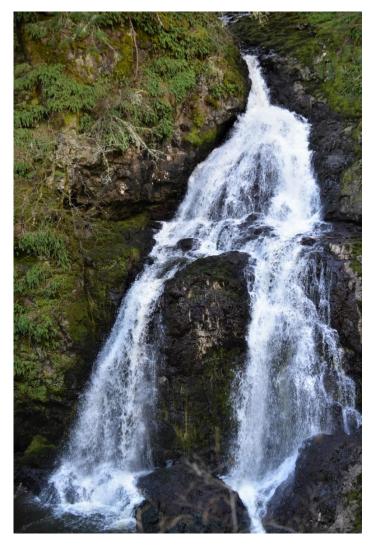
Zinc Water Quality Guidelines - Freshwater Aquatic Life

Ministry of Water, Land, and Resources Stewardship Water Protection & Sustainability Branch







Erratum: Zinc Water Quality Guidelines – Freshwater Aquatic Life (2023)

Page 7 and Page 20, the 1997 guideline was denoted as dissolved zinc. This has been amended to reflect that the 1997 guideline was derived for total zinc.

Table 7.1, the 1997 guideline was denoted as dissolved zinc. This has been amended to reflect that the 1997 guideline was derived for total zinc.

Page 21, Figure 9.1 compared dissolved zinc to the 1997 zinc water quality guideline. This has been amended to compare total zinc to the 1997 zinc water quality guideline.

The Water Quality Guideline Series is a collection of British Columbia (B.C.) Ministry of Environment and Climate Change Strategy water quality guidelines. Water quality guidelines are developed to protect a variety of water values and uses: aquatic life, drinking water sources, recreation, livestock watering, irrigation, and wildlife. The Water Quality Guideline Series focuses on publishing water quality guideline technical reports and guideline summaries using the best available science to aid in the management of B.C.'s water resources. For additional information on B.C.'s approved water quality parameter specific guidelines, visit:

http://www2.gov.B.C.ca/gov/content/environment/air-land-water/water/waterquality/water-quality-guidelines/approved-water-quality-guidelines

ISBN: 978-1-0399-0042-4

Document citation:

B.C. Ministry of Water, Land, and Resource, Stewardship, 2023. Zinc Water Quality Guidelines - Freshwater Aquatic Life. Water Quality Guideline Series, WQG-19-1(1). Prov. B.C., Victoria B.C. Erratum for: B.C. Zinc Water Quality Guidelines - Freshwater Aquatic Life. Water Quality Guideline Series, WQG-19-1.

Cover Photograph:

Location: Sitting Lady Falls, Vancouver Island, B.C.

Disclaimer: The use of any trade, firm, or corporation names in this publication is for the information and convenience of the reader. Such use does not constitute an official endorsement or approval by the Government of British Columbia of any product or service to the exclusion of any others that may also be suitable. Contents of this report are presented for discussion purposes only. Funding assistance does not imply endorsement of any statements or information contained herein by the Government of British Columbia.

EXECUTIVE SUMMARY

The British Columbia Ministry of Water, Land and Resource Stewardship (WLRS) develops province-wide ambient Water Quality Guidelines (WQGs) for substances or physical attributes that are important for managing both the fresh and marine surface waters of British Columbia (B.C.). WQGs provide a basis for water quality assessments and inform decision-making in the natural resource sector. WQGs may be created for the protection of designated values, including aquatic life, wildlife, agriculture, drinking water sources, and recreation. This document presents updated zinc (Zn) WQGs for the protection of freshwater aquatic life.

Elevated concentrations of Zn can adversely affect aquatic and terrestrial life. While background Zn concentrations in B.C. are generally lower than the threshold for adverse effects to biota, anthropogonic activities such as mining can increase Zn concentrations to levels that can be harmful.

In 2018, Canadian Council of Ministers of Environment (CCME) published Canadian WQGs for dissolved Zn for the protection of freshwater aquatic life. B.C. has adopted these guidelines with the addition of an assessment factor to account for the uncertainty of extrapolating laboratory studies to field conditions. While the previous B.C. WQG considered only hardness, the updated guidelines also consider dissolved organic carbon (DOC) and pH. When compared to background Zn concentrations across the province exceedances from the updated WQG were less than 10%.

WQGs for the protection of marine aquatic life, agriculture (irrigation and livestock watering) and wildlife were derived in 1999 and remain unchanged. These technical documents as well as the one for source drinking water WQGs can be found on the B.C. WQG website.

A summary of the WQGs is presented in Table E.1. As the calculation of the aquatic life guideline requires information on the site-specific water chemistry it is listed below as variable though it is generally much lower than WQGs for the other listed values. For example, for a waterbody with pH of 7.5, DOC of 0.5 mg/L, and hardness of 50 mg/L the freshwater aquatic life WQG is 3.5 μ g/L.

-	Guideline	
Value	(μg/L)	Guideline Type
Freshwater Aquatic life	Variable	Long-term chronic
Marine Aquatic life	10	Long-term chronic
Livestock	2,000	Short-term acute
Irrigation	Soil pH < 6	Long-term chronic
Source drinking water	3,000	Maximum allowable concentration

Table E. 2.1. Summary of recommended water quality guidelines for Zinc.

CONTENTS

1.	INTRODUCTION	1
2.	SUBSTANCE IDENTITY	2
3.	SOURCES AND USES	2
4.	FATE, BEHAVIOUR, AND PARTITIONING IN THE ENVIRONMENT	2
5.	BACKGROUND CONCENTRATION OF ZINC IN BRITISH COLUMBIA	2
	5.1. Methods for Estimating Background Concentrations of Zinc in British Columbia Surface Waters5.2. Background Concentration Results	
6.	EFFECTS ON AQUATIC LIFE	7
	6.1. Effects	7
7.	CRITERIA FROM OTHER JURISDICTIONS	7
	7.1. British Columbia	7
	7.2. Canadian Council of Ministers of the Environment (CCME)	
	7.3. Provincial Water Quality Guidelines	
	7.4. USEPA Water Quality Criteria	
	7.5. European Union	
	7.6. Australia and New Zealand	-
8.	RECOMMENDED GUIDELINE	
	8.1. CCME Zinc Water Quality Guidelines	
	8.1.1. Toxicity Data 8.1.2. Toxicity-Modifying Factors	
	8.1.2. Toxicity-wouldying Factors	
	8.2. B.C. Zinc Water Quality Guidelines	
	8.2.1. Chronic long-term water quality guidelines	
	8.2.2. Acute short-term water quality guidelines	
9.	COMPARISON OF AMBIENT ZINC CONCENTRATIONS TO WATER QUALITY GUIDELINES	
RE	FERENCES	.23

LIST OF TABLES

Table E. 1.1. Summary of recommended water quality guidelines for Zinc	ii
Table 5.1. Statistical approach used to calculate station means	
Table 5.2. Summary statistics for station mean dissolved Zn in British Columbia.	4
Table 7.1. Summary of freshwater aquatic life water quality guidelines for Zn by jurisdiction	9
Table 8.1. Endpoints Used to Determine the Long-term Freshwater CWQG for Dissolved Zinc	12
Table 8.2. Endpoints Used to Determine the Short-term Freshwater Benchmark	15
Table 8.3. BC dissolved Zn WQGs (μg/L) at various levels of DOC, pH, and hardness	19
Table 8.4. Examples of short-term WQG (μg/l) for dissolved Zn in freshwater	20

LIST OF FIGURES

Figure 4.1. Distribution of station mean dissolved Zn at background stations in British Columbia	5
Figure 4.2. Distribution of station mean dissolved Zn by waterbody type in British Columbia	6
Figure 7.1. Long-term species sensitivity distribution (SSD) for dissolved zinc in fresh water	13
Figure 7.2. Short-term species sensitivity distribution (SSD) for dissolved zinc in fresh water	17
Figure 8.1. Ambient Zn concentrations compared to the 1997 chronic dissolved Zn WQGs	21
Figure 8.2. Ambient Zn concentrations compared to the 2023 chronic dissolved Zn WQGs	22

1. INTRODUCTION

The British Columbia Ministry of Water, Land, and Resource Stewardship (WLRS) develops province-wide ambient Water Quality Guidelines (WQGs) for substances or physical attributes that are important for managing both the fresh and marine surface waters of British Columbia (B.C.). WLRS defines a WQG as a scientifically derived numerical concentration or narrative statement considered to be protective of designated values in ambient conditions. WQGs provide a basis for water quality assessments and inform decision-making in the natural resource sector and may be derived for the protection of designated uses including aquatic life, wildlife, agriculture (livestock watering and irrigation), source drinking water, and recreation.

In B.C., WQGs are developed to protect the most sensitive endpoint associated with a given value (e.g., aquatic life, wildlife, livestock). For substances with sufficient toxicological data, both short-term acute and long-term chronic guidelines are developed. Interim WQGs are developed when the available toxicological data are insufficient (CCME, 1999; ENV, 2019).

WQGs are typically based on toxicological studies conducted under laboratory conditions. There are several uncertainties associated with applying WQGs to field conditions, including:

- o Laboratory to field differences in exposure conditions,
- Single contaminant tests in laboratories vs exposure to multiple contaminants in the field that may demonstrate additive, synergistic, or antagonistic effects,
- Toxicity of metabolites,
- Intra- and inter-specific differences between test species used to derive the WQG and those found in the field,
- o Indirect effects (e.g., behavioral responses, food web dynamics),
- Laboratory studies conducted on partial life cycle studies which may not include the most sensitive life stage,
- Delayed effects which may not occur within the life stage tested, or may occur across generations, and
- Exposure of organisms to cumulative effects of the various stressors, such as habitat loss and climate change.

Given these uncertainties, WQGs are a predicted no-effect concentration (i.e., no effects are expected if exposure concentrations are below the WQG). An exceedance of the WQGs presented in this document does not imply that unacceptable risks are present, but that the potential for adverse effects is increased and additional investigation and monitoring may be warranted. To that end, ongoing ecological monitoring is encouraged to ensure the WQG is indeed protective under field conditions.

In 2018, Canadian Council of Ministers of the Environment (CCME) published an updated Canadian WQG for dissolved Zn for the protection of freshwater aquatic life. B.C. has adopted this guideline with the addition of an assessment factor to account for the uncertainty of extrapolating laboratory study results to field conditions. This document provides information on CCME's derivation of the aquatic life guideline (replicated here verbatim and highlighted grey) as well as a discussion of background concentrations in B.C. and the choice of assessment factor. The 1997 guidelines for marine aquatic life, wildlife, livestock watering, and irrigation (ENV, 1997) have not been updated and are available in a separate document on the B.C. WQG website.

2. SUBSTANCE IDENTITY

Zinc (Zn; CAS 7440-66-6) is an essential metal found widely in nature. Its predominant oxidation state in the natural environment is Zn²⁺; the metallic form (Zn⁰) is found only in highly reducing environments (Lindsay 1979). Zinc is able to form complexes with a variety of organic ligands and has a variety of salts (WHO 2001). Zinc metal is insoluble in water, but several of its salts are freely soluble (Budavari 1996; Lide 2006).

3. SOURCES AND USES

Canada is one of the largest producers and exporters of zinc. Zinc generally occurs in association with copper and lead; therefore, mining and milling operations usually recover these metals as co-products (NRCan 2007). Zinc is used mainly to coat iron and steel products such as pipes, wire and sheet metal, to render them resistant to corrosion and rust (NRCan 2007). Additionally, many zinc compounds are used in dentistry, medicine, the rubber industry, paint, cosmetics and household products (ATSDR 2007).

4. FATE, BEHAVIOUR, AND PARTITIONING IN THE ENVIRONMENT

Zinc can occur in both suspended and dissolved forms in natural aquatic environments, but most is partitioned into suspended and bottom sediments (Eisler 1993). Several processes control zinc concentrations and mobility in water, and thus its bioavailability to aquatic organisms. Several abiotic variables influence the speciation of zinc, and thus the predominance among zinc forms: most importantly pH, alkalinity, redox potential (Eh), dissolved organic matter and salinity. The most common dissolved zinc species in natural waters under aerobic conditions are ZnOH⁺, Zn²⁺ and ZnCO₃ (Florence 1977; Stumm and Morgan 1981). Of all chemical species found in aquatic environments, the aqueous zinc ion (Zn²⁺) is believed to be the most toxic (ANZECC 2000). Less soluble forms of zinc, such as zinc hydroxide (Zn(OH)₂) and zinc carbonate (ZnCO₃), are considered to be non-toxic (Cairns et al. 1971; Spear 1981). Changes in environmental conditions that influence zinc speciation can result in changes to zinc toxicity.

5. BACKGROUND CONCENTRATION OF ZINC IN BRITISH COLUMBIA

Zinc is a naturally occurring element in aquatic and terrestrial ecosystems, therefore, background concentrations must be considered when deriving provincial Zn WQGs.

5.1. Methods for Estimating Background Concentrations of Zinc in British Columbia Surface Waters

Background (i.e., from non-impacted sites) dissolved Zn concentrations vary across B.C. as a function of local geology and hydrology, therefore, a regional approach was used to estimate background concentrations. Zinc concentrations were estimated following methods used in recent WQG derivation documents (e.g., ENV, 2021). Only data from the B.C. Environmental Management System (EMS) database were used as the Canadian Aquatic Biomonitoring Network (CABIN) database did not contain data for dissolved Zn.

EMS does not identify reference stations, so the database was screened to create a sub-set of water quality stations known to be minimally impacted. To do this, "background" water quality sampling stations that were sampled at least three times over 20 years for any water quality parameter (1996/01/01 to 2017/05/03) were extracted. Next, the list of stations with location information was given to ENV environmental impact assessment biologists to identify sites that they considered minimally impacted by

human activities. No strict definition of 'minimally impacted' was given to the biologists and station selection was left to their professional judgement.

The list of minimally impacted stations was then used to extract Zn data from the EMS database within the following date range: 2000/01/01 to 2023/05/24.

The dataset underwent several additional automated and manual data cleaning steps summarized below:

- Where lake samples were available at multiple depths, only surface samples were included;
- non-detect results with a method detection limit (MDL) ≥5 µg/L were removed as these would influence the results of the analysis; and
- samples were excluded where results were missing or reported as 0.

Arithmetic means were calculated for laboratory replicates (analytical replicates taken from one field sample) with the MDL substituted for values below detection. All field replicates were included as independent samples. The final data set consisted of 159 stations with 3603 samples.

Group	Conditions	Approach	Total Stations	Total Samples
1	% non-detects = 100	% of minimum station MDL	6	107
2	0 < % non-detects < 100 AND # detects < 3	Substitute ½ MDL for non- detects and calculate arithmetic mean for all samples	13	36
3	0 < % non-detects < 100 AND # detects ≥ 3	Regression on order statistics	81	399
4	% non-detects = 0	Arithmetic mean	59	3061

Table 5.1. Statistical approach used to calculate station means.

5.2. Background Concentration Results

The distribution of dissolved Zn concentrations by ENV administrative region is summarized in Table 5.2 and Figure 5.1. The median of station means ranged from 0.5 μ g/L (Skeena) to 3.3 μ g/L (Omineca). Of the 159 stations, 62 stations were on lakes and 97 were on rivers. The median of the distribution of station means in lakes (0.5 μ g/L) was similar to that of rivers (0.4 μ g/L).

	Number of Stations	Number		Concentration MDL Range	% Samples	Distribution of Station Means (µg/L)			
Region		of Samples	Date Range	Range Across all Samples (µg/L)	Across all Samples	< MDL	Median	10 th Percentile	90 th Percentile
Cariboo	51	1037	2000 – 2022	0.1 - 45.1	0.1 – 3	57	0.8	0.2	1.8
Kootenay	11	127	2001 – 2022	0.1–19	1-1	40	1.9	0.8	16.0
Lower Mainland	3	19	2002 – 2005	0.4 – 2	0.1 - 0.1	11	1.2	0.4	1.3
Okanagan	7	1176	2000 – 2022	0.1-88	0.1 - 3	59	0.9	0.4	1.3
Omineca	7	12	2005 – 2018	0.1-6	0.1-0.2	NA	3.3	0.6	6.0
Peace	4	44	2012 – 2018	0.1 - 2	0-1-0.1	NA	1.6	0.7	2.0
Skeena	28	390	2000 – 2022	0.1 - 18	0.1 - 2	42	0.5	0.2	2.7
Thompson	5	215	2001 – 2018	0.2 – 60	0.1 - 1	NA	0.6	0.4	6.9
Vancouver Island	43	584	2001 – 2022	0.1 – 152	0.1 – 4	21	0.7	0.3	2.9

Table 5.2. Summary statistics for station mean dissolved Zn at minimally impacted stations in British Columbia.

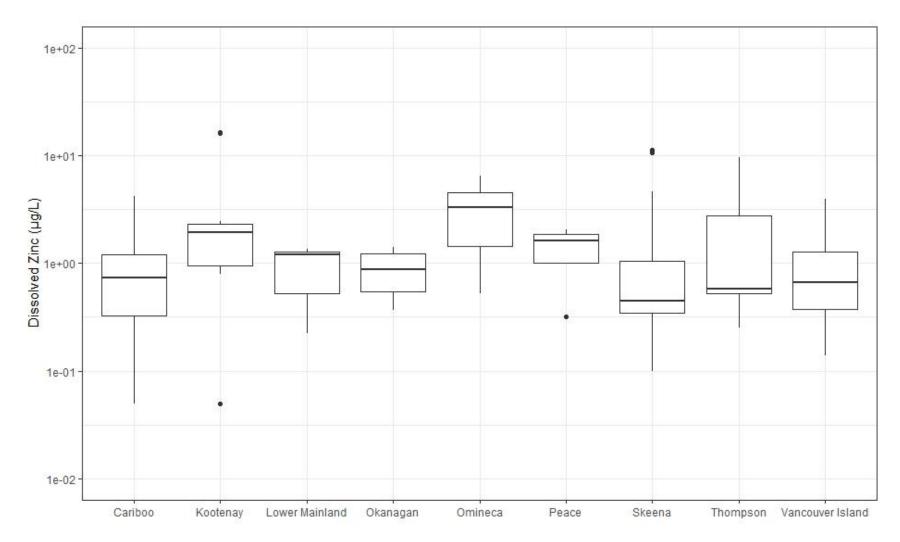


Figure 5.1. Distribution of station mean dissolved Zn at background stations in British Columbia by region. Solid horizontal bar and the lower and higher whiskers represent median, 10th and 90th percentile of station means.

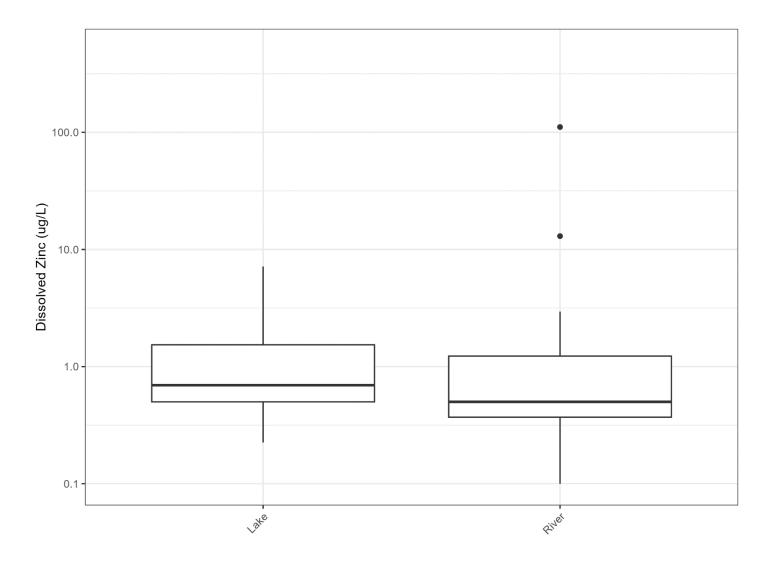


Figure 4.2. Distribution of station mean dissolved Zn at background stations in British Columbia by waterbody type. Solid horizontal bar and the lower and higher whiskers represent median, 10th and 90th percentile of station means.

6. EFFECTS ON AQUATIC LIFE

6.1. Effects

Zinc is an essential element for numerous biological functions. Environmental concentrations that are well below a species' optimal concentration range can disrupt homeostasis, and deficiencies have observable effects (Muyssen and Janssen 2002). At higher concentrations, however, zinc produces adverse chronic and acute effects on reproduction, biochemical and physiochemical reactions, and behavioural effects in aquatic organisms (WHO 2001). In fish, zinc interferes with gill uptake of calcium (Hogstrand et al. 1994; Spry and Wood 1985). Because calcium is also an essential element, this reduction in uptake causes calcium deficiency (Spry and Wood 1985). Zinc also disrupts calcium homeostasis in invertebrates (Muyssen et al. 2006) due to competition between zinc and calcium for the same uptake sites on the gill epithelium (Hogstrand et al. 1994; Hogstrand et al. 1998). Zinc also disturbs, to a lesser extent, sodium and chloride fluxes (Spry and Wood 1985). At higher zinc concentrations, lethal toxicity of zinc to aquatic organisms is caused by the irreversible destruction of the gill epithelium, which causes tissue hypoxia, osmoregulatory failure, acidosis and low oxygen tensions in the arterial blood (Hiltibran 1971; Skidmore 1970; Skidmore and Tovell 1972). In general, salmonids were found to be more sensitive than other types of fish in short-term studies.

7. CRITERIA FROM OTHER JURISDICTIONS

Zinc WQGs/criteria from eight provincial and national jurisdictions are summarized in Table 7.1. Three types of guidelines are used: a static number, a hardness-based equation, and MLR taking the influence of pH, hardness, and DOC into account. In general, most of the older WQGs are static, while more recent WQGs are calculated using MLR. While older guidelines exist for total Zn, more recent guidelines are based on dissolved Zn.

7.1. British Columbia

The B.C. Ministry of Environments, Land and Parks established a hardness-based Zn WQG in 1997 (ENV, 1997) for freshwater aquatic life. This guideline was for total Zn, is based on hardness, and provides both chronic and acute WQGS.

7.2. Canadian Council of Ministers of the Environment (CCME)

The CCME develops national WQGs for the protection of aquatic life and other values. The CCME aquatic life WQG is for dissolved Zn and is calculated using MLR. The short-term WQG takes hardness and DOC into account for guideline calculation (Table 7.1) whereas for the calculation of the long-term WQG pH, hardness and DOC are needed.

7.3. Provincial Water Quality Guidelines

Canadian provinces typically develop or adopt WQGs from another jurisdiction. Manitoba and Quebec have adopted the USEPA 1995 WQC (MWS, 2011; MDDEFP 2013;) and Alberta and Saskatchewan have adopted the previous CCME WQGS of 35 µg/L (Water Security Agency, 2015; AEP, 2018).

7.4. USEPA Water Quality Criteria

The USEPA developed a national WQC for the protection of aquatic life based on dissolved Zn (USEPA, 1995). The USEPA Zn WQC is hardness-based and has identical values for chronic and acute criteria and is calculated using the equation presented in Table 7.1.

7.5. European Union

The European union has derived a predicted no effect concentration (PNEC) of 3.1 μ g/L for waters with low hardness (less than 24 mg/l) and a PNEC of 7.8 μ g/L for water with high hardness (more than 24 mg/l) and is for dissolved Zn (EU, 2006).

7.6. Australia and New Zealand

Australia and New Zealand have joint WQGs, described as trigger values, that invoke a response if exceeded (ANZECC, 2000a; 2000b). Although four trigger values have been calculated to provide various levels of protection (i.e., 80-99% of species), ANZECC (2000a) recommends application of the 80%, 95% and 99% protection levels to protect highly disturbed ecosystems, slightly-moderately disturbed ecosystems, and high conservation/ecological value ecosystems, respectively (ANZECC, 2000a). The Zn trigger value is dependent on the site-specific hardness.

Table 7.1. Summary of freshwater aquatic life water quality guidelines for Zn by jurisdiction.

Jurisdiction	Chronic (µg/L)		Acute (μg/L)	Total/ dissolved	Year published
British Columbia	7.5 + 0.75 (water hardness-90)		33 + 0.75 (water hardness-90)	Total	1997
CCME	exp ^{(0.947[In(hardness)] - 0.815[pH] + 0.398[In(DOC)] + 4}	4.625)	exp ^{(0.833[in(hardness)] + 0.240[in(DOC)] + 0.526)}	Dissolved	2018
Alberta	30		NA	Total	2018
Ontario	30		NA	Total	1994
Saskatchewan	30		NA	Total	2015
USEPA	e ^{(0.8473 [In(hardness)]+ 0.884)} * 0.986		e ^{(0.8473 [In(hardness)]+ 0.884)} * 0.986	Dissolved	1995
European Union	Hardness <24 Hardness ≥24	3.1 7.6	NA	Dissolved	2006
Australia/ New Zealand	NA		a* x TV*(hardness/30) ^{0.85}	Dissolved	2000

TV*: trigger value varies depending on intended level of protection.

a*: constant varies based on hardness.

8. <u>RECOMMENDED GUIDELINE</u>

8.1. CCME Zinc Water Quality Guidelines

8.1.1. Toxicity Data

All zinc toxicity data were evaluated for scientific acceptability before being considered for or used in the derivation of the short-term benchmark concentration and CWQG. Data from toxicity studies were ranked as primary, secondary or unacceptable in terms of acceptability for guideline derivation. The ranking criteria are described fully in the CCME 2007 protocol and are briefly outlined here.

In order for a toxicity value to be considered primary, the concentration of the toxic substance must be measured at the beginning and end of the exposure period, and the measurement of water quality parameters (hardness, pH, temperature, etc.) must be reported. Adequate replication must be performed, suitable statistical procedures should be used and control mortality should be low (typically less than 10%). Secondary data are those that originate from studies where primary data cannot be generated but are still of acceptable quality and documentation. For example, a study may use calculated (rather than measured) substance concentrations, but the most relevant water quality parameters must be reported. Appropriate test replication is still necessary, but pseudo replication may be acceptable for secondary studies (e.g., all test organisms in only one aquarium per concentration). Unacceptable data are those that do not meet the criteria of primary or secondary data.

8.1.2. Toxicity-Modifying Factors

Water chemistry conditions influence how toxic Zn is to aquatic organisms by affecting its environmental fate, behaviour and bioavailability. Water hardness, dissolved organic carbon (DOC), and pH are the most important variables. Complete details of the assessment are available in the scientific criteria document (CCME 2018).

In general, increased hardness is protective (i.e., zinc is less toxic in harder waters), likely due to competitive interactions with Ca²⁺ and Mg²⁺ at binding sites. The effect of pH is not as clear. Authors report that toxicity to fish and algae increases with increasing pH in natural waters, but there are no consistent patterns for invertebrates (CCME 2018). Increased bioavailability of zinc is possible at high pH due to decreased binding by organic ligands. Dissolved organic matter is an important complexing agent for zinc; therefore, increased organic matter (measured as DOC) tends to have a protective effect. The toxicity modifying factors are described more fully in CCME (2018).

It is important to account for exposure and toxicity-modifying factors when deriving guidelines (CCME 2007). This can be done through single- or multi-factor equations, matrices or models. In this case, CCME used multiple linear regression (MLR) analysis to account for the simultaneous effects of water hardness, DOC and pH on zinc toxicity.

CCME derived empirical relationships for short-term and long-term exposure using forward stepwise MLR. The analysis identified which water chemistry variables explained a significant portion of variability in zinc toxicity. MLR analyses were conducted species by species. The best species model was selected based on how well it predicted toxicity, how well it explained variability in the dataset and, in the case of the longterm guideline, the protectiveness of the values calculated by the model.

For the short-term benchmark, hardness and DOC were both significant factors. For the long-term Canadian Water Quality Guideline (CWQG), hardness, pH and DOC all explained a significant portion of the variation and were used to derive the CWQG equation. CCME's short-term benchmark equation is based on a pooled Daphnia magna and *D. pulex* model containing variables for hardness and DOC. The

long-term CWQG equation is based on an Oncorhynchus mykiss model containing variables for hardness, pH and DOC. Accordingly, the CWQG and short-term benchmark for freshwater exposure to zinc are presented as multivariable equations that are a function of water hardness, DOC and pH and allow users to derive guidelines and benchmarks based on the water chemistry of the site under consideration.

8.1.3. Water Quality Guideline Derivation

For the derivation of the short-term guideline, effect concentrations were normalized to a hardness of 50 mg·L-1 as CaCO3 and a DOC concentration of 0.5 mg·L-1 using the pooled *Daphnia* MLR equation. For the derivation of the long-term guideline, effect concentrations were normalized to a hardness of 50 mg·L-1 as CaCO3 and a DOC concentration of 0.5 mg·L-1 and pH of 7.5 using the O. mykiss MLR. Total zinc concentrations were converted to dissolved using a conversion factor for laboratory toxicity data of 0.978 for the short-term guideline and 0.986 for the long-term guideline (USEPA 1996) and plotted (see Figures 1 and 2). For details and references regarding the data included in the SSDs, see the scientific criteria document for the CWQG for zinc and its spreadsheet appendix (CCME 2018).

8.1.3.1. Long-term Water Quality Guidelines

Long-term exposure guidelines identify waterborne concentrations intended to protect all forms of aquatic life for indefinite exposure periods. The minimum data requirements for the Type A guideline approach were met and a total of 29 data points were used to derive the guideline (Table 6). Each species for which appropriate long-term toxicity data were available was ranked according to sensitivity.

No data points fell below the fifth percentile value on the long-term SSD curve. CCME assessed the CWQG for zinc for protectiveness and found it achieved the intended level of protection as per CCME 2007 and 2018). Because the analysis of the long-term data found that water hardness, pH and DOC were significant toxicity-modifying factors, the CWQG is expressed as an equation into which the local water hardness, pH and DOC must be entered in order to produce an appropriate site-specific CWQG.

Equation 2.

CWQG = exp^{(0.947[In(hardness)] - 0.815[pH] + 0.398[In(DOC)] + 4.625)}

Where the CWQG is expressed in dissolved zinc concentration ($\mu g \cdot L^{-1}$), hardness is measured as CaCO3 equivalents in $mg \cdot L^{-1}$, pH is in standard units and DOC is in $mg \cdot L^{-1}$.

Species sensitivity distribution rank	Species	Endpoint	Normalized effect concentration ^a (µg dissolved
1	Chironomus riparius (chironomid)	11-w LOEC (development)	9.89
2	Ceriodaphnia dubia (water flea)	7-d MATC (reproduction)	11.3
3	Pseudokirchneriella subcapitata (green algae)	72-h EC ₁₀ (growth rate)	13.8
4	Daphnia magna (cladoceran)	21-d EC ₁₀ (reproduction)	15.0
5	Potamopyrgus jenkinsi (snail)	12-w MATC (growth)	19.1
6	Jordanella floridae (flagfish)	100-d MATC (growth)	27.9
7	Cottus bairdi (mottled sculpin)	30-d EC10 (mortality)	31.5
8	Brachionus havanaensis (rotifer)	18-d EC_{10} (population growth inhibition)	36.5
9	Phoxinus phoxinus (Eurasian minnow)	150-d LC10 (mortality)	51.0
10	Dreissena polymorpha (zebra mussel)	10-w LC10 (mortality)	51.1
11	Pimephales promelas (fathead minnow)	7-d IC10 (growth)	68.2
12	Brachionus calyciflorus (rotifer)	48-h EC ₁₀ (intrinsic rate of population increase)	73.0
13	Oncorhynchus mykiss (rainbow trout)	30-d LC ₁₀ (mortality)	101
14	Lampsilis siliquoidea (fatmucket)	28-d IC ₁₀ (length)	104
15	Bufo boreas (boreal toad)	4-w MATC (development)	108
16	Lymnaea stagnalis (snail)	28-d EC10 (growth)	113
17	Salmo trutta (brown trout)	58-d MATC (weight)	130
18	<i>Prosopium williamsoni</i> (mountain white fish)	90-d IC10 (biomass)	133
19	Salvelinus fontinalis (brook trout)	24-w IC10 (egg fragility)	161
20	Oncorhynchus clarkii pleuriticus (cutthroat trout)	30-d MATC (biomass)	169
21	Chlorella sp. (green algae)	48-h IC ₅₀ (growth rate)	225
22	Physa gyrina (snail)	30-d NOEC/L (mortality)	344
23	<i>Lemna minor</i> (duckweed)	7-d EC ₁₀ (growth)	400
24	Lyngbya sp. (cyanobacteria)	18-d EC ₁₀ (growth rate)	415
25	<i>Cyclotella meneghiniana</i> (diatom)	5-d EC_{10} (growth rate)	477
26	Ceratophyllum demersum (hornwort)	15-d LOEC (chlorophyll content and biomass)	1116
27	Chlamydomonas sp. (green algae)	10-d EC ₁₀ (growth rate)	1428
28	Scenedesmus quadricauda (green	5-d EC ₁₀ (growth rate)	1628
29	Rhithrogena hageni (mayfly)	10-d EC ₁₀ (mortality)	1696

Table 8.1. Endpoints Used to Determine the Long-term Freshwater CWQG for Dissolved Zinc (from CCME 2018).

^a Normalized for hardness and dissolved organic carbon—see text for details.

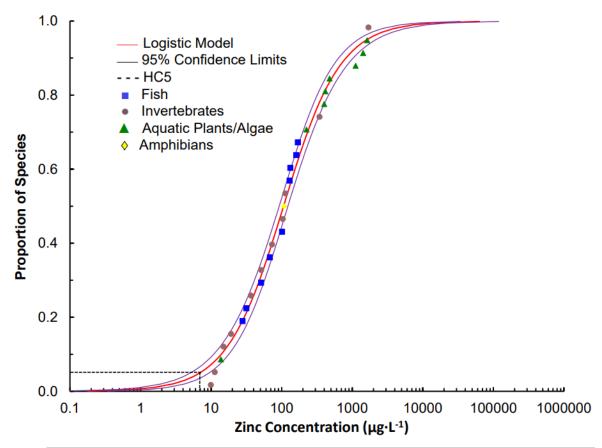


Figure 8.1. Long-term species sensitivity distribution (SSD) for dissolved zinc in fresh water, derived by fitting the logistic model to the long-term endpoints of 29 aquatic species. Data were normalized to hardness, dissolved organic carbon and pH, and converted to dissolved concentrations. The fifth percentile on the long-term SSD is 7.0 μ g·L⁻¹.

8.1.3.2. Short-term Water Quality Guidelines

CCME derives short-term benchmark concentrations using severe effects data (such as lethality) for defined short-term exposure periods. These benchmarks are estimators of severe effects to the aquatic ecosystem and are intended to give guidance on the impacts of severe but transient situations, such as spill events and infrequent releases of short-lived or non-persistent substances. Short-term benchmark concentrations do not provide guidance for protective levels of a substance in the aquatic environment, as they are levels that do not protect against adverse effects. The minimum data requirements for the Type A guideline approach were met and CCME used a total of 81 data points to derive the benchmark concentration (Table 4). Each species was ranked according to sensitivity.

Because water hardness and DOC were significant toxicity-modifying factors in the short-term analysis, CCME expresses the short-term benchmark as an equation into which the local water hardness and DOC must be entered in order to produce an appropriate site-specific benchmark concentration. Full details of the derivation are provided in CCME (2018). See Table 5 for examples.

Equation 1.

Short-term benchmark = exp^{(0.833[In(hardness)] + 0.240[In(DOC)] + 0.526)}

where the benchmark is expressed in dissolved zinc concentration ($\mu g \cdot L^{-1}$), hardness is measured as CaCO3 equivalents in mg·L-1 and DOC is measured in mg·L-1.

Species	Socies Species	Endpoint	Normalized effect
sensitivity			concentration ^a (µg
distribution rank			dissolved Zn·L ⁻¹)
1	Daphnia magna (water flea)	96-h LC ₅₀	22.7
2	Ceriodaphnia dubia (water flea)	48-h LC ₅₀	34.0
3	Pseudokirchneriella subcapitata (green algae)	4-h EC₅₀ (growth)	36.2
4	Ceriodaphnia reticulata (water flea)	48-h LC ₅₀	67.2
5	Chlorella pyrenoidosa (green algae)	24-h EC ₅₀ (growth)	76.3
6	Oncorhynchus mykiss (rainbow trout)	5-d LC ₅₀	84.9
7	Daphnia pulex (water flea)	48-h LC ₅₀	94.6
8	Oncorhynchus tshawytscha (Chinook	96-h LC ₅₀	99.6
9	<i>Oncorhynchus clarkii virginalis</i> (Rio Grande cutthroat trout	96-h LC ₅₀	120
10	Cottus bairdi (mottled sculpin)	96-h LC ₅₀	121
11	Salvelinus confluentus (bull trout)	5-d LC ₅₀	123
12	Morone saxatilis (striped bass)	96-h LC ₅₀	141
13	Salmo trutta (sea trout)	96-h LC ₅₀	147
14	Daphnia ambigua (cladoceran)	48-h LC ₅₀	150
15	Rhinichthys chrysogaster (longfin dace)	96-h LC ₅₀	152
16	Thymallus arcticus (Arctic grayling)	96-h LC ₅₀	171
17	Lampsilis rafinesqueana (Neosho mucket)	48-h EC50 (survival)	175
18	Pimephales promelas (fathead minnow)	96-h TLM	194
19	Daphnia longispina (cladoceran)	48-h EC50 (immobility)	210
20	Daphnia carinata (cladoceran)	48-h LC50	224
21	Oncorhynchus clarkii pleuriticus (Colorado River cutthroat trout	96-h LC50	245
22	Simocephalus vetulus (cladoceran)	48-h EC50 (immobility)	246
23	Daphnia galeata (cladoceran)	48-h EC₅₀ (immobility)	262
24	Simocephalus exspinosus (cladoceran)	48-h EC₅₀ (immobility)	307
25	Prosopium williamsoni (mountain whitefish)	96-h LC ₅₀	327
26	<i>Oncorhynchus clarkii stomias</i> (greenback cutthroat trout	96-h LC ₅₀	328
27	Acroperus elongatus (cladoceran)	48-h EC ₅₀ (immobility)	423
28	Chydorus ovalis (cladoceran)	48-h EC50 (immobility)	426
29	Ceriodaphnia pulchella (cladoceran)	48-h EC ₅₀ (immobility)	443
30	Lampsilis siliquoidea (fatmucket)	96-h LC ₅₀ (survival)	470
31	Chydorus sphaericus (cladoceran)	48-h EC50 (immobility)	516

Table 8.2. Endpoints Used to Determine the Short-term Freshwater Benchmark Concentration for Dissolved Zinc^a.

Species sensitivity	Species	Endpoint	Normalized effect concentration ^a (μg
distribution rank			dissolved Zn·L ⁻¹)
32	Ptychocheilus lucius (Colorado pikeminnow)	96-h LC ₅₀	533
33	Bufo boreas (boreal toad)	96-h LC ₅₀	535
34	Oncorhynchus nerka (sockeye salmon)	115-h LC ₅₀	717
35	Oncorhynchus kisutch (coho salmon)	96-h LC ₅₀	834
36	Culicoides furens (midge)	96-h LC ₅₀	888
37	Chironomus plumosus (midge)	96-h LC ₅₀	999
38	Physa heterostropha (snail)	96-h LC ₅₀	1021
39	Moina macrocopa (cladoceran)	48-h LC ₅₀	1144
40	Tubifex tubifex (worm)	96-h LC ₅₀	1245
41	Xyrauchen texanus (razorback sucker)	96-h LC ₅₀	1286
42	Physa gyrina (snail)	96-h LC ₅₀	1356
43	Rhinichthys cataractae (longnose dace)	96-h LC ₅₀	1382
44	Brachionus havanaensis (rotifer)	24-h LC ₅₀	1428
45	Gila elegans (bonytail)	96-h LC ₅₀	1505
46	Lymnaea luteola (snail)	96-h LC ₅₀	1542
47	Salvelinus fontinalis (brook trout)	96-h LC ₅₀	1712
48	Platygobio gracilis (flathead chub)	96-h LC ₅₀	1809
49	Hydra viridissima (green hydra)	96-h LC50	2003
50	Lirceus alabamae (isopode)	96-h LC ₅₀	2077
51	Cyprinus carpio (carp)	96-h LC₅₀	2496
52	Spirodela polyrrhiza (greater duckweed)	4-d IC ₅₀ (growth)	2505
53	Azolla pinnata (mosquito fern)	4-d IC ₅₀ (growth)	2540
54	Catostomus commersoni (white sucker)	96-h LC ₅₀	2688
55	Lepomis macrochirus (bluegill)	96-h LC ₅₀	3155
56	Catostomus latipinnis (flannelmouth sucker)	24-h LC ₅₀	3604
57	Corbicula fluminea (bivalve)	96-h LC ₅₀	3696
58	Brachydanio rerio (zebrafish)	96-h LC ₅₀	3761
59	Caecidotea bicrenata (isopode)	96-h LC ₅₀	3897
60	Gambusia holbrooki (eastern mosquitofish)	96-h LC ₅₀	4192
61	Rana hexadactyla (green pond frog)	96-h LC ₅₀	4404
62	Hydra vulgaris (pink hydra)	96-h LC ₅₀	4928
63	Bufo melanostictus (Asian toad)	96-h LC ₅₀	4945
64	Morone americana (white perch)	48-h TLm	5253
65	Daphnia magna (water flea)	96-h LC ₅₀	5420
66	Ceriodaphnia dubia (water flea)	72-h LC ₅₀	5928
67	Pseudokirchneriella subcapitata (green	96-h LC ₅₀	7293

Species sensitivity distribution rank	Species	Endpoint	Normalized effect concentration ^a (μg dissolved Zn·L ⁻¹)
68	Ceriodaphnia reticulata (water flea)	96-h TLm	7542
69	Chlorella pyrenoidosa (green algae)	24-h LC ₅₀	7666
70	Oncorhynchus mykiss (rainbow trout)	96-h LC ₅₀	8429
71	Daphnia pulex (water flea)	96-h TLm	9987
72	Oncorhynchus tshawytscha (Chinook	96-h TLm	10455
73	Oncorhynchus clarkii virginalis (Rio Grande cutthroat trout	48-h LC ₅₀	11076
74	Xenopus laevis (African clawed frog)	4-d LC ₅₀	18947
75	Lepidostoma sp. (caddisfly)	96-h LC ₅₀	35215
76	Carassius auratus (goldfish)	24-h LC ₅₀	39517
77	Rhithrogena hageni (mayfly)	96-h LC ₅₀	40479
78	Drunella doddsi (mayfly)	96-h LC50	46625
79	Chloroperlidae (stonefly)	96-h LC ₅₀	49058
80	Cinygmula sp. (mayfly)	96-h LC ₅₀	49058
81	Ephemerella sp. (mayfly)	96-h LC₅₀	49058

^a Normalized for hardness and dissolved organic carbon—see text for details.

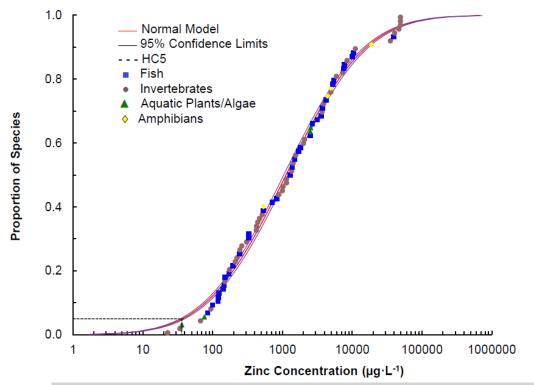


Figure 8.2. Short-term species sensitivity distribution (SSD) for dissolved zinc in fresh water. Derived by fitting the log-normal model to the short-term LC/EC50 values of 81 aquatic species. Data were normalized to hardness and

dissolved organic carbon and converted to dissolved concentrations. The fifth percentile on the short-term SSD is 37 μg·L⁻¹.

8.2. B.C. Zinc Water Quality Guidelines

The CCME Zn WQGs are based on SSDs that use an MLR approach to incorporate the toxicity modifying factors of pH, hardness, and DOC in the calculation of the WQG (CCME, 2018). In the sections below the chronic and acute CCME WQGs are reviewed and the selected assessment factor is discussed.

8.2.1. Chronic long-term water quality guidelines

The toxicity dataset used to derive the Zn long-term WQG consists of data for one bacterium, one diatom, six plants, 11 invertebrates, nine fishes, and one amphibian, the boreal toad (*Bufo boreas*) which is native to Canada. The chronic dataset used for the Zn FEQG fulfills the minimum number of species required for a type A2 guideline (ENV, 2019). It should be noted that of the 29 species with chronic data for Zn, four are not native to Canada. However, the minimum data requirement using data for only Canadian species.

The CCME Zn WQG meets the conditions specified by the B.C. derivation protocol for Type A2 WQG. However, to account for the sources of uncertainty associated with WQG derivation, an assessment factor (AF) must be applied to the calculated HC5 (ENV, 2019). The minimum AF to be applied to Type A WQGs is 2 which accounts for the extrapolation of lab results to field conditions and the cumulative effects of other environmental stressors. Considering that the database has data on a relatively large number of species including amphibians and important benthic invertebrates (i.e., Ephemeroptera, Plecoptera, Tricoptera or EPT), and data are available on all endpoints (e.g., survival, growth, and reproduction) the minimum AF of 2 was applied to the calculated HC₅.

The chronic long-term WQG for dissolved zinc for the protection of freshwater aquatic life is calculated using the equation:

Chronic long-term WQG (μ g L⁻¹) = $\frac{\exp(0.947[\ln(\text{hardness})] - 0.815[\text{pH}] + 0.398[\ln(\text{DOC})] + 4.625)}{2}$

The equation is valid between hardness 23.4 and 399 mg CaCO3·L⁻¹, pH 6.5 and 8.13 and DOC 0.3 to 22.9 mg·L⁻¹.

		00			1, 1	,
рН 6.5		Hardness (mg CaCO3·L ⁻¹)				
DOC (mg/L)	25	50	75	100	200	399*
0.5	4.1	7.9	11	15	29	56
2.0	7.1	13	20	26	51	97
5.0	10	19	29	34	73	140
10.0	13	26	38	50	96	185
22.9*	18	36	53	70	134	258
рН 7.0						
	25	50	75	100	200	399*
0.5	2.7	5.2	7.7	10	19	37
2.0	4.7	9.1	13	17	34	65
5.0	6.8	13	19	25	48	93
10.0	8.9	17	25	33	64	123
22.9*	12	24	35	46	89	171
рН 7.5						
	25	50	75	100	200	399*
0.5	1.8	3.5	5.1	6.7	13	25
2.0	3.1	6.1	8.9	11	22	43
5.0	4.5	8.7	13	17	32	62
10.0	5.9	11	17	22	42	82
22.9*	8.3	16	23	31	59	114
рН 8.0						
	25	50	75	100	200	399*
0.5	1.2	2.3	3.4	4.5	8.6	17
2.0	2.1	4.0	5.9	7.8	15	29
5.0	3.0	5.8	8.5	11	21	41
10.0	3.9	7.6	11	14	28	54
22.9*	5.5	10	15	20	39	76

Table 8.3. BC long term chronic WQGs for dissolved Zn (μ g/L) at various levels of DOC (farthest left column), pH (separate blocks), and hardness (values ranging from 25 to 399 across the top; adapted from CCME 2018).

*The B.C. WQG equation is valid between hardness 23.4 and 399 (mg CaCO3/L), pH 6.5 and 8.13 and DOC of 0.3 to 22.9 (mg/L).

8.2.2. Acute short-term water quality guidelines

The CCME acute WQG also used MLR to account for toxicity modifying factors, however, only hardness and DOC were included in the multiple linear regression model. The toxicity dataset used to derive the CCME short-term WQG for dissolved Zn consists of data for four plants, 40 invertebrates, 34 fishes, and four amphibians. The acute dataset consists of Canadian and non-Canadian species, however if only the Canadian species are considered, the database fulfills the minimum number of species required for a B.C. Type A2 guideline (ENV, 2019).

Considering the large number of species in the data including data on all aquatic taxa (e.g., amphibians, EPT, and plants), the minimum AF of 2 is sufficient.

The B.C. short term WQG for dissolved Zn is calculated using the following equation:

Acute short-term WQG (μ g L⁻¹) = $\frac{\exp(0.833[\ln(hardness)] + 0.240[\ln(DOC)] + 0.526)}{\exp(0.833[\ln(hardness)] + 0.240[\ln(DOC)] + 0.526)}$

The equation is valid between hardness 13.8 and 250.5 mg CaCO3·L⁻¹ and DOC 0.3 and 17.3 mg·L⁻¹

	Hardness (mg CaCO3·L ⁻¹)							
DOC (mg/L)	15	25	50	75	100	150*	200	250.5*
0.5	7	10.5	18.5	26	33	46.5	59	71.5
2.0	9.5	14.5	26	36.5	46.5	65	82.5	99.5
5.0	12	18	32.5	45.5	57.5	81	103	124
10.0	14	22.5	38.5	53.5	68	95.5	122.5	146.5
17.3*	16	25.5	43.5	61	77.5	109	138.5	167

Table 8.4. Examples of short-term WQG (μ g/I) for dissolved Zn in freshwater at various levels of hardness and DOC.

* The short-term benchmark equation is valid between hardness 13.8 and 250.5 (mg CaCO3/L) and from DOC 0.3 to 17.3 (mg/L).

9. COMPARISON OF AMBIENT ZINC CONCENTRATIONS TO WATER QUALITY GUIDELINES

Water quality guidelines are commonly used to determine the potential risk of toxicity to aquatic life from a given substance in ambient conditions. In general, if ambient concentrations are below the WQG the risk is assumed to be low. It is important to understand how the assessment of risk to aquatic life will change with the updated Zn WQG. To answer this question, water quality data (dissolved Zn, pH, DOC, and hardness) collected over the past 22 years from freshwater sites were extracted from the EMS database. Non-detect data for Zn were not included if the MDL was greater than 5 μ g/L. Non-detect data for DOC were not included if the MDL was greater than 0.5 mg/l. Results reported as "<MDL", were given the value of the MDL. A total of 722 records were retrieved to calculate the 1997 WQG (i.e., hardness, DOC, and dissolved Zn).

Total Zn exceeded the 1997 WQGs (based on total Zn) concentrations 4.1% of the time (30/722) (Figure 9.1) and dissolved zinc exceeded the updated WQG 8.6% of the time (596/6913) (Figure 9.2). The higher rate of exceedance of the updated WQG compared to the 1997 WQG is mainly because the updated WQG could be as low as 1.35 μ g/L when pH is high and DOC and hardness are low, whereas the lowest value for 1997 WQG is 7.5 μ g/L. In addition, some of the exceedances from the updated WQG are due to the non detect data that were replaced with the MDL of 5 ug/L (the horizontal dots at 5 ug/L in Figure 9.2)

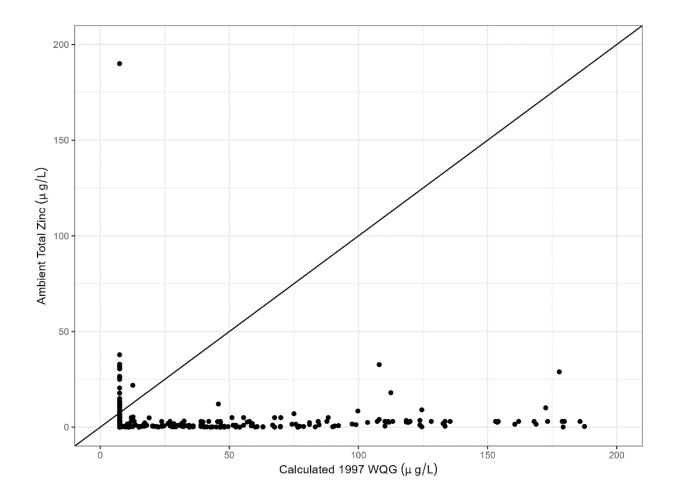


Figure 9.1. Ambient Zn concentrations compared to the 1997 chronic total Zn WQGs. Points above the solid 1:1 line represent exceedances. The minimum and maximum values for the 1997 chronic WQG are 7.5 and 187.5 μ g/L respectively.

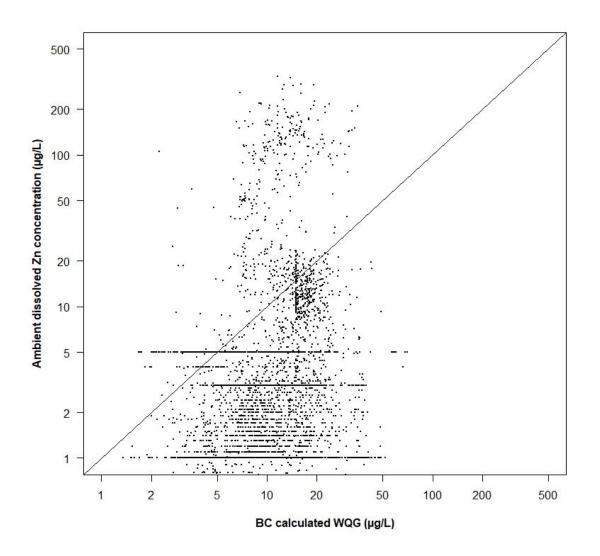


Figure 9.2. Ambient Zn concentrations compared to the 2023 chronic dissolved Zn WQGs. Points above the solid 1:1 line represent exceedances.

REFERENCES

- AEP (Alberta Environmental Protection) 2018. Environmental Quality Guidelines for Alberta Surface Waters. Water Policy Branch, Alberta Environment and Parks. Edmonton, Alberta.
- ATSDR (Agency for Toxic Substances and Disease Registry), 2007. Toxicological profile: Zinc. Toxicological profiles series. Available from https://www.atsdr.cdc.gov/ToxProfiles/tp.asp?id=302&tid=54
- ANZECC (Australian and New Zealand Environment and Conservation Council and Agriculture and Resource Management Council of Australia and New Zealand). 2000. National water quality management strategy: An introduction to the Australian and New Zealand guidelines for fresh and marine water quality. Available from <u>http://www.agriculture.gov.au/water/quality/guidelines</u>
- Budavari, S. 1996. The Merck index. Merck & Co., Inc., Rahway, NJ.
- Cairns, J., Jr., Bahns, T.K., Burton, D.K., Dickson, K.L., Sparks, R.E., and Waller, W.T. 1971. The effects of pH, solubility and temperature upon the acute toxicity of zinc to the bluegill sunfish (Lepomis macrochirus Raf.). Trans. Kans. Acad. Sci. 74(1): 81–92.
- Canadian Council of Ministers of the Environment. 2007. A protocol for the derivation of water quality guidelines for the protection of aquatic life. In Canadian environmental quality guidelines, Canadian Council of Ministers of the Environment, 1999. Canadian Council of Ministers of the Environment, Winnipeg, MB. Available from www.ccme.ca.
- Canadian Council of Ministers of the Environment. 2018. Scientific criteria document for the development of the Canadian water quality guidelines for the protection of aquatic life: zinc. Canadian Council of Ministers of the Environment, Winnipeg, MB. Available from www.ccme.ca.
- Clement Associates, U.S. Agency for Toxic Substances and Disease Registry, and U.S. Environmental Protection Agency. 1989. Toxicological profile for zinc. Agency for Toxic Substances and Disease Registry, U.S. Public Health Service, Atlanta, GA.
- Eisler, R. 1993. Zinc hazards to fish, wildlife, and invertebrates: A synoptic review [Biological report 10; contaminant hazard reviews report 26]. U.S. Department of the Interior Fish and Wildlife Service, Washington, D.C.
- ENV (formally: British Columbia Ministry of Environment, Land and Parks), 1997. Water quality criteria for zinc. Water management branch.
- ENV (British Columbia Ministry of Environment and Climate Change Strategy), 2019. Derivation of Water Quality Guidelines for the Protection of Aquatic Life in British Columbia. Water Quality Guideline Series, WQG-06. Prov. B.C., Victoria B.C.
- ENV (British Columbia Ministry of Environment and Climate Change Strategy), 2021. Molybdenum Water Quality Guideline for the Protection of Freshwater Aquatic Life, Livestock, Wildlife and Irrigation -Technical Report. Water Quality Guideline Series, WQG-07. Prov. B.C., Victoria B.C.
- European Union. 2006. Draft risk assessment report zinc metal (Final draft of September 2006). Ministry of Housing, Spatial Planning and the Environment, European Union.
- Evans, L.J. 2000. Fractionation and aqueous speciation of zinc in a lake polluted by mining activities, Flin Flon, Canada. Water Air Soil Pollut. 122(3–4): 299–316.
- Florence, T.M. 1977. Trace metal species in fresh waters. Water Res. 11: 681–687.

- Government of Ontario. n.d. Provincial (Stream) Water Quality Monitoring Network [2014 dataset]. Retrieved May 2, 2016, from https://www.ontario.ca/data/provincial-stream-water-qualitymonitoring-network.
- Hiltibran, R.C. 1971. Effects of cadmium, zinc, manganese, and calcium on oxygen and phosphate metabolism of bluegill liver mitochondria. J. Water Pollut. Control Fed. 43(5): 818–823.
- Hogstrand, C., Webb, N., and Wood, C.M. 1998. Covariation in regulation of affinity for branchial zinc and calcium uptake in freshwater rainbow trout. J. Exp. Biol. 201(11): 1809–1815.
- Hogstrand, C., Wilson, R.W., Polgar, D., and Wood, C.M. 1994. Effects of zinc on the kinetics of branchial calcium uptake in freshwater rainbow trout during adaptation to waterborne zinc. J. Exp. Biol. 186: 55–73.
- Lide, D.R. 2006. CRC handbook of chemistry and physics. CRC Press, Boca Raton, FL.
- Lindsay, W.L. 1979. Chemical equilibria in soils. John Wiley & Sons, Inc., New York, NY.
- Malle, K.G. 1992. Zinc in the environment. Acta Hydroch. Hydrob. 20(4): 196–204.
- Muyssen, B.T.A., and Janssen, C.R. 2002. Accumulation and regulation of zinc in Daphnia magna: Links with homeostasis and toxicity. Arch. Environ. Contam. Toxicol. 43: 492–496.
- Muyssen, B.T.A., De Schamphelaere, K.A.C., and Janssen, C.R. 2006. Mechanisms of chronic waterborne Zn toxicity in Daphnia magna. Aquat. Toxicol. 77(4): 393–401.
- MDDEFP (Ministère du Développement durable, de l'Environnement, de la Faune et des Parcs), 2013. Critères de qualité de l'eau de surface, 3e edition, Québec, Direction du suivi de l'état de l'environnement. Gouvernment du Québec. 510 p.
- MWS (Manitoba Water Stewardship) 2011. Manitoba Water Quality Standard, Objectives, and Guidelines. Water Science and Management Branch, Manitoba Water Stewardship.
- NRCAN (Natural Resources Canada), 2007. Minerals and metals: A world to discover. Main minerals and metals produced in Canada. Available from http://www.nrcan.gc.ca/mining-materials/publications/17686
- Newhook, R., Hirtle, H., Byrne, K., and Meek, M.E. 2003. Releases from copper smelters and refineries and zinc plants in Canada: Human health exposure and risk characterization. Sci. Total Environ. 301: 23–41.
- Nriagu, J.O., Lawson, G., Wong, H.K.T., and Cheam, V. 1996. Dissolved trace metals in Lakes Superior, Erie and Ontario. Environ. Sci. Technol. 30: 178–187.
- Nova Scotia Environment. 2015. Surface water: Monitoring and reporting. Available from https://novascotia.ca/nse/surface.water/surfacewater.monitoring.reporting.asp
- Regional Aquatics Monitoring Program. n.d. Monitoring database: Water quality. Retrieved May 2, 2016, from http://www.ramp-alberta.org/data/Water/water.aspx
- Reimann, C., and De Caritat, P. 1998. Chemical elements in the environment: Factsheets for the geochemist and environmental scientist. Verlag Berlin Heidelberg, Germany, Springer.
- Skidmore, J.F. 1970. Respiration and osmoregulation in rainbow trout with gills damaged by zinc sulphate. J. Exp. Biol. 52: 481–494.
- Skidmore, J.F., and Tovell, W.A. 1972. Toxic effects of zinc sulphate on the gills of rainbow trout. Water Res. 6: 217–230.

- Spear, P.A. 1981. Zinc in the aquatic environment: Chemistry, distribution and toxicology. National Research Council of Canada, Ottawa, ON.
- Spry, D.J., and Wood, C.M. 1985. Ion flux, acid-base status, and blood gases in rainbow trout, Salmo gairdneri, exposed to toxic zinc in natural soft water. Can. J. Fish. Aquat. Sci. 42(8): 1332–1341.
- Stumm, W., and Morgan, J.J. 1981. Aquatic chemistry: An introduction emphasizing chemical equilibria in natural waters. John Wiley & Sons, Inc., New York, NY.
- Tri-Star Environmental Consulting. 2006. The statistical approaches and data availability for a case study comparison: Natural background levels and the CCME WQI [Report to Environment Canada see contact information below].
- USEPA (United States Environmental Protection Agency), 1995. Water quality criteria documents for the protection of aquatic life in ambient water. Office of water, Washington, DC.
- U.S. Environmental Protection Agency, Office of Water. 1996. The metals translator: Guidance for calculating a total recoverable permit limit from a dissolved criterion [EPA 823-B-96-007]. Available from https://www.epa.gov/npdes/pubs/metals translator.pdf

Water Security Agency, 2015. Surface Water Quality Objectives. Water Security Agency EPB356.

- Wayland, M., and Crosley, R. 2006. Selenium and other trace elements in aquatic insects in coal mineaffected streams in the Rocky Mountains of Alberta, Canada. Arch. Environ. Contam. Toxicol. 50(4): 511–522.
- WHO (World Health Organization). 2001. Environmental health criteria 221: Zinc. World Health Organization, Geneva, Switzerland. Available from http://www.who.int/ipcs/publications/ehc/ehc_221/en/