2	Hydrogen sulphide H2S
S#1 Loc'n 1 Bottom of Pit	May 18/06	605200001	<	<
S#2 Loc'n 1 Upstream Pipe	May 18/06	605200002	<	<
S#3 Loc'n 2 3900	May 18/06	605200003	<	<
S#4 Loc'n 3 961/962	May 18/06	605200004	<	<
S#5 Loc'n 4 Gerry Soron	May 18/06	605200005	<	<
DETECTION LIMIT UNITS			0.1 mL/cu. m	0.5 mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

CANTEST

#### Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	S#1 Loc'n 1 Bottom	S#2 Loc'n 1 Upstream	S#3 Loc'n 2 3900	S#4 Loc'n 3 961/962	
	of Pit	Pipe			
DATE SAMPLED:	May 18/06	May 18/06	May 18/06	May 18/06	
CANTEST ID:	605200001	605200002	605200003	605200004	DETECTION
ANALYSIS DATE:	May 20/06	May 20/06	May 20/06	May 20/06	LIMIT
Benzene		<	<	<	0.01
Bromodichloromethane	<	<	<	<	0.01
Bromoform	<	<	<	<	0.02
Bromomethane	<	<	<	<	0.08
2-Butanone	<	<	<	<	0.5
Carbon Tetrachloride	<	<	<	<	0.01
Chlorobenzene	<	<	<	<	0.01
Chloroethane	<	<	<	<	0.04
Chloroform	<	<	<	<	0.03
Chloromethane	<	<	<	<	0.04
Dibromochioromethane	<	<	<	<	0.01
1,2-Dibromoethane	<	<	<	<	0.01
Dibromomethane	<	<	<	<	0.02
Dichlorodifluoromethane	<	<	<	<	0.02
1,2-Dichlorobenzene	<	<	<	<	0.01
1,3-Dichlorobenzene	<b> </b> <	<	<	<	0.01
1,4-Dichlorobenzene	<	<	<	<	0.01
1,1-Dichloroethane	<	<	<	<	0.01
1,2-Dichloroethane	<	<	<	<	0.04
1,1-Dichloroethene	<	<	<	<	0.01
cis-1,2-Dichloroethene	<	<	<	<	0.01
trans-1,2-Dichloroethene	<	<	<	<	0.01
1,2-Dichloropropane	<	<	<	<	0.01
cis-1,3-Dichloropropene	<	<	<	<	0.01
trans-1,3-Dichloropropeпе	<	<	<	<	0.01
Ethylbenzene	<	<	<	<	0.01
2-Hexanone	<	<	<	<	0.5
4-Methyl-2-pentanone	<	<	<	<	0.2
Methylene Chloride	<	<	<	<	0.6
Styrene	<	<	<	<	0.01
1,1,2,2-Tetrachloroethane	<	<	<	<	0.02

(Continued on next page)

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

# CANTEST

#### Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	S#1 Loc'n 1 Bottom of Pit	S#2 Loc'n 1 Upstream Pipe	S#3 Loc'n 2 3900	S#4 Loc'n 3 961/962	
DATE SAMPLED:	May 18/06	May 18/06	May 18/06	May 18/06	
CANTEST ID:	605200001	605200002	605200003	605200004	DETECTION
ANALYSIS DATE:	May 20/06	May 20/06	May 20/06	May 20/06	LIMIT
Tetrachloroethene	<	<	<	<	0.01
Toluene	<	<	<	<	0.01
1,1,1-Trichloroethane	<	<	<	<	0.01
1,1,2-Trichloroethane	<	<	<	<	0.01
Trichloroethene	<	<	<	<	10.0
Trichlorofluoromethane	<	<	<	<	0.005
Vinyl Chloride	<	<	<	<	0.02
Xylenes	<	<	<	<	0.01

Results expressed as milligrams per cubic meter (mg/cu. m) < = Less than detection limit

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

# CANTEST

# Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	S#5 Loc'n 4 Gerry Soron	The state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the state of the s
DATE SAMPLED:	May 18/06	
CANTEST ID:	605200005	DETECTION
ANALYSIS DATE:	May 20/06	LIMIT
Benzene	<	0.01
Bromodichloromethane	<	0.01
Bromoform	<	0.02
Bromomethane	<	0.08
2-Butanone	<	0.5
Carbon Tetrachloride	<	0.01
Chlorobenzene	<	0.01
Chloroethane	<	0.04
Chloroform	<	0.03
Chloromethane	<	0.04
Dibromochloromethane	<	0.01
1,2-Dibromoethane	<	0.01
Dibromomethane	<	0.02
Dichlorodifluoromethane	<	0.02
1,2-Dichlorobenzene	<	0.01
1,3-Dichlorobenzene	<	0.01
1,4-Dichlorobenzene	<	0.01
1,1-Dichloroethane	<	0.01
1,2-Dichloroethane	<	0.04
1,1-Dichloroethene	<	0.01
cis-1,2-Dichloroethene	<	0.01
trans-1,2-Dichloroethene	<	0.01
1,2-Dichloropropane	<	0.01
cis-1,3-Dichloropropene	<	0.01
trans-1,3-Dichloropropene	<	0.01
Ethylbenzene	<	0.01
2-Hexanone	<	0.5
4-Methyl-2-pentanone	<	0.2
Methylene Chloride	<	0.6
Styrene	<	0.01
1,1,2,2-Tetrachloroethane	<	0.02

(Continued on next page)

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001



## Volatile Organic Compounds in Gas

CLIENT SAMPLE IDENTIFICATION:	S#5 Loc'n 4 Gerry Soron	New York Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of t
DATE SAMPLED:	May 18/06	
CANTEST ID:	605200005	DETECTION
ANALYSIS DATE:	May 20/06	LIMIT
Tetrachioroethene	T <	0.01
Toluene	<	10.0
1,1,1-Trichloroethane	<	0.01
1,1,2-Trichloroethane	<	10.0
Trichloroethene	<	0.01
Trichlorofluoromethane	<	0.005
Vinyl Chloride	<	0.02
Xylenes	<	0.01

Results expressed as milligrams per cubic meter (mg/cu, m) < = Less than detection limit

Teck Cominco Ltd.

CANTEST

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

# Analysis of Components in Gas

CLIENT SAMPLE IDENTIFICATION:	S#1 Loc'n 1 Bottom of Pit	S#2 Loc'n 1 Upstream Pipe	S#3 Loc'n 2 3900	S#4 Loc'n 3 961/962		
DATE SAMPLED:	May 18/06	May 18/06	May 18/06	May 18/06	DETECTION	UNITS
CANTEST ID:	605200001	605200002	605200003	605200004		
Methane	10.6	1680	2.2	<	2	mL/cu. m
Carbon Monoxide	<	<	<	<	1	mL/cu. m
Carbon Dioxide	-	-	611	498	20	mL/cu. m
Carbon Dioxide	6	6	_	-	1	% by vol.
Oxygen	4	3	20	20	1	% by vol.
Nitrogen	90	90	78	78	[ 1	% by vol.
Nitrous Oxide	11.5	11.6	0.4	0.4	0.3	mL/cu. m
Trichlorotrifluoroethane	<	<	<	<	0.1	mL/cu. m
Chlorodifluoromethane	· ·	<	<	<	0.05	mL/cu. m
Dichlorodifluoromethane	<	<	<	<	0.1	mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

% by vol. = percent by volume

Teck Cominco Ltd.

REPORT DATE:

May 20, 2006

GROUP NUMBER: 70520001

### Analysis of Components in Gas

CLIENT SAMPLE IDENTIFICATION:	S#5 Lac'n 4 Gerry Saron		
DATE SAMPLED:	May 18/06	DETECTION	UNITS
CANTEST ID:	605200005	LIMIT	
Methane	2,1	2	mL/cu. m
Carbon Monoxide	<	1	mL/cu. m
Carbon Dioxide	2190	20	mL/cu. m
Oxygen	20	1	% by vol.
Nitrogen	78	1	% by vol.
Nitrous Oxide	1.3	0.3	mL/cu. m
Trichlorotrifluoroethane	<	0.1	mL/cu, m
Chlorodifluoromethane	<	0.05	mL/cu. m
Dichlorodifluoromethane	<	0.1	mL/cu. m

mL/cu. m = mL/cubic meter or ppm (v/v) < = Less than detection limit

% by vol. = percent by volume

CANTEST

# APPENDIX C ALS AIR QUALITY ANALYSIS REPORT (MINISTRY OF ENERGY AND MINES, MAY 18, 2006)



#### **Environmental Division**

	PRELIM	INARY RESULTS		
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Lab Work Order #:	L390464		Date Received:	19-MAY-06
Project P.O. #: Job Reference: Legal Site Desc: CofC Numbers:	PIPE AIR 222509			
Comments: Water	Samples not preserved or filtered.			
			•	•
AF	PROVED BY:			

THIS REPORT SHALL NOT BE REPRODUCED EXCEPT IN FULL WITHOUT THE WRITTEN AUTHORITY OF THE LABORATORY ANY REMAINING SAMPLES WILL BE DISPOSED OF AFTER 30 DAYS FOLLOWING ANALYSIS PLEASE CONTACT THE LAB IF YOU REQUIRE ADDITIONAL SAMPLE STORAGE TIME

ALS Canada Limited **Environmental Division** Part of the ALS Laboratory Group 9936-67 Avenue, Edmonton, AB T6E 0P5 Phone: +1 780 413 5227 Fax: +1 780 437 2311 www.alsglobal.com

A Compbell Brothers Limited Company



Sample Delails	s/Parameters	Result	Qualifier*	D.L.	Units	Extracted	Analyzed	Ву	Batch
L390464-2	1B ~ PIPE AIR	:				•			•
Sample By:	. on 19-MAY-06 @ 0	1:07							
Matrix:	AIR	<u> </u>							
YIGUIA.	7.11.1	1			ĺ		00 14414 00		R400919
	Carbon Monoxide	<b>,</b> <1		1	ppm		20-MAY-06	•	1
	Carbon Dioxide	59000		10	ppm		20-MAY-06		R400919
	Nitrogen	. 92		1	%		20-MAY-06		R400919
	Oxygen	2		1	%		120-MAY-06		R400919
000454.6	28 ~ 1' < GRADE (PIT)	ر د د د د د د د د د د د د د د د د د د د	1						i l
390464-4	on 19-MAY-06 @ C	1:07							1
Sample By:			ŧ		;				Ī
Matrix:	AIR		•		:		į		
	Carbon Monoxide	<1		1	ppm		20-MAY-06		R400919
	Carbon Dioxide	57000		10	ppm		20-MAY-06		R400919
	Nitrogen	91	ì	1	%		20-MAY-06		R400919
	<del>-</del>	2	ı	1	%		20-MAY-06		R400919
	Oxygen	1 4		, practical manufactions	Shown pulling helicological man				İ
L390464-6	3B ~ 5' > GRADE	24.07			i	•	1		
Sample By:	on 19-MAY-06 @ (	<i>)</i> <sub>i</sub> 1:07	;						4 **
Matrix:	AIR		į				]		
		-14	•	1	ppm	•	20-MAY-06		R400919
	Carbon Monoxide	<1		10	ppm	1	20-MAY-06		R400919
	Carbon Dioxide	10000			۱۱۱۰م %		20-MAY-08		R400919
	Nitrogen	80		1	% %	:	20-MAY-06		R400919
	Oxygen	18		1	70	and the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second of the second o	.20-14/-11-00		1
L390464-8	CLEAN AIR 2 ~FRESH AIR	i		ŧ	T.	4			i i
Sample By:	on 19-MAY-06 @	0 1:07				ţ			
Matrix:	AIR	1		,		<u> </u>			
		1					20-MAY-06		R400919
	Carbon Monoxide	<1		1	, ppm		20-MAY-06		R400919
	Carbon Dioxide	490		10	ppm		20-MAY-06	ı	R400919
	Nitrogen	78		1	%			;	
	Oxygen	21		1	%		20-MAY-06		R400919
L390464-9	SULIVAN MINE WATER SAMPLE				:				t
Sample By:	on 19-MAY-06 @	01:07			;	1		Į.	
Matrix	WATER	•						:	
	ed Metals	1		1				i	
				1	•	00 443 / 0	6 20-MAY-06	ł	R400964
	GC/MS Characterization of Extractables	<ul> <li>See Appendix A</li> </ul>	•						
	GC/MS Characterization of Volatiles	See Appendix B				,20-MAY-0	6 20-MAY-06	i .	, R400987
Routing	Water Analysis		<u> </u>						Antiquipment of the second extension
			albadalaav						i
	* Refer to Referenced Information for Q	lualmers (II any) and M	eriloocody		4		1		ļ
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# Reference Information

Sample Number Cli	ient ID	Qualifier	Description	
	JLIVAN MINE WA	ATER SAN SFU SRU	Sample fillered to remove particulate - Dissolved Sample Received Unpreserved - Dissolved meta	metals filtered before analysis
Methods Listed (if	applicable):			Analytical Method Reference(Based On)
LS Test Code	Matrix	Test Description	Preparation Method Reference(Based On)	
C-DIOXIDE-ED	Air	Compressed Breathir	ng Air	GC-TCD/FID
AIHA Accredited a	analysis in Gas Ch	nromatography Field of Te	esting	
-MONOXIDE-ED	Air	Carbon Monoxide		OSHA ID 210-GC-FID
AIHA Accredited a	analysis in Gas Cl	nromatography Field of Te	esting	
CL-IND-ED	Water	Chloride (CI)		APHA 4110 B-lon Chromatography
ETL-ROUTINE-ICP-I	ED Water	ICP metals and SO4 water	for routine	APHA 3120 B-ICP-OES
ONBALANCE-ED	Water	Ion Balance Calculati	on	APHA 1030E
MET1-DIS-LOW-ED	Water	Dissolved Trace Meta	als (Low	EPA 6020
MET2-DIS-ED	Water	Level) Dissolved Major Meta	als	EPA 200.7
N2-GC-ED	Air	Compressed Breathin	ng Air	GC-TCD/FID
AIHA Accredited	analysis in Gas C	hromatography Field of To	esting	
NH4-ED	Water	Ammonia-N		APHA4500NH3F Colorimetry
				APHA 4110 B Ion Chromatography
NO2-IND-ED	Water	Nîtrite-N		APHA 4110 B Ion Chromatography
NO3-IND-ED	Water	Nitrate-N		
O2-ED	Air	Percent Oxygen in A	ir	OSHA 6601-O2 METER
Not an AlHA or C	AEAL accredited	test service.		0)2040 0070
OPENSCAN-EXT-E	D Waler	GC/MS Characteriza Extractables	ation of	SW846 8270
OPENSCAN-VOC-E	ED Water	GC/MS Characteriza	ation of EPA 5021	EPA5021/8260
PH/EC/ALK-ED	Water	pH, Conductivity and Alkalinity	i Total	APHA 4500-H, 2510, 2320
SULPHIDE-ÉD	Water	Sulphide		APHA 4500 -S E-Auto-Colorimetry
		anning a specimen specimen in the specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen and specimen a	"* Laboratory Methods employed follow in generally based on nationally or internatio	n-house procedures, which are inally accepted methodologies.
Chain of Custody	numbers:			
222509				
	ers of the above t	est code(s) indicate the	laboratory that performed analytical analysis for th	hat test. Refer to the list below:
Laboratory Defini	tion Code La	aboratory Location	Laboratory Definition Code	Laboratory Location
ED	AL	S LABORATORY GROU MONTON, ALBERTA, C	JP - CANADA	

# Reference Information

Surr - A surrogate is an organic compound that is similar to the target analyte(s) in chemical composition and behavior but not normally detected in environmental samples. Prior to sample processing, samples are fortified with one or more surrogate compounds. The reported surrogate recovery value provides a measure of method efficiency. The Laboratory control limits are determined under column heading D.L.

mg/kg (units) - unit of concentration based on mass, parts per million mg/L (units) - unit of concentration based on volume, parts per million

< Less than

D.L. - Detection Limit

N/A - Result not available. Refer to qualifier code and definition for explanation

Test results reported relate only to the samples as received by the laboratory.

UNLESS OTHERWISE STATED, ALL SAMPLES WERE RECEIVED IN ACCEPTABLE CONDITION
UNLESS OTHERWISE STATED, SAMPLES ARE NOT CORRECTED FOR CLIENT FIELD BLANKS. Although test results are generated under strict QA/QC protocols, any unsigned test reports, faxes, or emails are considered preliminary.

ALS Laboratory Group has an extensive QA/QC program where all analytical data reported is analyzed using approved referenced procedures followed by checks and reviews by senior managers and quality assurance personnel. However, since the results are obtained from chemical measurements and thus cannot be guaranteed, ALS Laboratory Group assumes no liability for the use or interpretation of the results.

**SAMPLE DESCRIPTION:** 

Analyze four gasbag samples by Entech canister system - gas

chromatography/mass spectrometry to characterize all VOCs.

LAB SAMPLE #	CLIENT I.D.	RESULTS
L390464-1	1A ~ PIPE AIR	Please see below.
L390464-3	2A ~ 1' < GRADE (PIT)	Please see below.
L390464-5	3A ~ 5' > GRADE	Please see below.
L390464-7	CLEAN AIR 1 ~ FRESH AIR	Please see below.

#### METHODOLOGY:

A portion of the sample from the gasbag was withdrawn through the Entech Model 7032A Autosampler and cryofocused with the Entech 7100A Preconcentrator. The effluent was directed to the gas chromatograph/mass spectrometer (GC/MS) for separation, detection and identification of eluting compounds. The initial temperature was 35°C and programmed ramp to 220°C. Chromatograms are attached.

A pre-cleaned 6L Canister was spiked with 51 ppb (164 ug/m³) hexane. Quantitation of each compound detected in the sample was performed by comparison to the hexane response to provide a semi-quantitative estimate of amount of each compound. Ethylbenzene\_d10 was spiked on-line to serve as an internal standard.

Each compound detected was identified from the mass spectra and a degree of confidence value assigned based on the analyst's confidence in the identification. A legend of these values follows.

#### TABLE OF CONFIDENCE OF IDENTIFICATION

- NUMBER 1: Very confident, clean spectra, excellent match with library spectrum.
- **NUMBER 2:** Some small spectra differences with library match. A good comparison. Reference standard needed for positive identification.
- **NUMBER 3:** Possible compound or very similar to. Spectrum is definitely contaminated. Reference standard needed for positive identification.
- NUMBER 4: Evidence of possible structure and molecular weight. Compound not identified.
- **NUMBER 5:** Unable to identify.

LAB SAMPLE #

CLIENT I.D.

L390464-1

1A ~ PIPE AIR

RETENTIO TIME (min.	N COMPOUND IDENTIFICATION )	CONFIDENCE OF IDENTIFICATION	Air Conc. (ug/m³)
1.9-2.2	Air	*	•
5.0	Carbon dioxide*	1	-
5.2	Carbonyl sulfide	3	3
5.6	Isobutane	2	3
6.8	Acetonitrile*	2	1
8.4	Carbon disulfide*	1-2	2
9.4	C5 cycloalkane or C5 alkene	3	1
10.5	Hexene	2	1
18.8	C8 alkene or C9 alkane	· 3	1
19.7	Internal standard	-	•
20.2	C9 alkane or C6 alkanol / Ethylbenzene	2-3	2
26.1	Unknown ether or alkane	3	1
27.1	C10 or C11 alkane	2	1
		Detection Limit:	1
		Total VOC's:	21

<sup>\*</sup>Possible laboratory or instrument contaminant.

LAB SAMPLE #

CLIENT I.D. 2A ~ 1' < GRADE (PIT)

L390464-3 2A ~ 1

2000404-0	ZA ~ I < GRADE (FIL)		
	N COMPOUND IDENTIFICATION	CONFIDENCE OF	Air Conc.
TIME (min.	.)	IDENTIFICATION	(ug/m³)
1.9-2.2	Air	-	-
4.9-5.0	Carbon dioxide*	1	-
5.3	Carbonyl sulfide / Propane	3	33
5.6	Isobutane / Methanol	2-3	15
5.9	C4 alkene	1-2	8
6.0	Butane	1	16
6.7	Acetonitrile*	1-2	73
7.1	Acetone / C5 cycloalkane or C5 alkene	2-3	18
7.2	Trichloromonofluoromethane	1-2	1
7.3	Isopropyl alcohol	1-2	19
7.6	Pentane	1	28
7.7	C5 alkadiene or Ethylidene cyclopropane	2-3	1
8.0	Methylene chloride* / Dimethyl cyclopropane or C5 alkene	2	9
8.4	Carbon disulfide*	1	14

LAB SAMPLE #

CLIENT I.D. 2A ~ 1' < GRADE (PIT) L390464-3

RETENTION	N COMPOUND IDENTIFICATION	CONTINUENCE OF	A: O
TIME (min.		CONFIDENCE OF IDENTIFICATION	Air Conc.
8.6	Unknown cycloalkene	3-4	(ug/m³)
9.4	C5 alkene or C5 cycloaikane	3	4
9.5	C6 alkane	1 <b>-</b> 2	19
9.7	Butanal	2	2
10.0	C6 alkane	2	1
10.5	Hexane	1	14
10.7	Chloroform	1-2	2
11.9	1,1,1-Trichloroethane	1-2	1
12.8	Cyclohexane	2	3
13.0	C7 alkane	2	1
13.1	C6 alkene	2	2
13.4	C5 alkanal / C7 alkene	2-3	5
14.2	Hexene	2	3
15.2	Methyl Isobutyl Ketone / Methyl cyclohexane	2-3	4
15.6	C7 alkane	2	1
16.5	Toluene	1	4
16.7	3,4-Dihydro-2H-pyran	2-3	2
17.9	C8 alkane	2-3	1
18.2	Tetrachloroethylene	1	1
18.6	Hexamethyl cyclotrisiloxane	2	3
18.8	C9 alkane	1-2	20
19.7	Internal standard	<u>-</u>	-
20.0	C9 alkane	1-2	3
20.2	C9 alkane	1-2	34
20.5	Cyclohexanone	2-3	1
23.2	Phenol	. 1	2
24.2	Unidentified compounds	5	8
24.5	C9 or C10 alkane	2-3	1
24.9	C7 alkanol	2	3
25.1	C4 alkyl benzene	2	1
25.4	Limonene or similar terpene	1-2	8
25.7	Unknown alkene / alkane	. 3	1
26.0	C10 alkane	2-3	1
26.1	C11 alkane	2	21
26.2	C11 alkane	3	2
26.3	C11 alkane	2	11
26.6	Unknown cycloalkane	2-3	2
27.0	C11 alkane	2	5
27.1	C12 alkane	2	37
		<del></del>	01

LAB SAMPLE #

CLIENT I.D.

L390464-3

2A ~ 1' < GRADE (PIT)

RETENTIO	N COMPOUND IDENTIFICATION	CONFIDENCE OF IDENTIFICATION	Air Conc. (ug/m³)
27.2	C12 alkane	2	16
27.3	C12 alkane	2	9
27.7	C12 alkane	2-3	2
28.0	Unknown silyl compound	3	5
28.7	Naphthalene	1-2	1
28.8	Dodecane	2	1
30.3	C13 alkane	2	2
31.0	Unknown silyl compound / C13 alkane	3	2
	•	Detection Limit:	1
		Total VOC's	550

<sup>\*</sup>Possible laboratory or instrument contaminant.

LAB SAMPLE #

CLIENT I.D.

390464-5	3A ~ 5'	> GRADE

L390464-5	3A ~ 5 > GRADE		
	I COMPOUND IDENTIFICATION	CONFIDENCE OF	Air Conc.
TIME (min.)		IDENTIFICATION	(ug/m³)
1.9-2.2	Air	•	-
5.0	Carbon dioxide*	1	-
5.3	Carbonyl sulfide	3	4
5.7	Methanol	1-2	11
6.0	Butane	3	2
6.8	Acetonitrile*	1-2	8
7.0	Acetone	1-2	9
7.2	Trichloromonofluoromethane	1-2	1
7.3	1-Methoxy-2-propanone	2	4
7.6	Pentane	2	8
8.0	Methylene chloride*	1	10
8.4-8.6	Carbon disulfide*	1	210
9.5	C6 alkane	1-2	3
10.5	Hexane	2	5
16.4	Toluene	1	3
18.8	C9 alkane	1-2	3
19.7	Internal standard	-	-
20.2	C9 alkane / Ethylbenzene	2	9
24.2	Unknown alkyl benzene	3	2
25.4	Limonene or similar terpene	2	3
26.1	C11 alkane	2	5
			-

LAB SAMPLE #

1390464-5

CLIENT I.D. 3A ~ 5' > GRADE

RETENTION COMPOUND IDENTIFICATION TIME (min.)		CONFIDENCE OF IDENTIFICATION		
26.3	C11 alkane	2-3	2	
27.0	C11 alkane	2-3	2	
27.1	C12 alkane	2	10	
27.2	C12 alkane	2	7	
27.3	C12 alkane	2	4	
28.8	Dodecane	2	4	
20.0	Dodouile	Detection Limit:	1	
		Total VOC's:	400	

<sup>\*</sup>Possible laboratory or instrument contaminant.

LAB SAMPLE #

L390464-7	CLEAN AIR 1 ~ FRESH AIR		
RETENTION	COMPOUND IDENTIFICATION	CONFIDENCE OF	Air Conc.
TIME (min.)		IDENTIFICATION	(ug/m³)
1.9-2.2	Air	,	-
5.0	Carbon dioxide*	1	-
5.3	Carbonyl sulfide	. 2	16
5.7	Isobutane / Methanol	2-3	9
5.9	C4 alkene	2-3	2
6.5	Ethanol	2-3	1
6.8	Acetonitrile*	1-2	10
7.0	Acetone	1-2	9
7.2	Trichloromonofluoromethane	1-2	1
7.3	Isopropyl alcohol	1-2	12
7.6	Pentane	1	3
8.0	Methylene chloride*	1	7
8.4	Carbon disulfide*	1	67
9.4	C5 alkene or C5 cycloalkane	3	2
9.5	C6 alkane	1-2	11
10.5	Hexane	1	8
11.6	C6 alkene or C6 cycloalkane	2-3	1
12.8	Cyclohexane	2	2
16.5	Toluene	1	2
18.6	C8 alkane / Unknown siloxane	2-3	1
18.8	C9 alkane	1-2	15
19.7	Internal standard	-	-
20.0	C9 alkane	2	2

LAB SAMPLE #

1390464-7

CLIENT I.D. CLEAN AIR 1 ~ FRESH AIR

F390404-1	OLEAN AID 1 ~ FRESHAIR		
RETENTION	N COMPOUND IDENTIFICATION	CONFIDENCE OF	Air Conc.
TIME (min.)		IDENTIFICATION	(ug/m³)
20.2	C9 alkane	1-2	24
24.2	Unknown alkyl benzene	3	2
24.5	C9 or C10 alkane	2-3	2
24.9	Unknown alkene	3	2
25.4	Limonene or similar terpene	1-2	17
26.0	C10 alkane	2-3	1
26.1	C11 alkane	2	10
26.2	C11 alkane	3	1
26.3	C11 alkane	2-3	4
26.6	Unknown cycloalkane	3-4	1
27.0	C11 alkane	2	3
27.1	C12 alkane	2	17
27.2	C12 alkane	2	8
27.3	C12 alkane	2	4
28.0	Unknown silyl compound	3	1
28.7	Naphthalene / C12 alkane	3	1
28.8	C12 alkane	2-3	2
		Detection Limit: Total VOC's:	1 330

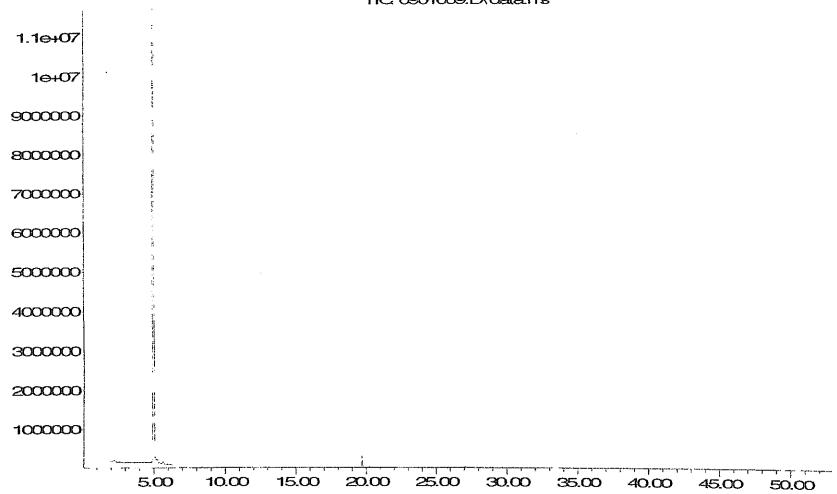
<sup>\*</sup>Possible laboratory or instrument contaminant.

L390464 Gasbag VOCs Analysis Page 8 of 11 DATE ANALYZED: May 19, 2006

# ALS LABORATORY GROUP ANALYSIS REPORT

L390464-1 Abundance

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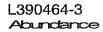


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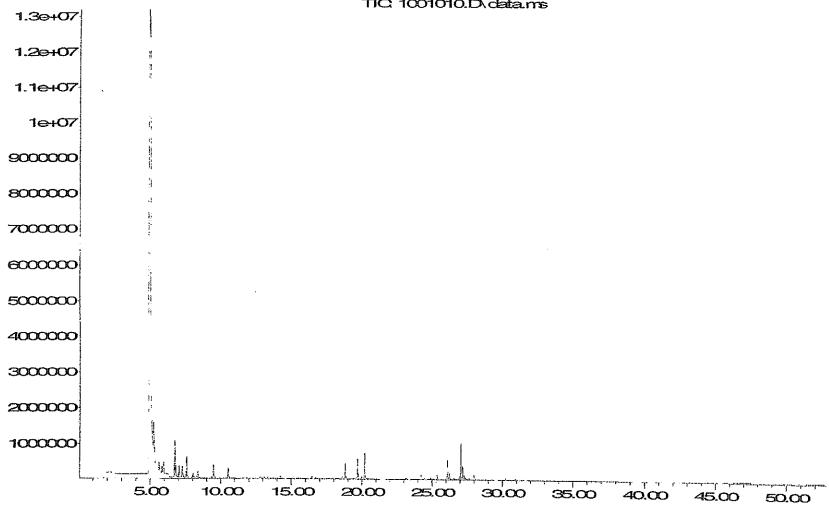
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Gasbag VOCs Analysis Page 9 of 11 DATE ANALYZED: May 19, 2006

# ALS LABORATORY GROUP ANALYSIS REPORT

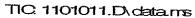


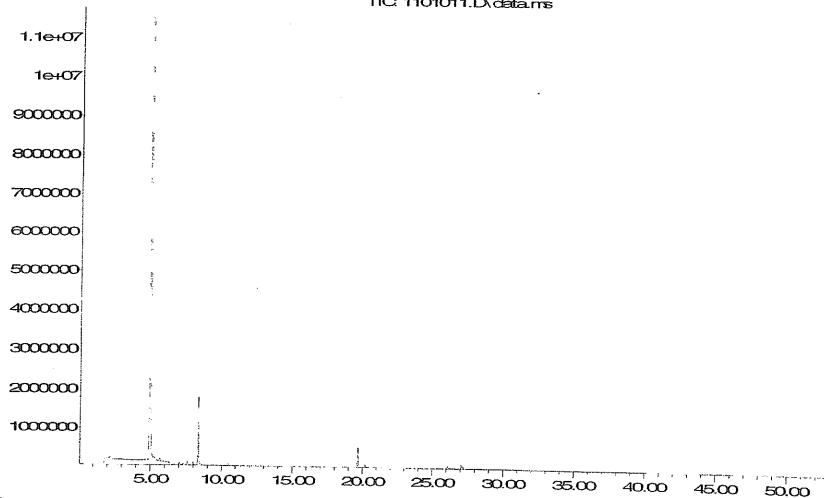




Time->

L390464-5 Abundance



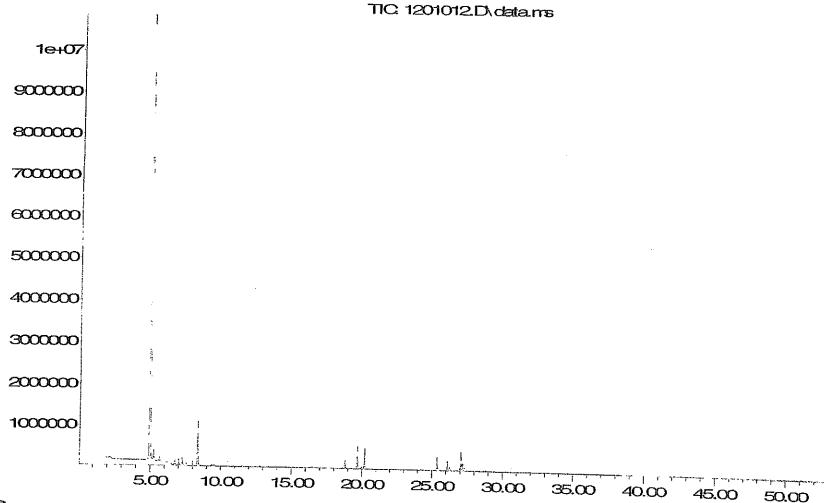


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L030464 Gasbag VOCs Analysis Page 11 of 11 DATE ANALYZED: May 19, 2006

# ALS LABORATORY GROUP ANALYSIS REPORT

L390464-7 Abundance



Time->

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L390464

Page 1 of 1 -

ANALYSIS: Hydrogen Sulfide

in Gasbags

DATE ANALYZED: 19-May-06

# ALS LABORATORY GROUP CHEMICAL ANALYSIS REPORT

ALS LAB SAMPLE ID: L390464-1

CLIENT SAMPLE ID: 1A - PIPE AIR

COMPOUND:	DETECTION LIMIT	UNITS	AIR CONCENTRATION	
			(ppb)	(ug/m³)
Hydrogen Sulfide	10	ppb	<10	<20

ALS LAB SAMPLE ID: L390464-3

CLIENT SAMPLE ID: 2A ~ 1' < GRADE (PIT)

COMPOUND:	DETECTION	UNITS	AIR CONCI	AIR CONCENTRATION	
	LIMIT		(ppb)	(ug/m³)	
Hydrogen Sulfide	10	ppb	<10	<20	

ALS LAB SAMPLE ID: L390464-5

CLIENT SAMPLE ID: 3A ~ 5' > GRADE

COMPOUND:	DETECTION LIMIT	UNITS	AIR CONCENTRATION	
			(ppb)	(ug/m³)
Hydrogen Sulfide	10	ppb	<10	<20

ALS LAB SAMPLE ID: L390464-7

CLIENT SAMPLE ID: CLEAN AIR 1 ~ FRESH AIR

COMPOUND:	DETECTION	UNITS	AIR CONC	NTRATION
	LIMIT		(ppb)	(ug/m³)
Hydrogen Sulfide	10	ppb	<10	<20

METHODOLOGY:

EPA TO15 - GC/MASS SPEC. (modified for gasbags)

# APPENDIX D ALS WATER QUALITY ANALYSIS REPORT (RESCAN, MAY 20, 2006)



# ALS Environmental



# **CERTIFICATE OF ANALYSIS**

Date:

May 25, 2006

ALS File No.

X6065

Report On:

MEM 246-0 Water Analysis

Report To:

**Rescan Environmental Services** 

Sixth Floor

1111 West Hastings Street

Vancouver, BC V6E 2J3

Attention:

Received:

May 23, 2006

**ALS ENVIRONMENTAL** 

per:

#### REMARKS



These samples were handled following protocols dictated by ALS' Level 1 Legal Handling procedures. Samples arrived in a sealed cooler, and were tracked after arrival on the ALS Legal Chain of Custody. All samples were kept in locked storage when not in use. Copies of the ALS Legal COC are attached.

The detection limits for some total and dissolved metals were increased for the samples reported due to interferences encountered during analysis.

For some of the submitted water samples, the measured concentration of specific dissolved parameters is greater than the corresponding total parameters concentration. The explanation for these findings is one or a combination of the following:

- laboratory method variability;
- field sampling method variability;
- bias introduced during general handling, storage, transportation and/or analysis of the sample;
- field sample grab bias where separate grab samples are processed to produce total and dissolved samples;
- field sample split bias where total and dissolved parameters samples are produced from the same grab sample.

For further clarification on any of the above information, please contact your ALS representative.

# **RESULTS OF ANALYSIS - Water**



Sample ID			1A	1B	Travel Blank
Sample Date Sample Time ALS ID			06-05-20 08:20 1	06-05-20 08:25 2	3
Physical Tests Colour Conductivity Total Dissolved S Hardness pH	(CU) (uS/cm) olids CaCO3		75.7 16000 30800 8070 3.27	73.7 15900 34300 8030 3.27	<5.0 <2.0 <10 <0.54 5.89
Total Suspended Turbidity	Solids (NTU)		21.1 7.82	44.4 6.75	<3.0 <0.10
Dissolved Anions Acidity (to pH 8.3) Alkalinity-Total Bromide Chloride Fluoride	Br Cl F	CaCO3 CaCO3	11300 <2.0 <0.50 40.5 18.5	11600 <2.0 <0.50 40.5 20.8	1.8 <2.0 <0.050 <0.50 <0.020
Sulphate	SO4		19500	18900	<0.50
Nutrients Ammonia Nitroger Total Kjeldahl Nitro Nitrate Nitrogen Nitrite Nitrogen Total Nitrogen	n ogen	N N N N	6.26 6.05 31.4 0.046 43.2	5.05 6.24 31.2 0.030 44.6	<0.0050 <0.050 <0.0050 <0.0010 <0.050
Total Phosphate		P	0.0633	0.0673	<0.0020
<u>Cyanides</u> Total Cyanide	CN		0.0013	<0.0010	

Remarks regarding the analyses appear at the beginning of this report. Results are expressed as milligrams per litre except where noted. < = Less than the detection limit indicated.

# **RESULTS OF ANALYSIS - Water**



Sample ID		1A	1B	Travel Blank
Sample Date Sample Time ALS ID		06-05-20 08:20 1	06-05-20 08:25 2	3
Total Metals Aluminum Antimony Arsenic Barium Beryllium	T-AI	843	875	<0.0010
	T-Sb	<0.10	<0.10	<0.00010
	T-As	<0.10	<0.10	<0.00010
	T-Ba	<0.050	<0.050	<0.000050
	T-Be	<0.50	<0.50	<0.00050
Bismuth	T-Bi	<0.50	<0.50	<0.00050
Boron	T-B	<10	<10	<0.010
Cadmium	T-Cd	2.98	3.09	<0.000050
Calcium	T-Ca	442	442	<0.050
Chromium	T-Cr	<0.50	<0.50	<0.0050
Cobalt	T-Co	2.22	2.30	<0.00010
Copper	T-Cu	1.83	1.87	<0.00010
Iron	T-Fe	122	121	<0.030
Lead	T-Pb	0.292	0.286	<0.000050
Lithium	T-Li	<5.0	<5.0	<0.0050
Magnesium	T-Mg	1670	1670	<0.10
Manganese	T-Mn	673	712	<0.000050
Mercury	T-Hg	0.000268	0.000167	<0.000050
Molybdenum	T-Mo	<0.050	<0.050	<0.000050
Nickel	T-Ni	4.25	4.44	<0.00050
Phosphorus	T-P	<3.0	<3.0	<0.30
Potassium	T-K	30	30	<2.0
Selenium	T-Se	<1.0	<1.0	<0.0010
Silicon	T-Si	33.6	33.5	<0.050
Silver	T-Ag	<0.010	<0.010	<0.000010
Sodium	T-Na	35	35	<2.0
Strontium	T-Sr	1.39	1.44	<0.00010
Thallium	T-TI	<0.10	<0.10	<0.00010
Tin	T-Sn	<0.10	<0.10	<0.00010
Titanium	T-Ti	<0.10	<0.10	<0.010
Uranium	T-U	0.171	0.177	<0.000010
Vanadium	T-V	<1.0	<1.0	<0.0010
Zinc	T-Zn	4140	4320	<0.0010

Remarks regarding the analyses appear at the beginning of this report. Results are expressed as milligrams per litre except where noted. < = Less than the detection limit indicated.

# **RESULTS OF ANALYSIS - Water**



Sample ID			1A	1B
Sample Date Sample Time ALS ID			06-05-20 08:20 1	06-05-20 08:25 2
<u>Dissolved Me</u> Aluminum Antimony Arsenic Barium Beryllium	etals D-Al D-Sb D-As D-Ba D-Be		838 <0.10 <0.10 <0.050 <0.50	848 <0.10 <0.10 <0.050 <0.50
Bismuth	D-Bi		<0.50	<0.50
Boron	D-B		<10	<10
Cadmìum	D-Cd		3.05	2.88
Calcium	D-Ca		440	441
Chromium	D-Cr		<0.50	<0.50
Cobalt	D-Co		2.23	2.27
Copper	D-Cu		1.94	1.86
Iron	D-Fe		120	121
Lead	D-Pb		0.292	0.273
Lithium	D-Li		<5.0	<5.0
Magnesium	D-Mg		1690	1680
Manganese	D-Mn		676	688
Mercury	D-Hg		0.000235	0.000205
Molybdenum	D-Mo		<0.050	<0.050
Nickel	D-Ní		4.48	4.27
Phosphorus	D-P		<3.0	<3.0
Potassium	D-K		30	30
Selenium	D-Se		<1.0	<1.0
Silicon	D-Si		33.4	33.5
Silver	D-Ag		<0.010	<0.010
Sodium	D-Na		35	35
Strontium	D-Sr		1.41	1.39
Thallium	D-TI		<0.10	<0.10
Tin	D-Sn		<0.10	<0.10
Titanium	D-Ti		<0.10	<0.10
Uranium	D-U		0.176	0.173
Vanadium	D-V		<1.0	<1.0
Zinc	D-Zn		4190	4200
Organic Paran Total Organic	<u>neters</u> Carbon	С	19.2	18.7

Remarks regarding the analyses appear at the beginning of this report. Results are expressed as milligrams per litre except where noted. < = Less than the detection limit indicated.

### Appendix 1 - METHODOLOGY



Outlines of the methodologies utilized for the analysis of the samples submitted are as follows

#### Colour in Water

This analysis is carried out using procedures adapted from APHA Method 2120 "Color". Colour (true colour) is determined by filtering a sample through a 0.45 micron membrane filter followed by analysis of the filtrate using the platinum-cobalt colourimetric method. The analysis is carried out without pH adjustment.

Recommended Holding Time:

Sample: 2 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### Conductivity in Water

This analysis is carried out using procedures adapted from APHA Method 2510 "Conductivity". Conductivity is determined using a conductivity electrode.

Recommended Holding Time:

Sample: 28 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### Solids in Water

This analysis is carried out using procedures adapted from APHA Method 2540 "Solids". Solids are determined gravimetrically. Total dissolved solids (TDS) and total suspended solids (TSS) are determined by filtering a sample through a glass fibre filter, TDS is determined by evaporating the filtrate to dryness at 180 degrees celsius, TSS is determined by drying the filter at 104 degrees celsius. Total solids are determined by evaporating a sample to dryness at 104 degrees celsius. Fixed and volatile solids are determined by igniting a dried sample residue at 550 degrees celsius.

Recommended Holding Time:

Sample: 7 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### pH in Water

This analysis is carried out using procedures adapted from APHA Method 4500-H "pH Value". The pH is determined in the laboratory using a pH electrode.

# Appendix 1 - METHODOLOGY - Continued

Recommended Holding Time:

Sample: 2 hours Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### **Turbidity of Water**

This analysis is carried out using procedures adapted from APHA Method 2130 "Turbidity". Turbidity is determined by the nephelometric method.

Recommended Holding Time:

Sample: 2 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### **Acidity in Water**

This analysis is carried out using procedures adapted from APHA Method 2310 "Acidity". Acidity is determined by potentiometric titration to a specified endpoint.

Recommended Holding Time:

Sample: 14 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

### Alkalinity in Water by Colourimetry

This analysis is carried out using procedures adapted from EPA Method 310.2 "Alkalinity". Total Alkalinity is determined using the methyl orange colourimetric method.

Recommended Holding Time:

Sample: 14 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### Dissolved Anions in Water by Ion Chromatography

This analysis is carried out using procedures adapted from APHA Method 4110 "Determination of Anions by Ion Chromatography" and EPA Method 300.0 "Determination of Inorganic Anions by Ion Chromatography". Anions are determined by filtering the sample through a 0.45 micron membrane filter and injecting the filtrate onto a Dionex IonPac AG17 anion exchange column with a hydroxide eluent stream. Anions routinely determined by this method include: bromide, chloride, fluoride, nitrate, nitrite and sulphate.



# Appendix 1 - METHODOLOGY - Continued



Recommended Holding Time:

Sample: 28 days (bromide, chloride, fluoride, sulphate)

Sample: 2 days (nitrate, nitrite) Reference: APHA and EPA

Laboratory Location: ALS Environmental, Vancouver

#### Ammonia in Water by Selective Ion Electrode

This analysis is carried out, on sulphuric acid preserved samples, using procedures adapted from APHA Method 4500-NH3 "Nitrogen (Ammonia)". Ammonia is determined using an ammonia selective electrode.

Recommended Holding Time:

Sample: 28 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

### Total Kjeldahl Nitrogen in Water

This analysis is carried out using procedures adapted from APHA Method 4500-Norg "Nitrogen (Organic)". Total kjeldahl nitrogen is determined by sample digestion at 367 celcius with analysis using an ammonia selective electrode.

Recommended Holding Time:

Sample: 28 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

# Total Kjeldahl Nitrogen and Total Nitrogen in Water

This analysis is carried out using procedures adapted from ASTM Method D 5176-91 "Standard Test Method for Total Chemically Bound Nitrogen in Water by Pyrolysis and Chemiluminescence detection." Total Nitrogen is determined directly by pyrolysis with chemiluminescence detection using automated instrumentation. Total Kjedahl Nitrogen is determined by calculation.

Recommended Holding Time:

Sample: 28 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

# Appendix 1 - METHODOLOGY - Continued



#### Phosphate in Water

This analysis is carried out using procedures adapted from APHA Method 4500-P "Phosphorus". All forms of phosphate are determined by the ascorbic acid colourimetric method. Dissolved ortho-phosphate (dissolved reactive phosphorous) is determined by direct measurement. Total phosphate (total phosphorous) is determined after persulphate digestion of a sample. Total dissolved phosphate (total dissolved phosphorous) is determined by filtering a sample through a 0.45 micron membrane filter followed by persulfate digestion of the filtrate.

Recommended Holding Time:

Sample: 2 days Reference: EPA

Laboratory Location: ALS Environmental, Vancouver

#### Cyanide Species in Water

This analysis is carried out using procedures adapted from APHA Method 4500-CN "Cyanide". Total or strong acid dissociable (SAD) cyanide and weak acid dissociable (WAD) cyanide are determined by sample distillation and analysis using the chloramine-T colourimetric method. Cyanate is determined by the cyanate hydrolysis method using an ammonia selective electrode. Thiocyanate is determined by the ferric nitrate colourimetric method.

Recommended Holding Time:

Sample: 14 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### Metals in Water

This analysis is carried out using procedures adapted from "Standard Methods for the Examination of Water and Wastewater" 20th Edition 1998 published by the American Public Health Association, and with procedures adapted from "Test Methods for Evaluating Solid Waste" SW-846 published by the United States Environmental Protection Agency (EPA). The procedures may involve preliminary sample treatment by acid digestion, using either hotplate or microwave oven, or filtration (EPA Method 3005A). Instrumental analysis is by atomic absorption/emission spectrophotometry (EPA Method 7000 series), inductively coupled plasma - optical emission spectrophotometry (EPA Method 6010B), and/or inductively coupled plasma - mass spectrometry (EPA Method 6020).

Recommended Holding Time:

Sample: 6 months Reference: EPA

# Appendix 1 - METHODOLOGY - Continued



Laboratory Location: ALS Environmental, Vancouver

#### Mercury in Water

This analysis is carried out using procedures adapted from "Standard Methods for the Examination of Water and Wastewater" 20th Edition 1998 published by the American Public Health Association, and with procedures adapted from "Test Methods for Evaluating Solid Waste" SW-846 published by the United States Environmental Protection Agency (EPA). The procedure involves a cold-oxidation of the acidified sample using bromine monochloride prior to reduction of the sample with stannous chloride. Instrumental analysis is by cold vapour atomic fluorescence spectrophotometry (EPA Method 245.7).

Recommended Holding Time:

Sample: 28 days Reference: EPA

Laboratory Location: ALS Environmental, Vancouver

#### Carbon in Water

This analysis is carried out using procedures adapted from APHA Method 5310 "Total Organic Carbon (TOC)". All fractions of carbon are determined by the combustion-infrared method. Total carbon includes organic carbon (covalently bonded in organic molecules) and inorganic carbon (carbonate, bicarbonate and dissolved carbon dioxide). Total organic carbon is the calculated difference between the total carbon and the inorganic carbon determination. Dissolved carbon fractions are determined by filtering the sample through a 0.45 micron membrane filter prior to analysis.

Recommended Holding Time:

Sample: 28 days Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

#### Ammonia in Water by Colourimetry

This analysis is carried out, on unpreserved samples, using procedures adapted from APHA Method 4500-NH3 "Nitrogen (Ammonia)". Ammonia is determined using the phenate colourimetric method.

Recommended Holding Time:

Sample: 1 day Reference: APHA

Laboratory Location: ALS Environmental, Vancouver

# File No. X6065 Appendix 1 - METHODOLOGY - Continued



Results contained within this certificate relate only to the samples as submitted.

This Certificate Of Analysis shall only be reproduced in full, except with the written approval of ALS Environmental.

**End of Report** 

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APPENDIX E

MICROMED MICROBIAL ANA

S REPORT

(RESCAL.

.Y 20, 2006)



### IG MicroMed Environmental !

190 - 12860 Clarke Place, Richmond, B.C. V6V Tel: (604) 279-0666 Fax: (604) 279-

Environmental Engineer Rescan Environmental Services Ltd. Sixth Floor, 111 West Hastings Street Vancouver, B.C. V6E 2J3

12 June, 2006

Ph: (604) 689-9460

ence No: 117962.

These are the results of the samples received May 23.

Product Sampled: Two various samples were received in the laboratory for anal  $\pm$ 

Product:	Heterotrophic Plate Count	Sulphate Reducers	/drocarbon <u>Jers – Diesel Fuel</u>
Sample #: 1 1A 8:10 a.m.	7	Not seen microscopically.	<50
Sample #: 2 1B 8:10 a.m.	9	Not seen microscopically.	<50

Note: All positive and negative controls, media sterility and media performance.c<sup>1</sup> satisfactory.

are found to be

All results are for cfu's per ml or per 100 mls of sample.

Teck Cominco Limited Technology Division #600 - 200 Burrard Street Vancouver, B.C. V6C 3L9



### **MEMORANDUM**

Memo To:

Walter Kuit

Date:

June15th, 2006

From:

Re:

Airborne Thermal Infrared Survey, Sullivan Waste Dump No. 1

An airborne survey was arranged to collect thermal infrared data over the No. 1 waste dump at the Sullivan Mine in the days following the incident that led to four fatalities at the mine site between May 15<sup>th</sup> and May 17<sup>th</sup>, 2006. The objective was to identify any thermal anomalies that might indicate a heat source beneath the till cover that had been applied to seal and re-slope the underlying waste dump material. The survey was completed on May 30<sup>th</sup> and May 31<sup>st</sup> and the data for the flight lines over the accident site and waste dump has been arriving over the past 2 or 3 weeks and compiled. We also were able to locate detailed orthophotos and digital contour maps with 0.5m accuracy from a survey done by McElhaney in 2005. This was used for reference and ground control.

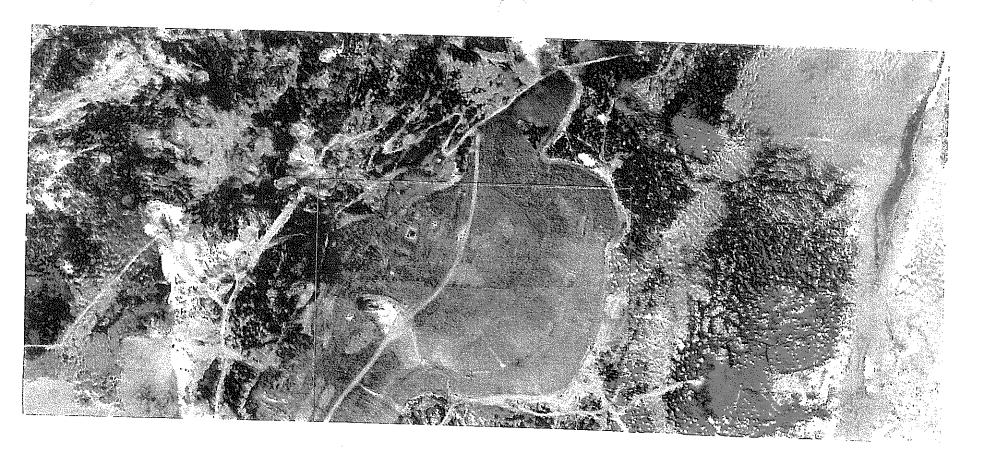
Four areas were covered by the survey but we have only received data for the important No. 1 waste dump to date. The image analysis and statistical work that I have been doing with the raw data provided seems to indicate that there is no obvious "hot spot" that can be identified. There are irregularities and anomalous areas that should be explained for the sake of completeness, but in general the waste dump cover appears to be homogeneous and the temperatures measured over the dump varied by less than 2°C. I believe that most of the variations will be explained by moisture content and variations in the composition and texture of the till cover. However, we can't be certain of this until some of the areas identified have been checked on the ground.

Three sets of thermal IR data were collected: mid-wave thermal and long-wave thermal IR instruments were flown separately on the evening of May 30<sup>th</sup>, just after sunset; and a further mid-wave thermal survey was flown the following morning. The mid wave camera is far more sensitive, and was used to compare evening and morning readings to get an idea of the characteristic thermal inertia properties of materials on the ground. This is done by subtracting the morning readings from the evening readings to measure the rate of cooling. This information was gathered at a resolution of 0.75m pixels. The long wave instrument is capable of calibration to true ground temperatures but has a much coarser resolution of approximately 2m. All three data sets were corrected for aircraft altitude, pitch, yaw and roll, and then geometrically registered to the McElhaney orthophotos. I spent some time looking at the quality of the data and did various statistical analyses to look for any anomalous patterns over the dump or near the shed where the accident happened.

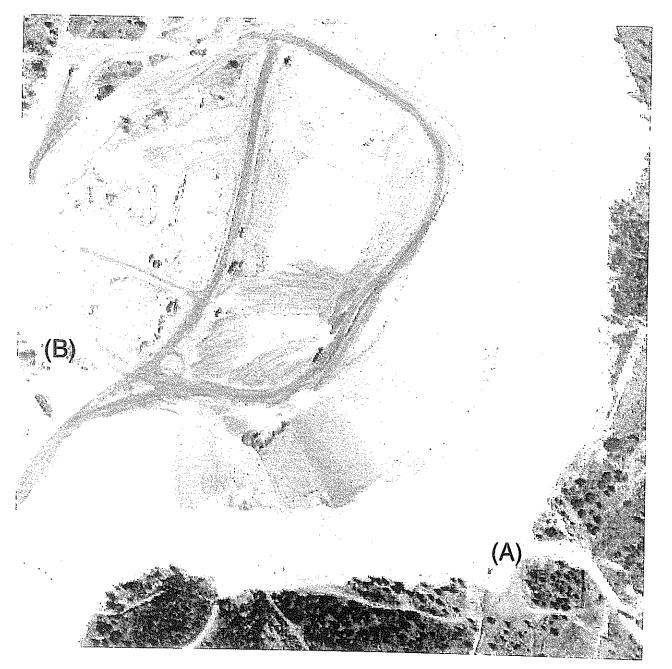
I've printed out various maps with this memo, and compiled some close up figures (Figures 1-5) over the area of interest to support my conclusions that although there are a few weak patterns and some anomalies, there is very little indication of anything going on in the dump material below the fill. The enclosed colour composite shows the tremendous variation both spatially and spectrally of the area surrounding the waste dump, while the dump itself at the larger scale is homogeneous in composition and has consistent spectral characteristics across the entire area. The colour pseudo-contour map (Figure 2) illustrates that all of the close-up patterns that are discussed are gleaned from calibrated temperatures that range over a maximum of 2°C and in fact mainly within a 1° range - so these are very weakly detectable. Figure 2 also shows a pattern of increasing temperatures (within the 2° range) as you descend to the lowest levels of the dump. This could be due to slight increases in moisture, but may also simply reflect the lower altitude and different atmospheric conditions as you move down toward the valley. Figures 3 to 5 show various other patterns in the midwave evening, morning and thermal inertia bands.

I can't be certain of any interpretation without some study or field checking, but I think the anomalies labeled A, B, and C on these last three figures are due to varying amounts of moisture in the till. Bruce Donald also suggested that some of them might be areas that were not recently ripped in preparation for seeding as a few areas are designated for trees. If the variations are due to moisture, then I would suspect that the dark areas (A) might represent seeps as they stay cool and change very little in temperature. The warmer patches above these (B) may be due to elevated moisture in the till above, where daytime sun can warm the moist soils up almost to the temperature of the surrounding till. After sunset, the drier soils around these patches cool faster and we see relatively elevated readings in the morning above the wetter "seeps" below, if that is what they are. The two areas on top of the dump (D) in Figures 4 and 5 may be similar damp The warmer slope shown as (C) on the figures is difficult to interpret for soils. three reasons: they may be close to outcrop in places at the western edge of the dump and we can see greater variation in the till patterns to the west of the road where bedrock is closer; parts of this area are close to the portal above where there seems to be an anomalous and difficult to explain pattern, possibly due to warmer air from old workings; and finally, a look at the orthophoto seems to show that the till cover over this particular area is slightly darker and redder than in other areas.

I suspect the three extra strips we are still waiting for will be easier to interpret with some confidence now that I've taken the time to look at the main area in detail. We will only have mid-wave thermal data (both morning and evening), but it will be similar in sensitivity and resolution. As they were taken at the same time as the main survey, it should be possible to approximately calibrate the data to absolute temperatures calculated by the long wave instrument over the waste dump. These last three lines were taken over the old Iron Tailings Pond, the Open Pit area, and the Lower Mine Yard Dumps. Data should be available next week.

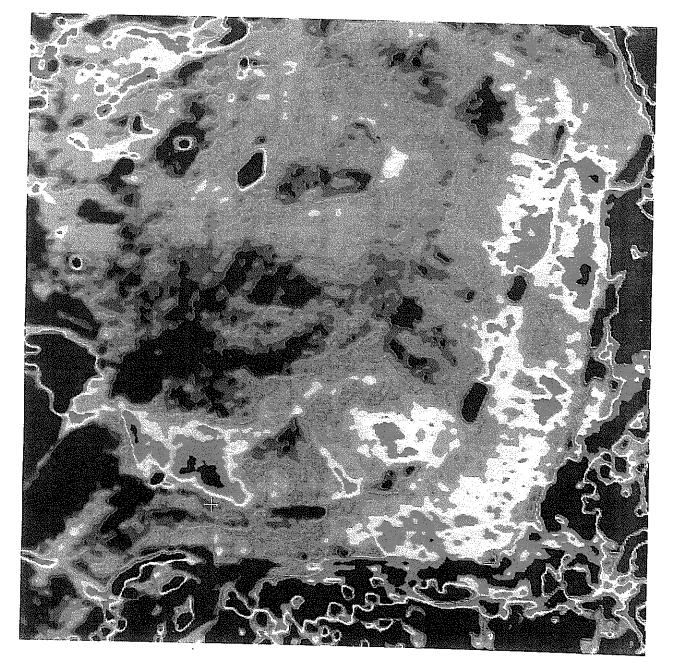


Airborne Thermal Infra-red Image Acquisition, May 30th and May 31st, 2006.



### Sullivan No. 1 Waste Dump Digital Orthophoto

Figure 1. Full resolution air photo flown over the Sullivan Lands in August, 2005. Distance across the close-up image is 500m. The shed (A) where the accidents occurred is on the southeast flank of the dump. The mine portal (B), still open to the underground workings is also shown for reference.

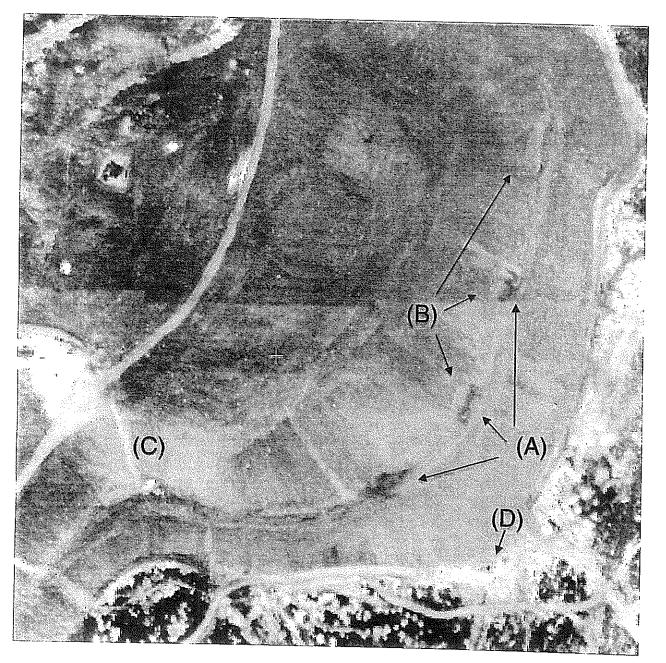


Sullivan No. 1 Waste Dump Long-wave Thermal IR Image, May 30th, 2006.

Figure 2. Calibrated surface temperatures from Long Wave Thermal Infra-red flight data acquired just after sunset.
Black areas within red areas slightly exceed 6° C while black within blue represents slightly less than 5° C. Temperature range across the entire scene is about 2° C. Generally, warmest areas are on the lower flanks of the dump, perhaps reflecting moisture content in the till or elevation differences.

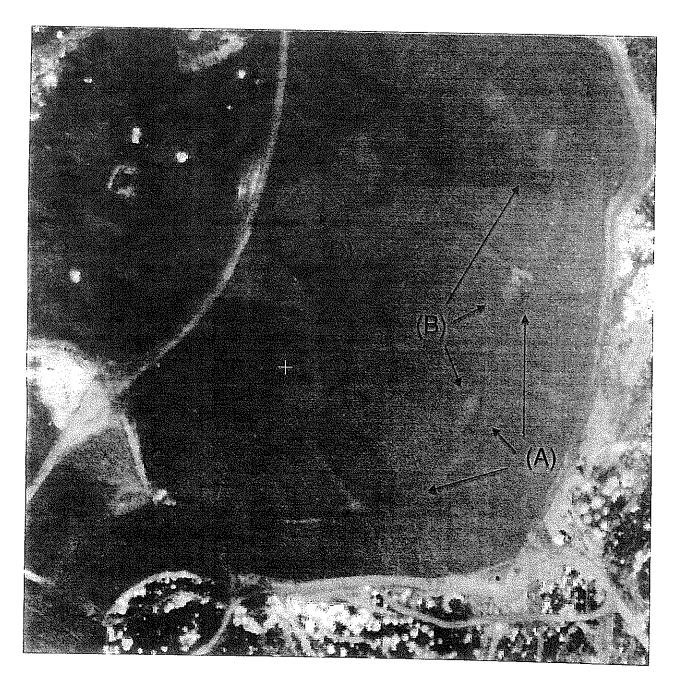
6° C

5° C



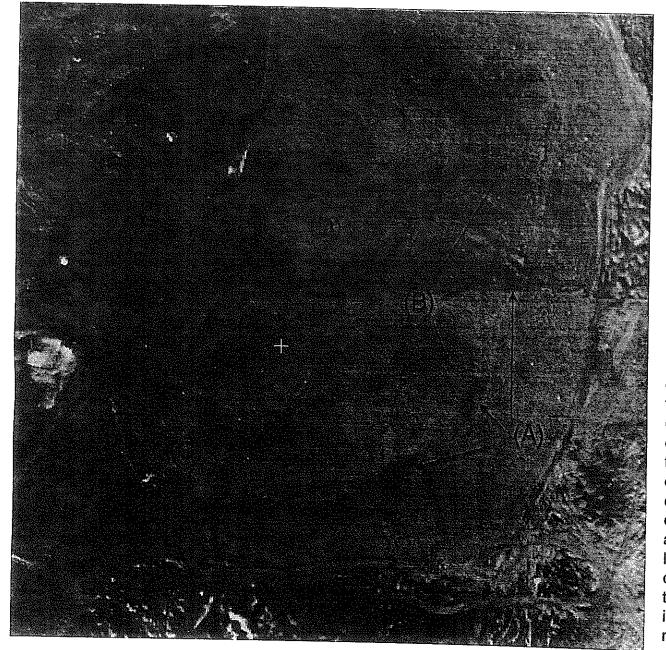
Sullivan No. 1 Waste Dump Evening Mid-wave Thermal IR Image, May 30<sup>th</sup>, 2006.

Figure 3. The till covering the dump appears homogeneous. Dark cool areas (A) at the base of the steep slope are probably due to moisture. Immediately above these patches (B), the till is very slightly warmer than surrounding material. These lighter, warm areas show clearly in the scene taken the next morning. The steep slope (C) just to the east of the main portal is the warmest area on the dump during the day time. The shed where the accidents occurred is the small dark square on the southeast flank of the dump (D). Also apparent on the image are the lighter drainages, or erosional channels that cut down the steep slope. The jagged horizontal line across the middle of the image is the "seam" where 2 flight lines have been joined together.



Sullivan No. 1 Waste Dump Morning Mid-wave Thermal IR Image, May 31st, 2006.

Figure 4. This mid-wave thermal image was acquired just before sunrise, about 6 hours after the evening image. The images are quite different, and the difference between them (Figure 5) can be used to determine the thermal inertia characteristics of different materials on the ground. In the morning image, the light, relatively warmer areas (B) at the base of the slope are now more pronounced than the darker areas (A) that showed up well on the evening shot. The slope at (C) has remained lighter than the surrounding till and two new warm spots (D) are now evident. These new anomalies are on the highest part of the dump where it is flat and probably represent moisture in the till where it collects runoff from the road. This image shows all of the major areas that should be field checked.



### Sullivan No. 1 Waste Dump Thermal Inertia Image.

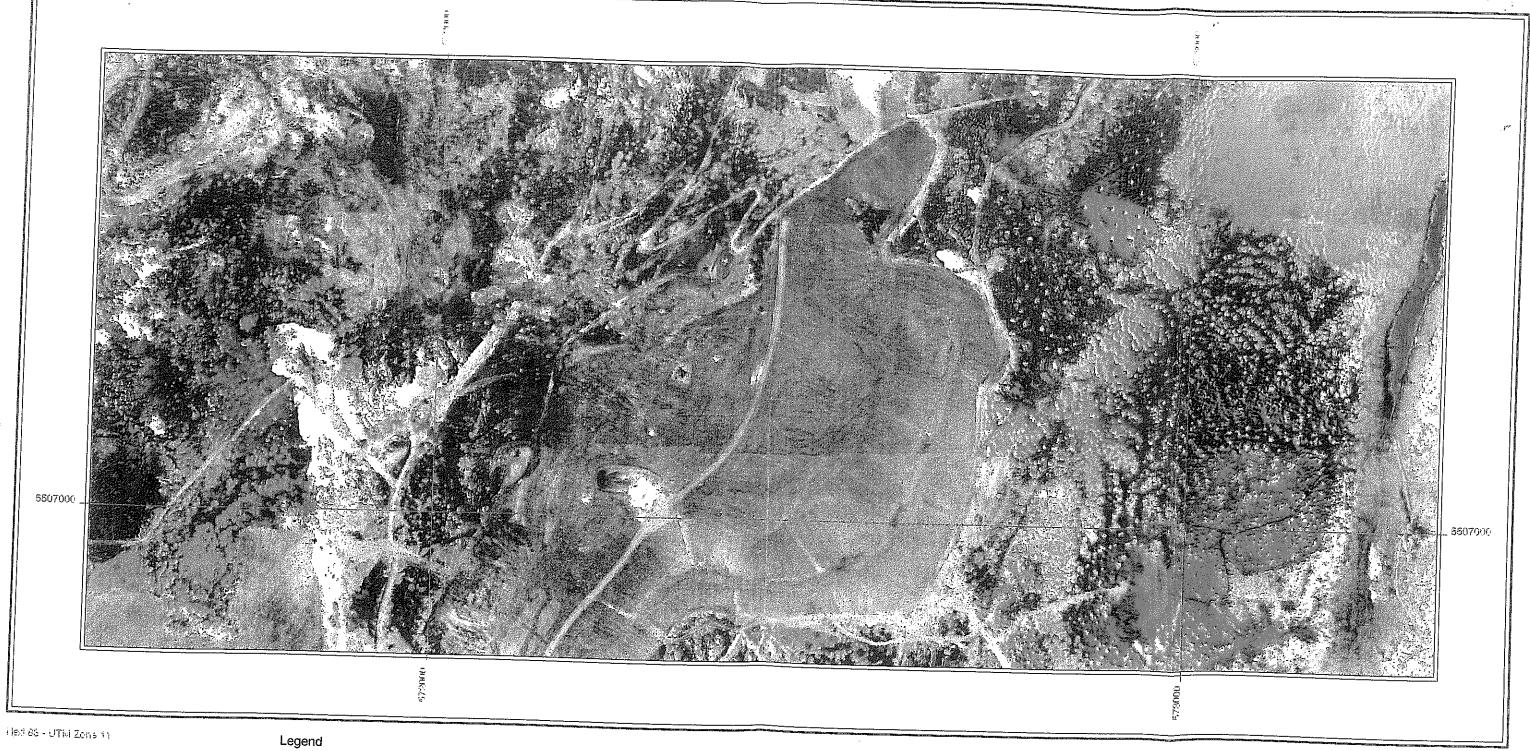
Figure 5. Dark areas indicate areas that have changed temperature the least between evening and morning, while lighter shades indicate the greatest amount of cooling. The dark areas at (A) have stayed cool throughout, while the (B) anomalies have stayed relatively warm as reflected in the morning mid-wave scene. The slope at (C) has stayed relatively warmer just east of the portal as the darker shade here indicates that this area has changed less than others on the steeper sloped part of the dump. The different characteristics on this part of the dump may be due to different till composition as the orthophoto seems to show a darker brown colour here. The area at (D) has cooled down less than other parts of the dump. The smooth outlines of this area seem to re-enforce the idea that this patch is related to moisture in the till.



August, 2005 Digital Orthophoto

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Airborne Thermal Infrared Survey Colour Composite

Scale 1:5 000



Transverse Mercator Projection



Airborne Evening Mid-wave Thermal Infrared Survey, May 30, 2006

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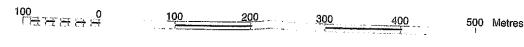
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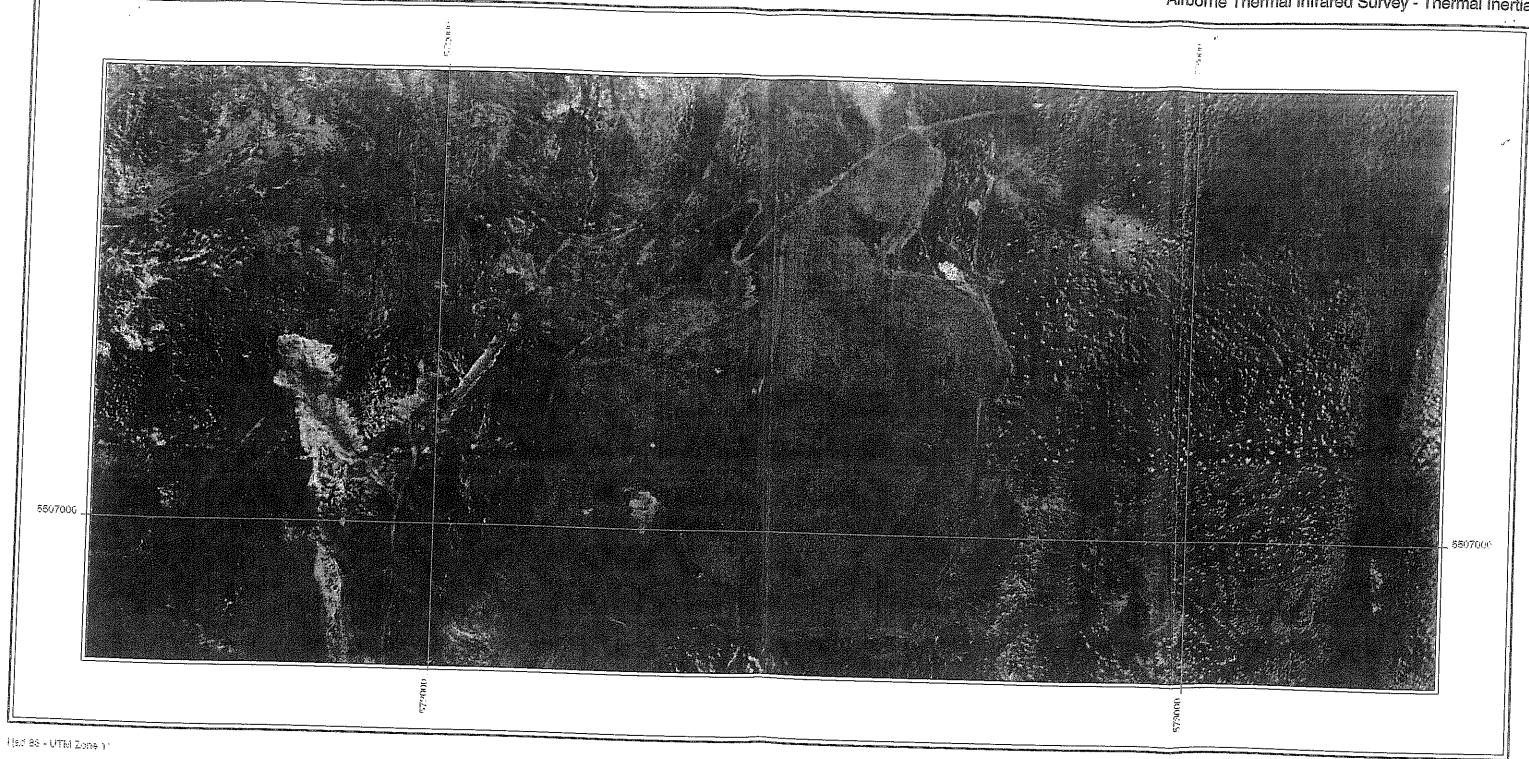
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Airborne Morning Mid-wave Thermal Infrared Survey, May 31, 2006

Scale 1:5 000





Airborne Thermal Infrared Survey - Thermal Inertia

Scale 1:5 000

Gas and Water Sampling Report #1 Waste Dump Monitoring Shed Sullivan Mine Fatalities 19 May 06

Al Hoffman, P.Eng.
Manager, Occupational Health
Ministry of Energy, Mines and Petroleum Resources
August 2006

#### 1. Introduction:

As a result of an accident involving four fatalities in a water monitoring shed at the Sullivan Mine #1 Waste Dump, gas and water samples were acquired in the early morning hours of Friday May 19, 2006 starting at 1:37 am,. Water samples were collected to determine if the water effluent flowing through the sump area of the shed could be a significant source of off-gassing and worker exposure.

This report confines itself to the gas and water sampling that was conducted by the Ministry of Energy, Mines and Petroleum Resources (MEMPR) and the Teck Cominco HAZMAT team in the early morning hours of Friday, May 19, 2006. Some broader conclusions will be drawn from work that was undertaken by Teck Cominco and Rescan Environmental Services Ltd.

#### 2. Methodology

A drawing illustrating No. 1 Waste Dump Monitoring Shed as-built drawing is attached as Appendix 1. A complete review of the water sampling shed location and sampling protocols is outlined in the Teck Cominco portion of the fatality report. The water sampling shed was located on the toe of the No. 1 Waste Dump. It was constructed over a weir that was used as a location to sample both the volume and quality of water that was seeping through the dump and collected in a French drain running along the toe of the dump. Water exiting the French drain was collected in a 400 mm HDPE pipe that was enclosed in a culvert and buried with rock and fines. The effluent then flowed across the sump floor and exited the monitoring shed via a smaller diameter 200 mm HDPE pipe. After leaving the immediate toe area of the dump, the effluent flowed downhill to another sampling shed located at the 3700 Level portal and eventually into the St. Mary's river treatment plant.

Workers periodically entered this shed to measure the volumetric flow rate of water passing through the shed and to and collect water samples for further analysis. A photograph showing a member of the HAZMAT team entering the sampling shed is shown in Figure 1 below:

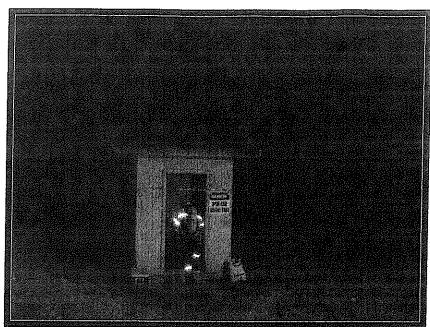
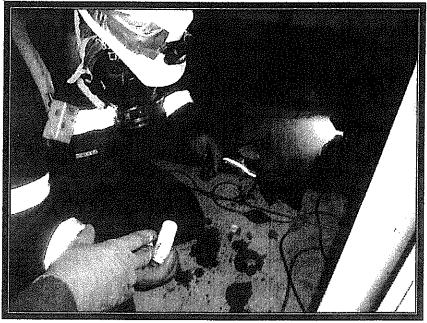


Figure 1 Team member enters the water monitoring shed.

A total of nine gas bag samples were collected from three locations in the sampling shed using a sampling train consisting of a metal probe, plastic tubing and a squeeze bulb. Three control samples were collected in fresh air approximately 200 yards upwind of the site. All samples were collected in one litre Tedlar gas bags.

Water samples were also collected from the inlet pipe at the bottom of the shed sump as shown in Figure 2.



All samples were collected by the Teck Cominco HAZMAT team. Team members were self-contained breathing apparatus and followed their applicable company emergency

procedures. After the samples were collected, they were sealed in coolers with legal tags and transported by road to the ALS Laboratory facilities in Edmonton, Alberta for analysis.

#### 3. Results:

The water sampling results did not show a significant source of off-gassing. No bubbling of the water effluent was noted at the time of collection.

The Teck Cominco HAZMAT team noted a significant inflow of air entering the shed via the upstream 400 mm HDPE pipe. Air velocities reaching 180 fpm exhausting the inlet pipe were estimated using a velometer. Given that the inlet pipe was approximately half full with sediment, the incasting volume of air was approximately 125 cfm.

Table I below summarizes the sampling location and analytical gas sampling results. The complete gas and water analytical results are attached in Appendix 2. One set of samples was retained for legal purposes by ASL Laboratories. The third set of samples was retained by Teck Cominco Ltd. for possible further analysis.

Table I Summary of Sampling Results Conducted Friday May 19th 2006 starting 1:37 am

Sample Location	Sample No	Oxygen ppm	Carbon Dioxide ppm	Nitrogen %	Carbon monoxide ppm	Hydrogen Sulphide ppb
Centre of Inlet Pipe	1B Pipe Air	2	59000	92	<1	<10
1ft < floor grade	2B below grade	2	57000	91	<1	<10
5ft> floor grade	3B above grade	18	10000	80	<1	<10
200 yards west of shed	Clean Air 2	21	49	78	<1	<10

The sampling results in the sump area illustrate that oxygen levels were far below normal levels of approximately 20.9%. The literature indicates that exposures to oxygen levels in the range of 1 -2 % can result in immediate collapse. In fact, the diffusive process of oxygenating blood via the alveoli actually reverses at these concentrations(1). Oxygen flows from the victim's blood stream back into the ambient air. Death can result very quickly.

It is interesting to note that at a sampling location approximating the breathing zone at five feet above the shed floor, oxygen levels had only been depleted to approximately 18%. (2) found oxygen levels above the floor grade in the range of 12 to 20% Therefore, it is possible that the victims or other individuals entering the shed felt no ill effects if they did not descend the ladder to the sump area.

Nitrogen levels were enriched from normal levels of approximately 78.1% to a range from 80 to 92%. Nitrogen gas is considered to be an inert simple asphixiant; thus, the

negative health effects of elevated levels are generally associated with depleted oxygen as is the case in this circumstance.

Carbon dioxide concentrations ranged from 10,000 to 59,000 ppm. 2) found concentrations in the shed above the floor grade ranging from approximately 2500 to 30,000 ppm. The effects of enriched carbon dioxide levels have also been described in the literature. Exposures at 30,000 ppm have resulted in reduced hearing acuity, increased blood pressure and pulse rate. Signs of intoxication were produced by a 30 minute exposure at 50,000 ppm (3). Although concentrations of carbon dioxide were elevated in the monitoring shed, it is very unlikely that the carbon dioxide levels found would solely cause the immediate collapse of those fatally injured.

The hydrogen sulphide (H<sub>2</sub>S) analytical results were less than the detectable limits of 10 ppb. The current MEMPR threshold limit value (TLV) for H<sub>2</sub>S is 10 ppm or 10,000 ppb (4).

Carbon monoxide levels were found to be less than the detectable limit of 1 ppm. The current MEMPR TLV for CO<sub>2</sub> is 25 ppm (4).

#### 4. Conclusions:

Oxygen depletion appears to be the immediate cause of collapse and consequent death of the victims. Carbon dioxide levels, although elevated, would probably not have solely resulted in the collapse of the four individuals. Neither elevated carbon monoxide nor hydrogen sulphide gas levels were found in the sampling shed. Nitrogen levels were elevated but this was a result of the oxygen depletion of incasting air from the waste dump.

#### 5. Recommendations

RESCAN, a consulting company collected additional water and air samples on the morning of Saturday May 20, 2006. A complete review of these results should be undertaken to see if the conclusions stated above are supported.

Teck Cominco is developing an onsite sampling program to determine the reaction that is occurring in the waste dump which resulted in the oxygen depletion of the incasting air. A permanent sampling facility will be build adjacent to the monitoring shed.

Specific recommendations to reduce the likelihood of similar fatalities are located elsewhere in the fatality report.

#### References

- 1. Hazards and Risks of Confined Spaces, Oxygen and Deficiency, Government of Western Australia SafetyLine Institute publication Level 1 Course 8.
- 2. Sullivan Mine Incident Investigation, May 15 to May 20<sup>th</sup> 2006, Prepared by Rescan Environmental Services.
- 3. Carbon Dioxide Threshold Limit Value Criteria, American Conference of Governmental and Industrial Hygienists, 2001.
- 4. Health, Safety and Reclamation Code for Mines in British Columbia, Part 2, Table 2-1, published April 2003.



### MINISTRY OF PUBLIC SAFETY AND SOLICITOR GENERAL

#### **BC CORONERS SERVICE**

#### Memo

To: Fed Hermann

FROM: Bruce Chamberlayne

TITLE: Chief Mining Inspector

TITLE: Regional Coroner

DEPT:

DEPT:

**DATE:** July 14, 2006

CC:

SUBJECT: Final Autopsy, Toxicology and Neuropathology Reports

Hi Fred,

This is a copy of the final autopsy, toxicology and neuropathology reports for Douglas Lloyd Erickson. The medical findings for Mr. Erickson and the other victims were identical. Our investigation confirms that the sampling site was low in oxygen and no other factors were causal or contributory towards their deaths.

I will be in touch in late August to set up a conference call with WorksafeBC and ourselves.

Regards,

Bruce

Bruce.Chamberlayne@gov.bc.ca